
7. A continuous fractionating column is to be designed for separating 10,000 kg/hr of a liquid mixture containing 40 mole% methanol & 60 mole% water in to an overhead product containing 97 mole% methanol and a bottom product having 98 mole% water. A molar reflux ratio of 3 is used.

Calculate:

- (i) Moles of overhead product obtained per hour &
- (ii) Number of ideal plates & location of the feed plate, if the feed is at its bubble point.

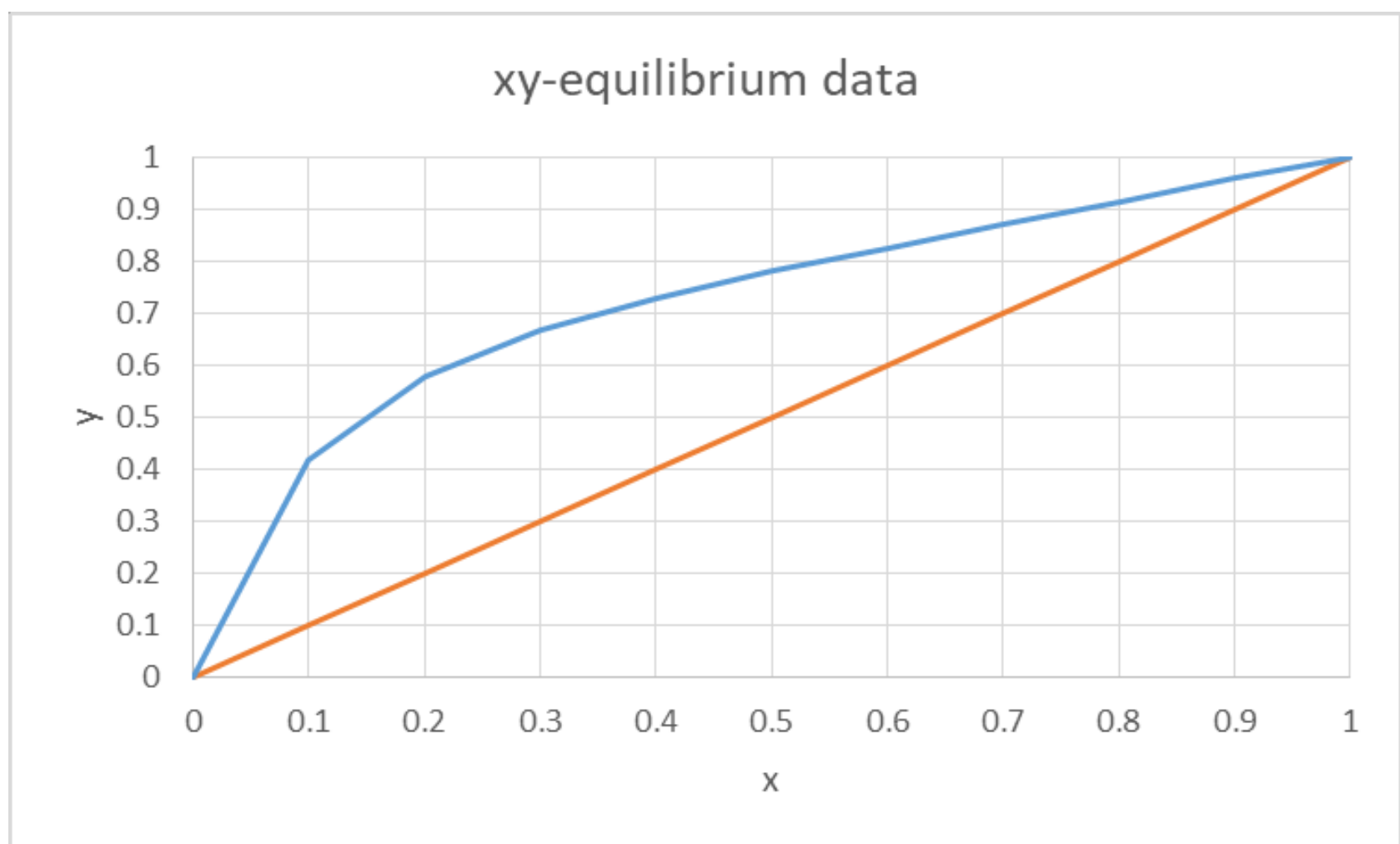
Equilibrium Data:

X	0.1	0.2	0.3	0.4	0.5	0.6	0.7	0.8	0.9
Y	0.417	0.579	0.669	0.729	0.78	0.825	0.871	0.915	0.959

x= mole fraction of methanol in liquid &

y= mole fraction of methanol in vapor phase





Solution:-

Feed
10,000 kg/hr
0.4 Methanol
0.6 water

Distilled
→
 $x_D = 0.97$

Bottom
→
 $x_B = 0.02$

given $R = 3$

↳ assume sat'd liquid
 $q = 1$ ← $\underline{q}$ gibso Gm

Const Mol flow

$$\rightarrow 10000 \frac{\text{kg} (\text{H}_2\text{O} + \text{Methanol})}{\text{hr}} \cdot \frac{1 \text{ kmol}}{21.8 \text{ kg}} = 458.7$$

$\overline{M_w} \Rightarrow$

$$\frac{1}{\overline{M_w}} = \frac{x_1}{M_{w1}} + \frac{x_2}{M_{w2}} =$$

$$\frac{1}{\overline{M_w}} = \frac{0.4}{32} + \frac{0.6}{18}$$

$$\overline{M_w} = 21.8$$

(1) Moles of overhead / hr (D)

by Material Balance

$$458.7 = D + B \quad - (1)$$

$$(0.4 + 458.7) = (0.97D) + (0.02B)$$

↳ Solve

$$\boxed{D = 180.7}$$

$$\boxed{B = 278.1}$$

(2) N_p

operating line of the top section

$$y = \frac{R}{(R+1)} x + \frac{x_D}{(R+1)} \Rightarrow y = 0.75x + 0.2425$$



Top section operating line:-

$$y = 0.75 + 0.2425x$$

Feed operating line:-

$$y = x$$

vertical line $\leftarrow$

Bottom section:-

As shown in the figure $\rightarrow$

