

HEAT INTEGRATION : PINCH TECHNOLOGY

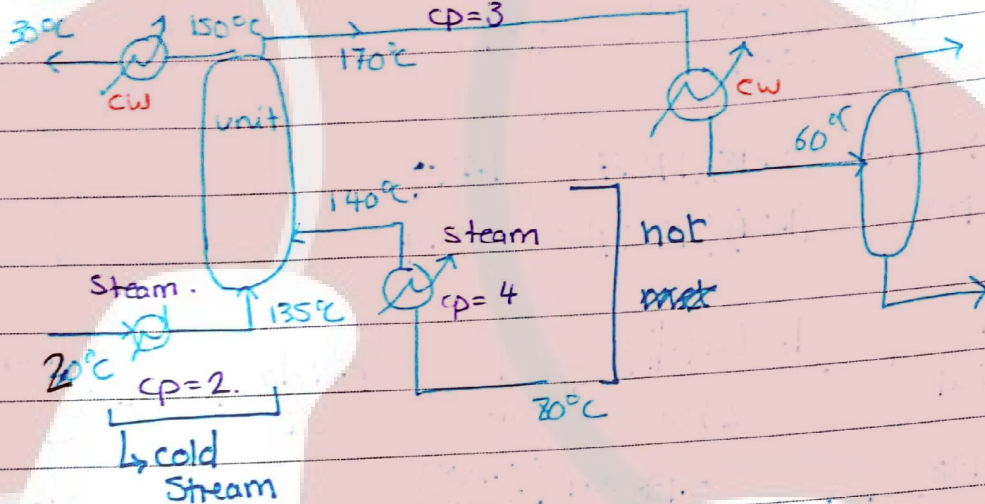
No.

Heating utility: ST

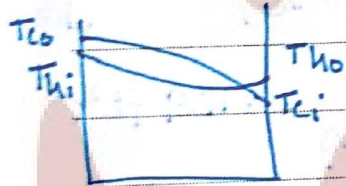
How to redesign equipments to have the least amount of exchanger

ILLUSTRATED

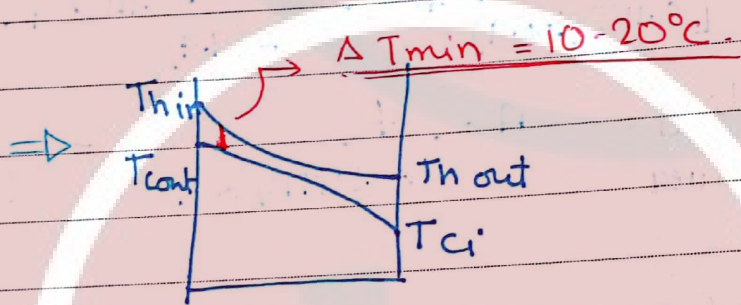
EXAMPLE



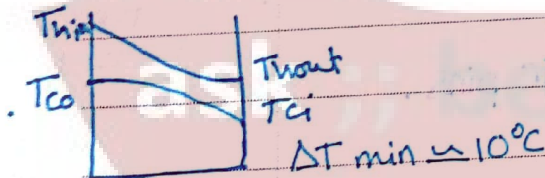
Sub case:



Can't be : haven't reach achieved.



originally :



hot → add $\frac{\Delta T_{min}}{2}$ to cold stream.
cold → subtract $\frac{\Delta T_{min}}{2}$ from hot stream.

↑ cp ↑ q.

for simplicity: all calculations are based on unit flow rate.

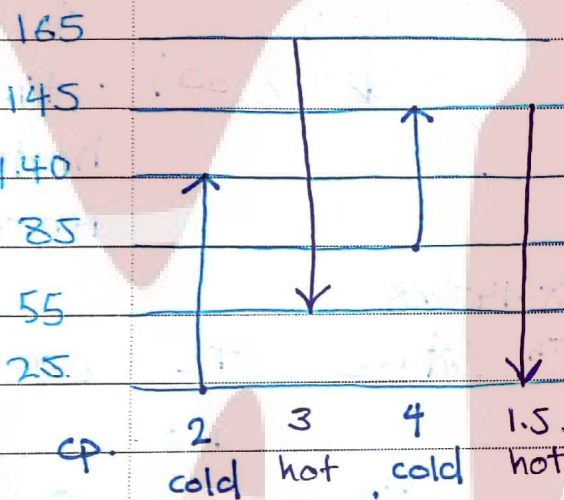
$$\frac{\Delta T_{\min}}{2} = \frac{10}{2} \begin{cases} \text{add to cold} \\ \text{subtract from hot.} \end{cases}$$

new values.

November 21th.

Streams	T_s	T_f	CP
cold	20 (25)	135 (140)	2
cold	80 (85)	140 (145)	4
hot	170 (165)	60 (55)	3
hot	150 (145)	30 (25)	1.5

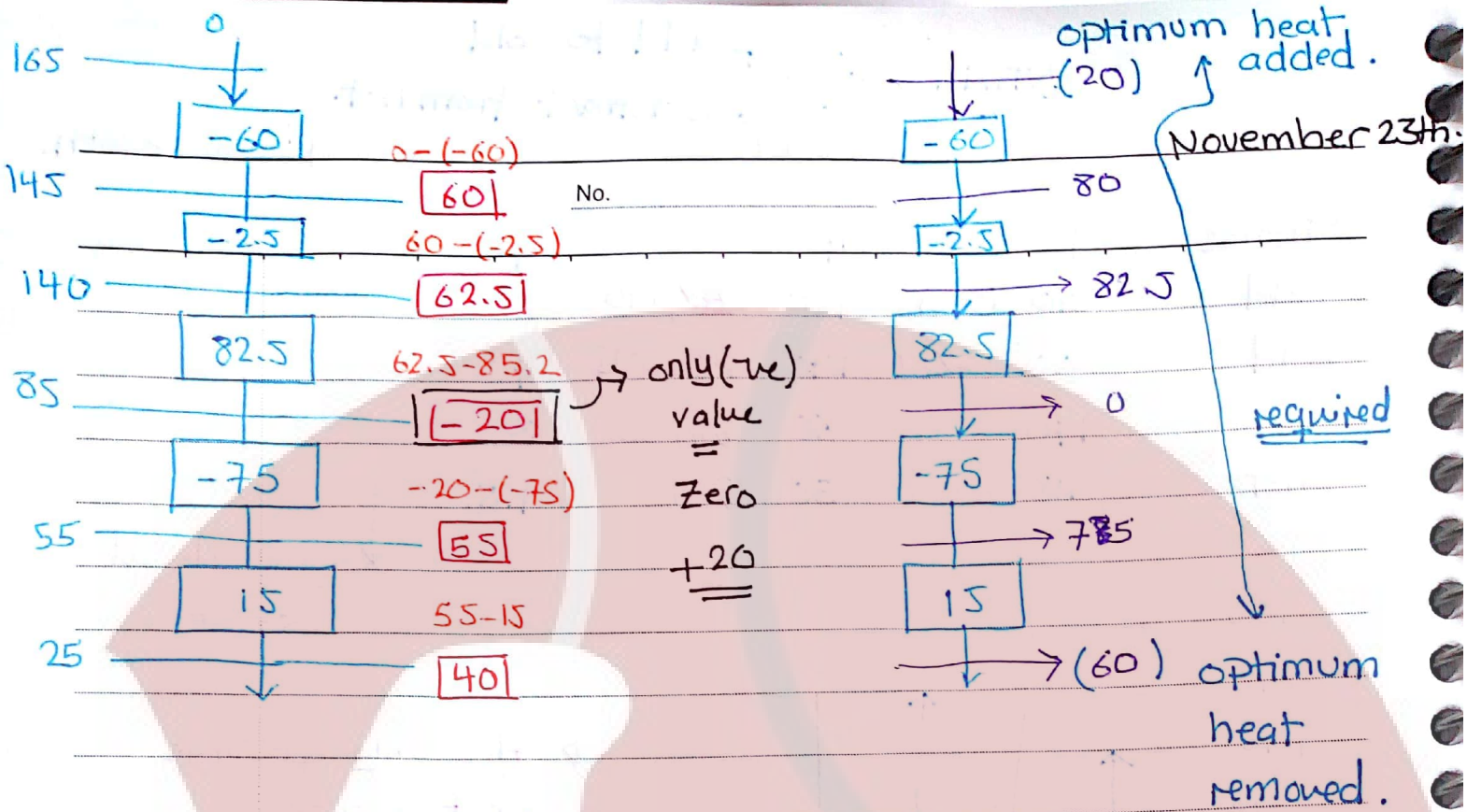
include all streams



⇒ the only possible heat exchange.

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T	interval	$T_i - T_{i+1}$	$\sum CP_c - \sum CP_h$	ΔT_i	excess heat Surplus Defisit
165	1	20	$\sum 0 - \sum 3 = -3$	-60 kW	S
145	2	5	$4 - (3 + 1.5) = -0.5$	-2.5 kW	S
140	3	55	$(4 + 2) - (3 + 1.5) = 1.5$	82.5 kW	D
85	4	30	$(2) - (3 + 1.5) = -2.5$	-75 kW	S
55	5	30	$(2) - 1.5 = 0.5$	15 kW	D
25					



Method I.

T	ΣC_{Pc}	ΣC_{Ph}	T_h	ΔT_{Th}	ΔT_{acc}	T_c	ΔH_c	ΔH_c^{acc}
165	0	3	170	60	510	160	0	470
145	4	4.5	150	22.5	450	140	20	470
140	6	4.5	145	247.5	427.5	135	330	450
85	2	4.5	90	135	180	50	60	120
55	2	1.5	60	45	45	20	60	60
25			30					

↑
accumulation
up wards.

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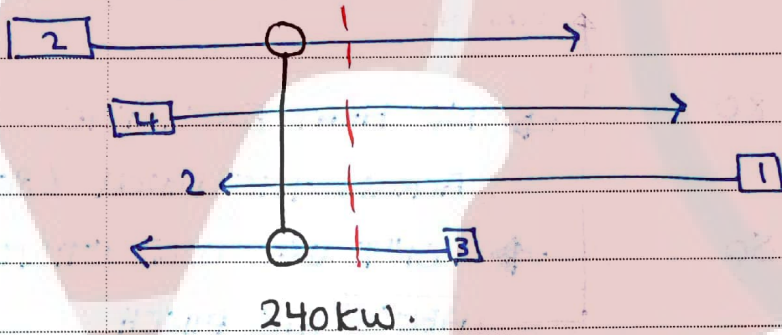
No. _____

$$C_{P_c} = 4 > C_{P_h} = 3.$$

$$Q_c = 4(140 - 80) = 240 \text{ kW} \quad \checkmark \text{ perfect match.}$$

$$Q_h = 3(170 - 90) = 240 \text{ kW}$$

Second exchanger:



$\Rightarrow$ THIS ALL FOR ABOVE the pinch.

⇒ BELOW THE PINCH:-

$$C_{PH} > C_{PC} \quad \max |C_{PH} - C_{PC}| \Delta t$$

Stream [2] with [1]

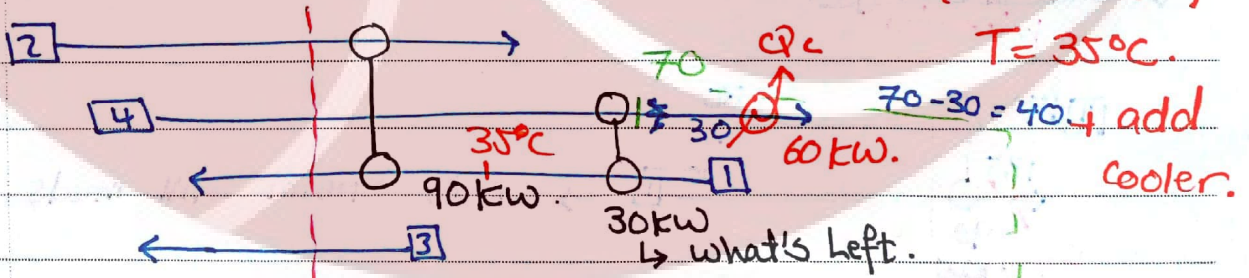
$$Q_h = C_{p_h} (T_2 - T_1) = 3(90 - 60) = 90 \text{ kW}.$$

$$Q_c = 2(80 - 20) = 120 \text{ kW} \rightarrow$$

isn't enough.

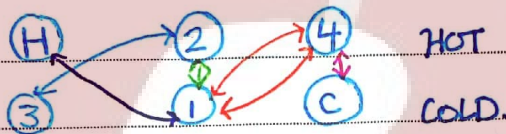
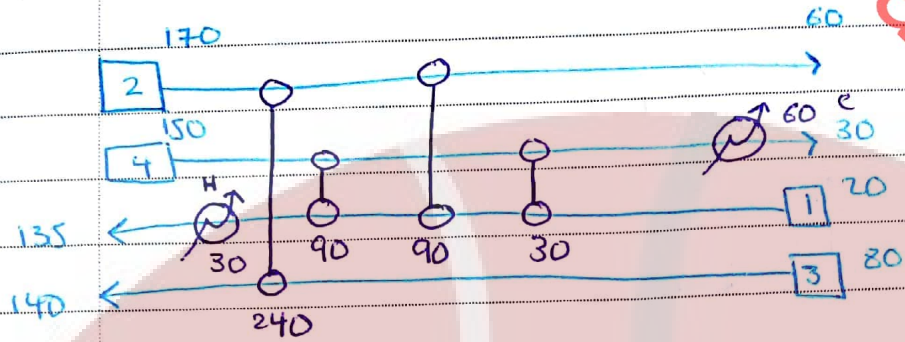
$$90 = 2(80 - T_{C \text{ new}})$$

$$T = 35^{\circ}\text{C}.$$

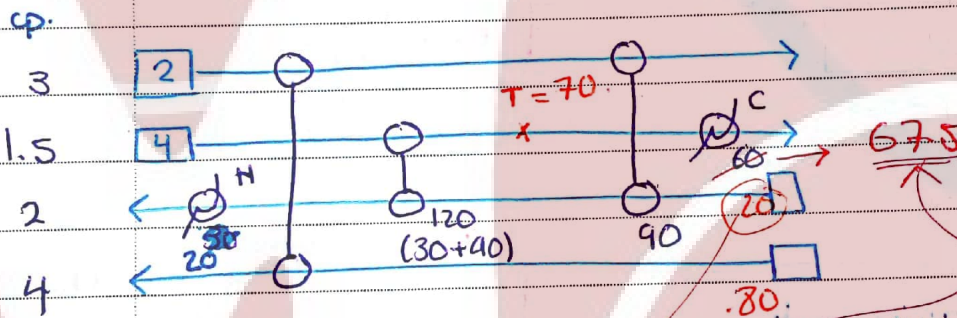


$$Q_h = C_p (90 - T_w) = 30^\circ\text{C}$$

$$90 - T_h = \frac{30}{1.5} = 20^\circ\text{C} \quad T_h = 90 - 20 = 70^\circ\text{C}$$



you can only manipulate streams with heaters/coolers



$$Q = 120 = 1.5(150 - T) = 70^\circ\text{C} \quad \Delta T_{\min} = 5 < 10 \text{ violation.}$$

$$Q = 90 = 2(T - 20) = 65^\circ\text{C} \quad \Rightarrow \text{need either to } \begin{matrix} 70 \rightarrow 75 \\ 65 \rightarrow 60 \end{matrix}$$

$$T_{\text{new}} = 78^\circ\text{C} : Q = 1.5(75 - 30) = 67.5$$

old duty.

$$Q = 112.5 \neq 2(120 - 65) = 110$$

$$Q = (112.5) = 2(T - 65) \Rightarrow T = 121^\circ\text{C}$$

$$\Rightarrow \text{new } Q_H = 2(135 - 121) = 27.5 \text{ kW}$$

$$\text{New } Q_C = 67.5 \text{ kW} - 27.5 \text{ kW} =$$

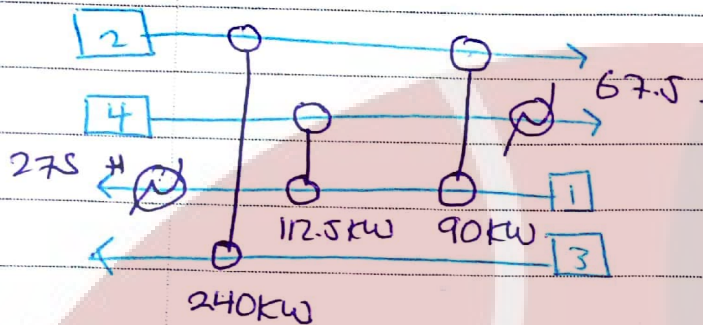
* Relaxation of energy to maintain the pinch point

⇒ 67.5: optimum/minimum value to have

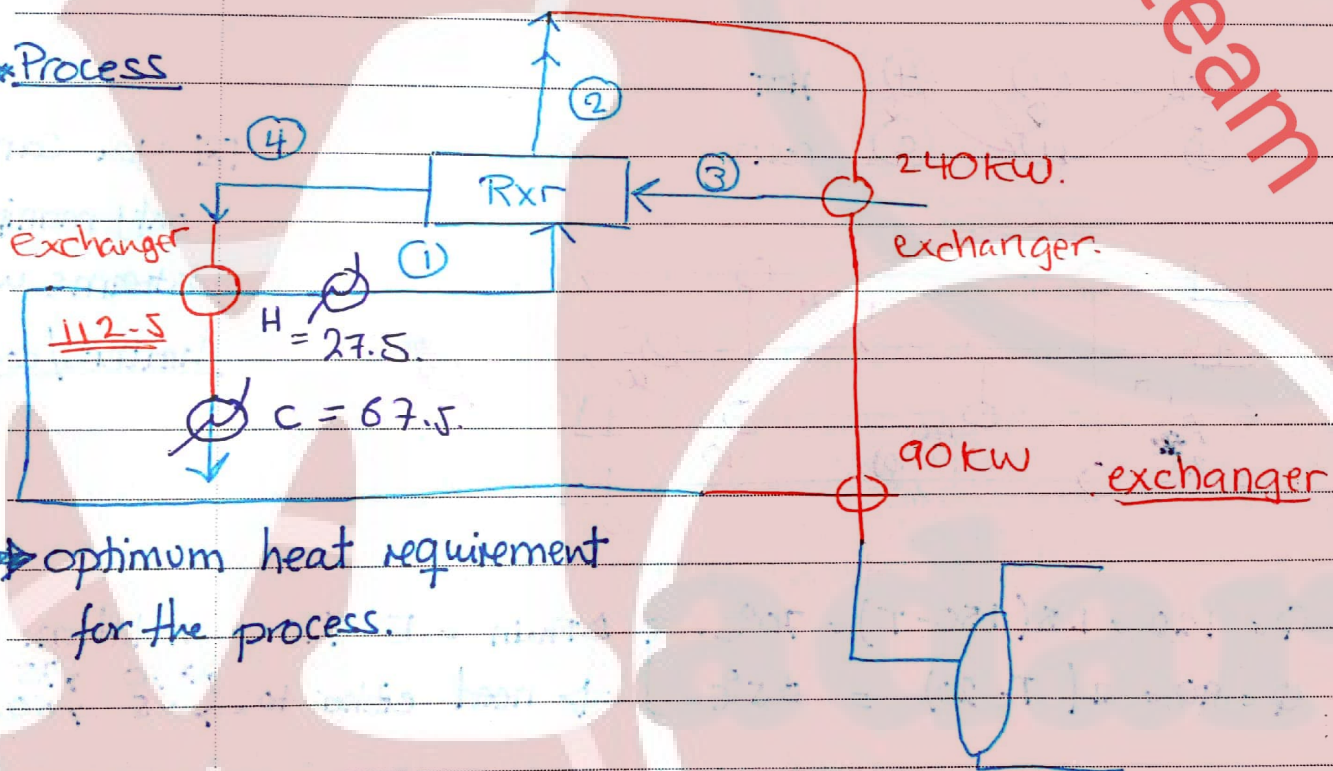
Nov. 28th

No.

Finally min ΔT .



Process



⇒ optimum heat requirement for the process.

$$\begin{aligned} \Rightarrow \text{Total heat duty} &= |hot| + |cold| \\ &= 240 + 90 + 112.5 + 67.5 + 27.5 \\ &= 537.5 \text{ kW} \end{aligned}$$

⇒ no. of exchangers = 3 + cooler + heater

Resolve:

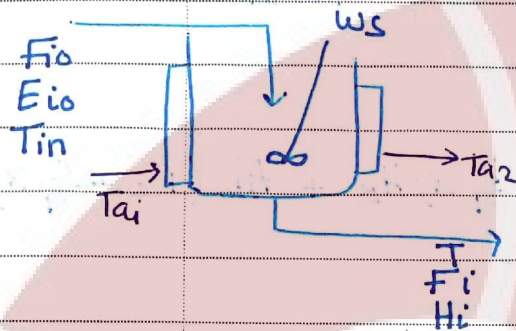
① $\Delta T_{min} = 20^\circ$
and the total heat duty.

* Reactor Design.

December 5th

No.

→ non isothermal CSTR design.



* energy balance =

$$\text{energy in} - \text{energy out} + \dot{Q} - \dot{W} = \frac{dE}{dt}$$

$$F_i E_{i, \text{in}} - F_o E_{o, \text{out}} + \dot{Q} - \dot{W} = \frac{dE}{dt}$$

$$E = U_i + KE + PE + hE^{\text{friction}}$$

$$E \approx U_i$$

enthalpy

$$\frac{dE_{\text{sys}}}{dt} = \dot{Q} - \dot{W} + \sum F_i H_{i, \text{in}} - \sum F_i H_{i, \text{out}}$$

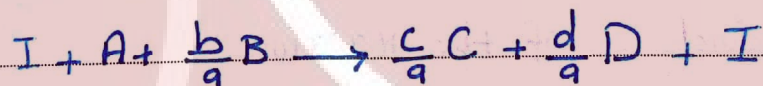
standard

NOTE

$$\sum F_i H_{i, \text{in}} = \sum F_i H_{i, \text{in}}^{\circ}$$

$\sum F_i H_{i, \text{out}} =$ all energy associated with rxn and heat in the stream.

~~STOICHIOMETRY~~



$$\sum F_i H_i = F_A H_A + F_B H_B + F_C H_C + F_D H_D + F_I H_I$$

$$F_A = F_{A0} (1 - x)$$

$$F_B = F_{A0} (\theta_B - \frac{b}{a} x) \quad \theta_B = \frac{F_{B0}}{F_{A0}}$$

$$F_C = F_{A0} (\theta_C + \frac{c}{a} x)$$

$$F_D = F_{A0} (\theta_D + \frac{d}{a} x)$$

out let stream

↳ we need to separate heat from rxn, and heat not associated.

$$\sum_{out} F_i H_i = F_{A0}(1-x)H_A + F_{A0}(\theta_B - \frac{b}{a}x)H_B + F_{A0}(\theta_C + \frac{c}{a}x)H_C + F_{A0}(\theta_D + \frac{d}{a}x)H_D$$

$$= (F_{A0}H_A + F_{A0}\theta_B H_B + F_{A0}\theta_C H_C + F_{A0}\theta_D H_D + F_{A0}\theta_I H_I + \dots)$$

$$\sum_{out} F_i H_i = \left(\sum_{i=1}^N F_i \theta_i H_i \right) + x F_{A0} \left[\frac{d}{a} H_D + \frac{c}{a} H_C - \frac{b}{a} H_B - H_A \right] + \text{inert}$$

$$\sum_{out} F_i H_i = \sum F_i \theta_i H_i + x F_{A0} \Delta H_R(T)$$

$$\dot{Q} - \dot{W}_S = F_{A0} \sum_{T_0}^T \theta_i C_{Pi} dT - [\Delta H_R(T)] F_{A0} X = 0$$

Split to two terms.

T_R standard of the reaction.
 $T_0 \rightarrow T_R$ $T_R \rightarrow T$

energy equation

$$\dot{Q} - \dot{W}_S = F_{A0} \sum_{i=1}^N \int_{T_0}^T \theta_i C_{Pi} dT - \left[\Delta H_R^\circ(T) + \int_{T_R}^T \Delta C_P dT \right] F_{A0} X$$

need a coupling equation.

⇒ Design equation of CSTR

Design equation: $V = \frac{F_{A0} - F_A}{-r_A} = \frac{F_{A0} X}{-r_A} = f(X, T).$

$r_A = f(T, X) \quad r_A = k C_A^\alpha \quad k = A e^{-\frac{E_A}{RT}} \text{ (polymath)}$

Q: changes with temp. of Jacket.
temp. of Jacket changes with reaction Temp.

need to find it $\leftarrow$
↓
Substituted in energy equation.
 $Q = UA \Delta T_{LM}$ multiplied with correction factor
Simple $[F=1]$

↓
LMTD = $\frac{(T - T_{a1}) - (T - T_{a2})}{\ln \left(\frac{T - T_{a1}}{T - T_{a2}} \right)}$ T_i : temp in CSTR

= $\frac{T_{a2} - T_{a1}}{\ln \left(\frac{T - T_{a1}}{T - T_{a2}} \right)}$

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energy balance across the Jacket

$$\text{Rate in} - \text{Rate out} = \text{Rate of heat transferred to Rx}$$

$$m \underline{C_p} (T_{a1} - T_R) - m \underline{C_p} (T_{a2} - T_R) = \frac{UA (T_{a1} - T_{a2})}{\ln \left(\frac{T - T_{a1}}{T - T_{a2}} \right)}$$

at inlet at outlet

energy

balance

$$\dot{Q} - \dot{W}_s - F_{A0} \sum_{T_0}^T \theta_i C_{p,i} dT - \left[\Delta H_R^\circ(T_R) + \int_{T_R}^T \Delta C_p dT \right] F_{A0} X = 0$$

relate it with $\dot{Q}$ for Jacket
 $\dot{Q}$ jacket changes with temp. $\rightarrow T_{A2} \rightarrow$ get an expression

energy

balance

across the
Jacket

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$$m_{cp} (T_{A1} - T_R) - m_{cp} (T_{A2} - T_R) = UA \frac{(T_{A1} - T_{A2})}{\ln \left(\frac{T_{A1} - T}{T_{A2} - T} \right)}$$

* Rewriting: $m_{cp} (T_{A1} - T_{A2}) = UA \frac{(T_{A1} - T_{A2})}{\ln \left(\frac{T_{A1} - T}{T_{A2} - T} \right)}$

T in side

$$T_{A2} = T - (T - T_{A1}) e^{\left(\frac{-UA}{m_{cp}} \right)}$$

the other

* expression for T_{A2}

in side

$$Q = m_{cp} (T_{A1} - T_{A2})$$

$$= m_{cp} \left(T_{A1} - \left(T - (T - T_{A1}) e^{\left(\frac{-UA}{m_{cp}} \right)} \right) \right)$$

$$Q = m_{cp} \left[(T_{A1} - T) \left\{ 1 - e^{\frac{-UA}{m_{cp}}} \right\} \right]$$

Simplify
using

$$e^{-x} = 1 - x + \dots$$

Taylor

Series

$$Q = m_{cp} (T_{A1} - T) \left[1 - \left(1 - \frac{UA}{m_{cp}} \right) \right]$$

$$\dot{Q} = UA (T_{A1} - T)$$

overall heat coefficient related
to $(U) \rightarrow$ local heat coefficient
effected by temp

NOT added to known (is included in volume)

December 7th.

No.

$$\Delta H_R^\circ(T_R) + \Delta C_P(T - T_R)$$

$$\frac{UA}{F_{A0}} (T_a - T) - \sum \theta_i C_{Pi} (T - T_0) - \Delta H_R(T) X = 0 \quad (\text{shaftwork} = 0)$$

Design equation

$$V = \frac{F_{A0} X}{-r_A}$$

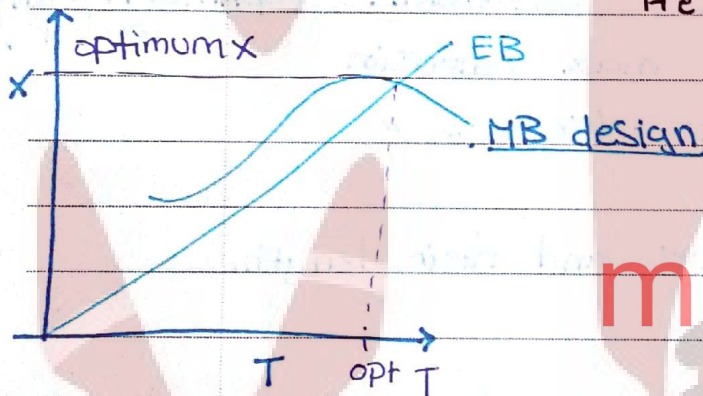
$$r_A = k C_A^A$$
$$k = A e^{-\frac{E_a}{RT}}$$

⇒ Assume a value of volume then Find $\begin{matrix} \boxed{X} \\ \nearrow \\ \boxed{T} \end{matrix}$

example:

$$r_A = -k C_A$$

$$V = \frac{F_{A0} X}{k C_A} = \frac{F_{A0} X}{A e^{-\frac{E_a}{RT}} C_A (1-X)} = f(T, X)$$



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* HW: Repeat the Design for $A + 2B \rightarrow C$ *

Design of steady state

December 10th, 2017

No. _____

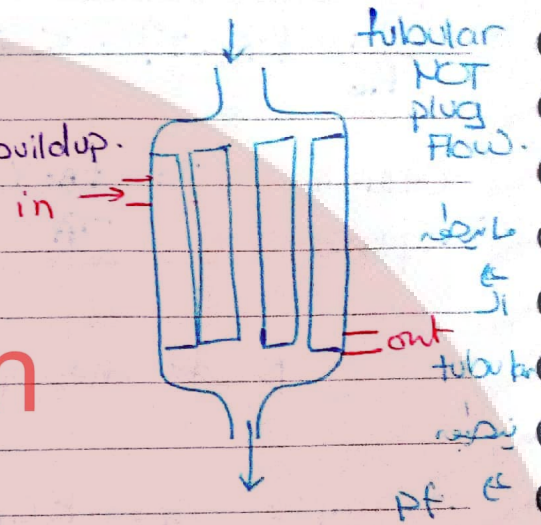
tubular reactor.

if exothermic $\Rightarrow$ going down $\uparrow$ ~~temp~~ Temp. buildup.
 $\therefore$ conversion $\downarrow$ with time

(X) (need it to stay constant.)
 $\Rightarrow$ and vice versa for endothermic.

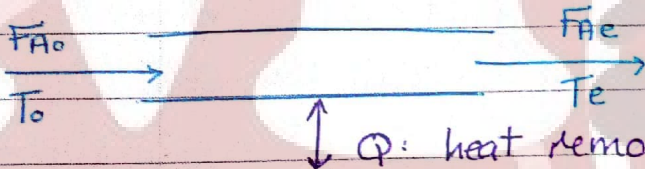
a Jacket is added.

to ~~de~~ maintain temp. constant.



- $\rightarrow$ length of tubes $\Rightarrow$ affects conversion. (optimum length) needed
- $\rightarrow$ optimise temp to achieve max. conversion
- $\rightarrow$ " no. of tubes to achieve " "

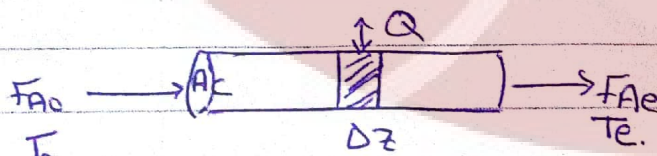
$\Rightarrow$ Ratio between no. of tubes and their length.



$\Rightarrow$ need an expression

energy balance :

$$\dot{Q} - F_{A0} \int_{T_0}^T \sum Q_i c_{pi} dT - \left[\Delta H_R^\circ (T_R) + \int_{T_R}^T \Delta c_p dT \right] F_{AX} = 0$$



$$\{\Delta V = A_c \times \Delta z\}$$

→ with respect to length.

December 10th

No.

→ variation of T/x with volume

$$\frac{dQ}{dv} - F_{A0} \sum Q_i C_{p_i} \frac{dT}{dv} - \left[\Delta H_R^\circ(T_R) F_{A0} \frac{dx}{dv} + \Delta C_p F_{A0} x \frac{dT}{dv} + \left(F_{A0} \int_{T_R}^T \Delta C_p dT \right) \frac{dx}{dv} \right] = 0$$

$$\dot{Q} = UA(T_a - T)$$

$$a = A/\Delta V \Rightarrow a = \frac{\pi D L}{\frac{\pi D^2 L}{4}} = \frac{4}{D}$$

unit area / unit volume

نسبة
بالنسبة
للتحريك

$$\frac{dQ}{dv} = Ua(T_a - T)$$

⌘ Rearrange: $\frac{dT}{dv} = Ua(T_a - T) + \left(F_{A0} \frac{dx}{dv} \right) \left[-\Delta H_R^\circ(T_R) - \int_{T_R}^T \Delta C_p dT \right]$

$$F_{A0} (\sum Q_i C_{p_i} + x \Delta C_p)$$

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Design equation: $-\frac{F_{A0} dx}{dv} = r_A$

$$\frac{dT}{dv} = Ua(T_a - T) \left(\frac{-r_A}{F_{A0}} \right) \left[-\Delta H_R^\circ(T_R) - \int_{T_R}^T \Delta C_p dT \right]$$

$$F_{A0} (\sum Q_i C_{p_i} + x \Delta C_p)$$

f(x, T)

couple this equation with other equation.

(II unknowns) (II equation).

other equation: $\frac{dx}{dv} = \frac{-r_A}{F_{A0}}$

Packed bed flow reactor Design

⇒ catalyst.

⇒ find the rate with respect to mass of catalyst.

(*) Ratio between (mass/volume) catalyst. ⇒ DENSITY.

$$\rho = \frac{m}{V} \Rightarrow V\rho = m$$

$$\rho dV = dm$$

$$\boxed{dV = \frac{dm}{\rho}}$$

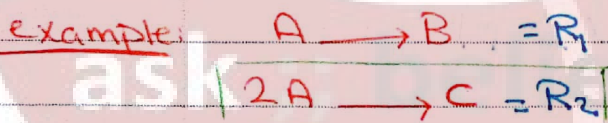
energy equation :

$$\frac{dT}{dw} = \frac{U_a(T_a - T)}{U_b \sum_{k=1}^n F_k C_{P_k}} + \sum_{i=1}^{n_{RXN}} (-r_{ij}) (-\Delta H_{Rij}(T))$$

k: no. of components.
w: weight of catalyst.
b: bulk

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$$-\frac{1}{2} \frac{dC_A}{dw} = \frac{dC_C}{dw}$$



$$\left[\begin{array}{l} r_1 A = -k_1 C_A \quad | \quad r_1 B = k_1 C_A \\ r_2 A = -k_2 C_A^2 \quad | \quad r_2 C = \frac{1}{2} k_2 C_A^2 \end{array} \right]$$

$$\frac{dT}{dw} = \frac{U_a}{U_b} (T_a - T) + \left[(-r_1) (\Delta H_{R1}^0(T_R)) + \int_{T_R}^T \Delta C_p dT \right]_{R_1}$$

$$+ \left[(-r_2) (\Delta H_{R2}^0(T_R)) + \int_{T_R}^T \Delta C_p dT \right]_{R_2}$$

$$F_A C_{P_A} + F_B C_{P_B} + F_C C_{P_C}$$

net rxn.

$$\left[\begin{array}{l} r_A = -k_1 C_A - k_2 C_A^2 \\ r_B = k_1 C_A \\ r_C = \frac{1}{2} k_2 C_A^2 \end{array} \right]$$

Δ VAPOR PHASE REACTION.

December 10th

No.

$$\rightarrow F_A = F_{A0}(1-X)$$

$$r_i = f(C_i)$$

Design equation..

$$\frac{dF_A}{dw} = r_A$$

$$C_A = C_{T0} \frac{F_A}{F_T} \frac{T_0}{T} \frac{P}{P_0} f(T, P, X)$$

(eqn for comp)

$$\frac{dF_B}{dw} = r_B$$

$$C_B = C_{T0} \frac{F_B}{F_T} \frac{T_0}{T} \frac{P}{P_0} f(T, P, X)$$

$$\frac{dF_C}{dw} = r_C$$

$$C_C = C_{T0} \frac{F_C}{F_T} \frac{T_0}{T} \frac{P}{P_0} f(T, P, X)$$

⋮

N-comp.

III equation III unknowns
 $\Rightarrow$ need to resolve with an equation

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Energy equation

$$\frac{dT}{dw} = \frac{V_a(T_a - T)}{V_b} + \frac{\sum_{i=1}^n (1 - \nu_{ij}) (-\Delta H_{Rij}(T))}{\sum_{i=1}^{n_{comp}} F_i C_{pi}}$$

December 12

$\Rightarrow$ Pressure drop equation

$$\frac{dP}{dw} = -\frac{\alpha}{2} \frac{T}{T_0} \frac{P_0}{(P/P_0)} \frac{F_T}{F_{T0}}$$

$$\alpha = \frac{2R_0}{A_c \rho_c (1-\phi) P_0}$$

$$R_0 = \frac{G(1-\phi)}{\rho_0 g_c D_p \phi^3} \left[\frac{150(1-\phi)\mu}{\Delta P} + 1.75G \right]$$

$$W = (1-\phi) A_c Z \times P_c$$