University of Jordan – Dept. of Chemical Engineering

0915321 ChE Thermodynamics (I)

Problem Set # (4)

Instructor: Dr. Ahmad AbuYaghi

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- These are suggested problems for you to work out.
- The problem numbers refer to those given at the end of Chapters in the Textbook: (Cengel & Boles, Thermodynamics: An Engineering Approach, 8th edition)
- The final answer(s) is/are provided.

Problem #	Final Answers
6-15	Efficiency = 45.4%
6-22	QH = 500 MW , Efficiency = 30%
6-39	COP = 2.22, QL = 4400 kJ/hour
6-55	COP = 2.64, QL = 1.96 kW
6-77	Efficiency = 70%, W = 9.33 kW
6-90	Compare actual COP to maximum COP
7-24	Sgen = 0
7-35	$\Delta h = -807.4 \text{ kJ/kG}$
7-44	q = 471 kJ/kg
7-50	w = 887.1 kJ/kg
7-62	$\Delta S_{total} = 4.08 \text{ kJ/K}$
7-72	+0.38 versus -0.21 kJ/kg/k
7-80	$\Delta S = -0.0025 \text{ kJ/kg/s}$
7-85	T ₂ = 576 K