

PERPARED BY:

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DR:

Motasem Saidan







Particle 1. in in les in les
and acteristics
The three most important characteristics of an individual particle are its:
Particle Composition: which determines such properties as density and conductivity, provided that the particle is completely uniform. In many cases, however, the particle is porous or it may
that the particle is completely uniform. In many cases, however, the particle is porous or it may
consist of a continuous matrix in which small particles of a second material are distributed. Tange!! no fixed Particle size!! Particle Size: which affects properties such as the surface per unit volume and the rate at which
a particle will settle in a fluid diameter.
Particle Shape: which may be regular, such as spherical or cubic, or it may be irregular as, for
example with a piece of broken glass. Regular shapes are capable of precise definition by
mathematical equations. موامع کیوب
Departions systems of particles include storage in hoppers, flow through orifices and pipes, and metering of flows. Lesign with using math! and solve + roubleshooting.
It is frequently necessary to reduce the size of particles, or alternatively to form them into
aggregates or sinters. row materials have to be crushing or enlargment
Sometimes it may be necessary to mix two or more solids, and there may be a requirement to
separate a mixture into its components or according to the sizes of the particles.
20 h so bloom Title the Plan of Solid so we
have to Characterize the particle so we can predict the
that could take me

.

Flow rate! و ایم التدبند و ایم التدبند و ایم التدبند و ایم التدبیر و ای Single ier particles The simplest shape of a particle is the sphere (because of its symmetry) lia Sphere particle looks exactly the same from whatever direction it is viewed and behaves in Sphere particle looks exactly the same from whatever and behaves the same manner in a fluid, irrespective of its orientation.

Frequently, the size of a particle of irregular shape is defined in terms of the size of an analysis of the si tar ea equivalent sphere although the particle is represented by a sphere of different sphere according to the property selected. Some of the important sizes of equivalent spheres are: (a) The sphere of the same volume as the particle. according to (b) The sphere of the same surface area as the particle. gour objective (c) The sphere of the same surface area per unit volume as the particle. (d) The sphere of the same area as the particle when projected on to a plane perpendicular to its direction of motion. المستقل بعدما بيقة لل بيعة المستقل ال Jying in its position of maximum stability such as on a microscope slide for example.

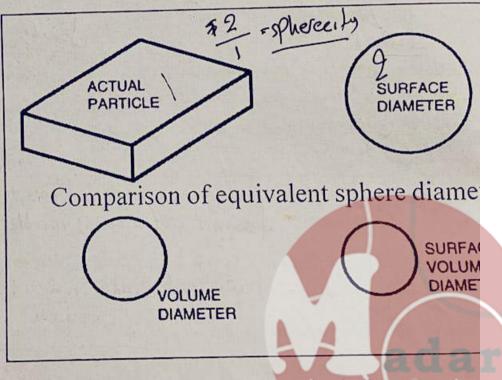
(f) The sphere which will just pass through the same size of square aperture as the particle, such as on a screen for example. (g) The sphere with the same settling velocity as the particle in a specified fluid. settle - cirin vila si./ so la si particle. 61 (int) & sire intrace area JI ze)

Describing the size of a single particle

Regular-shaped particles

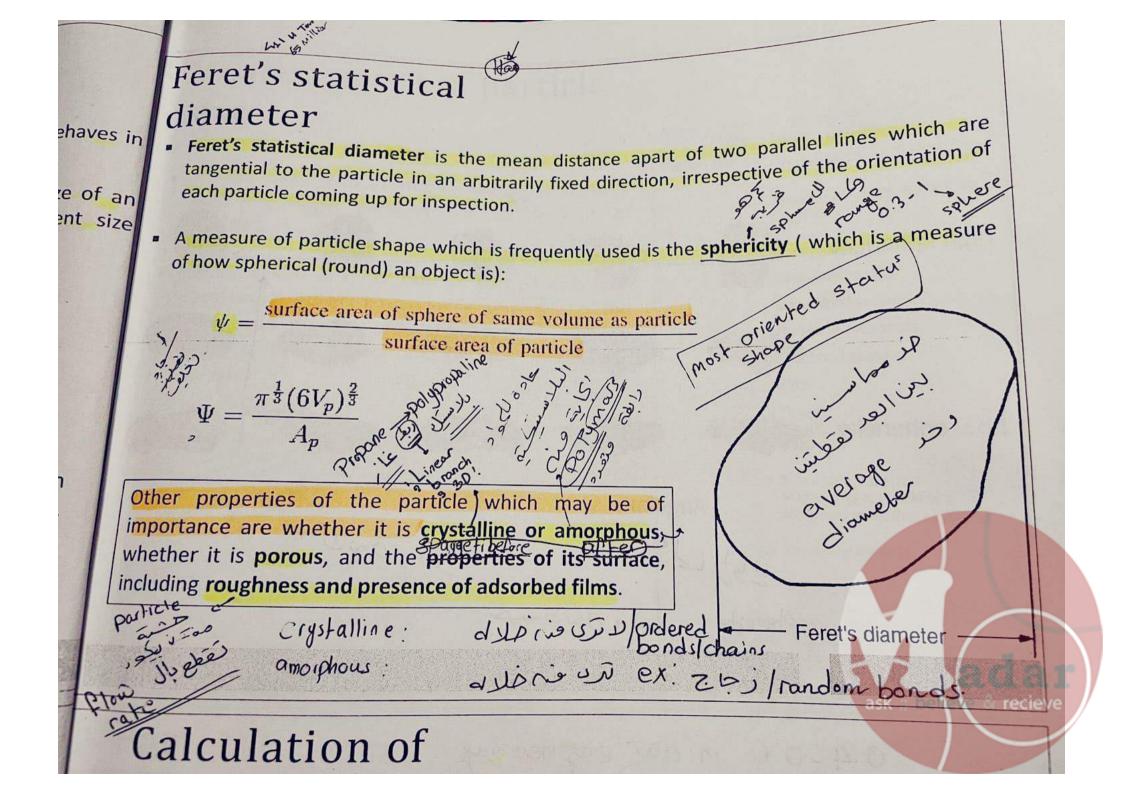
Shape	Sphere	Cube	Cylinder	Cuboid	Cone
Dimensions	Radius	Side length	Radius and height	Three side lengths	Radius and height

- The orientation of the particle on the microscope slide will affect the projected image and consequently the measured equivalent sphere diameter.
- Sieve measurement: Diameter of a sphere passing through the same sieve aperture.
- Diameter of a sphere having the same sedimentation velocity under the same conditions.



Liqui

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Calculation of Sphericity

Sphericity Φ_s is defined as $\Phi_s = \frac{6V_p}{D_p A_p}$ where V_p is the volume of the object, A_p is its surface area, and D_p is the diameter of a sphere with the same volume $(\pi D_p^3/6)$.

For a sphere of diameter d, $V_p = \pi d^3/6$, $A_p = \pi d^2$, $D_p = d$, and so $\Phi_s = 1$.

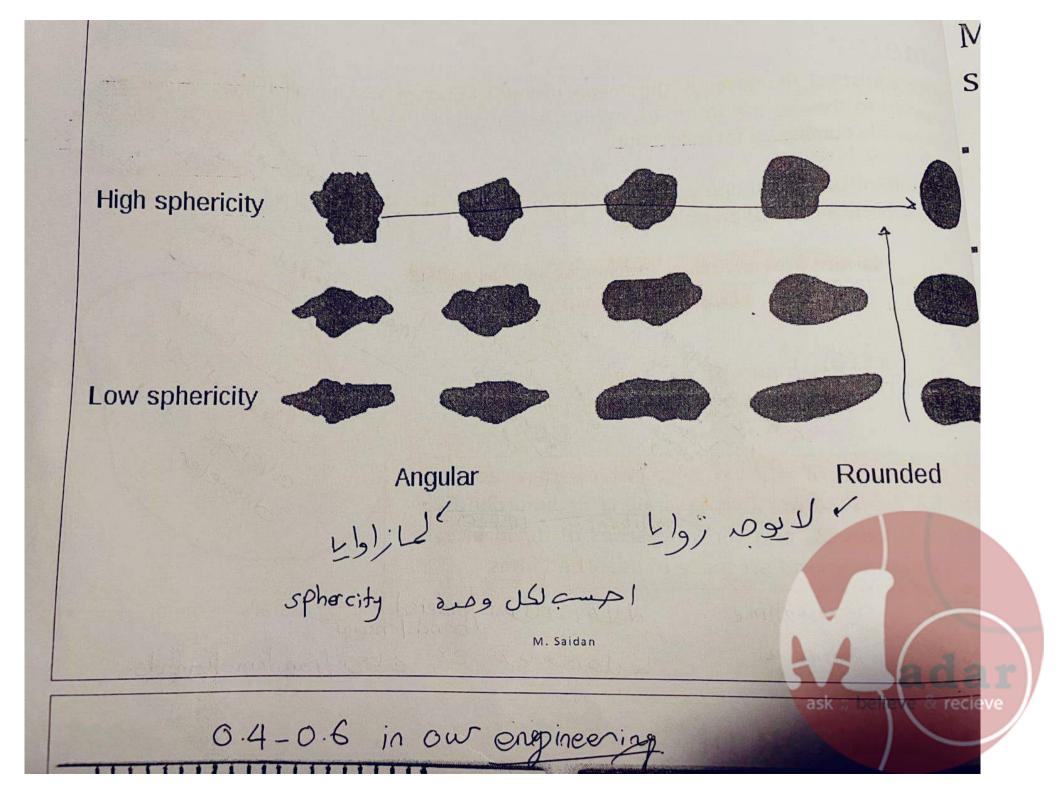
For a cylinder of diameter d and length L, $V_p = \pi d^2L/4$, $A_p = \pi dL + \pi d^2/2$, $D_p = (6d^2L/4)^{1/3}$,

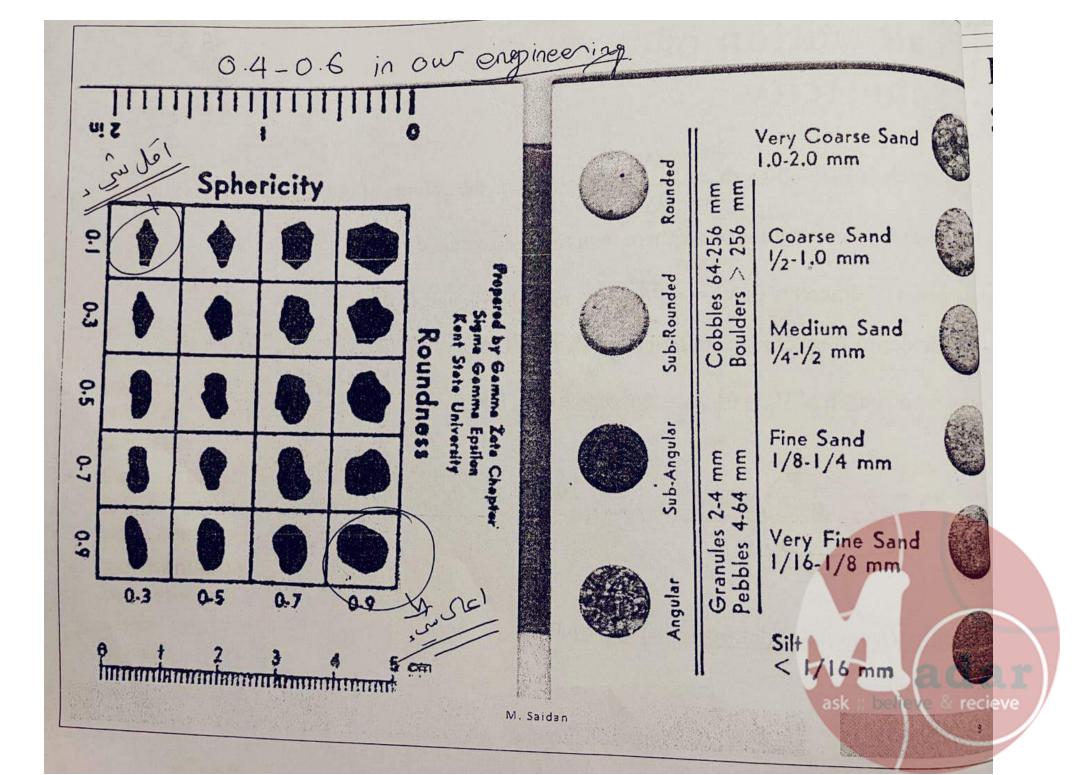
For a cube of width a, $V_p = a^3$, $A_p = 6a^2$, and $c = a^3$, and so

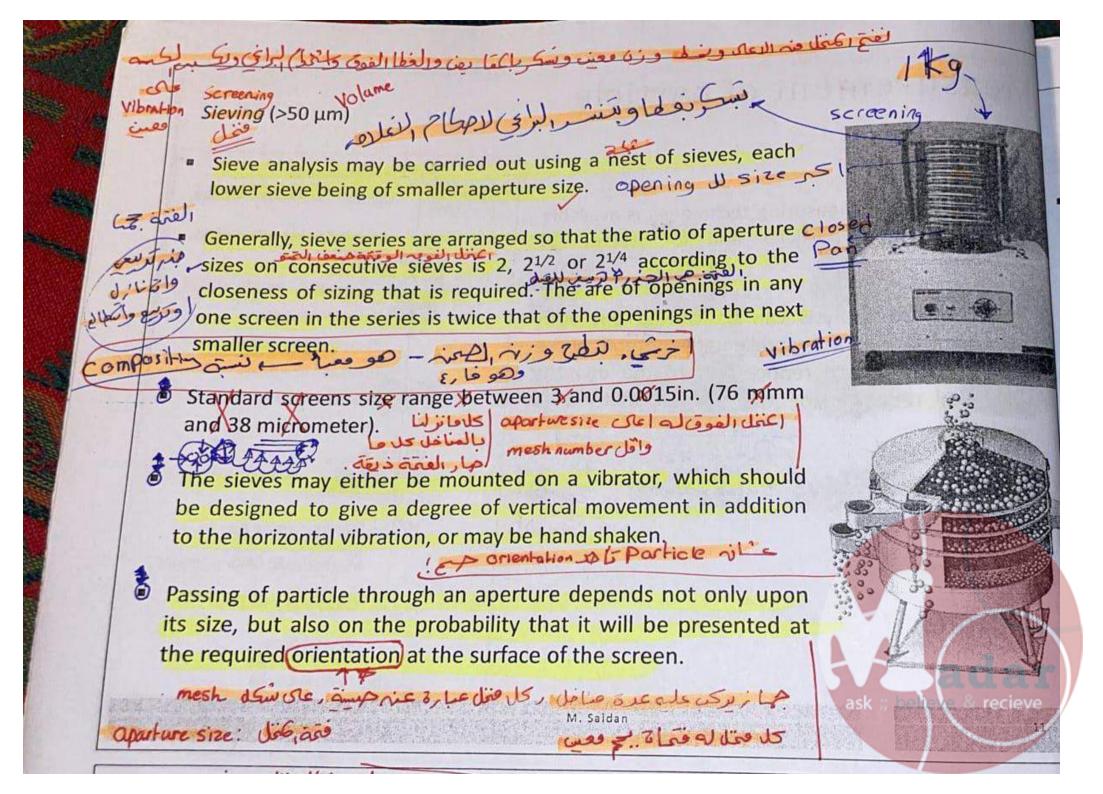
$$\Phi_{\rm S} = \frac{6a^3}{(6/\pi)^{(1/3)}a(6a^2)} = \left(\frac{\pi}{6}\right)^{(1/3)} = 0.806$$

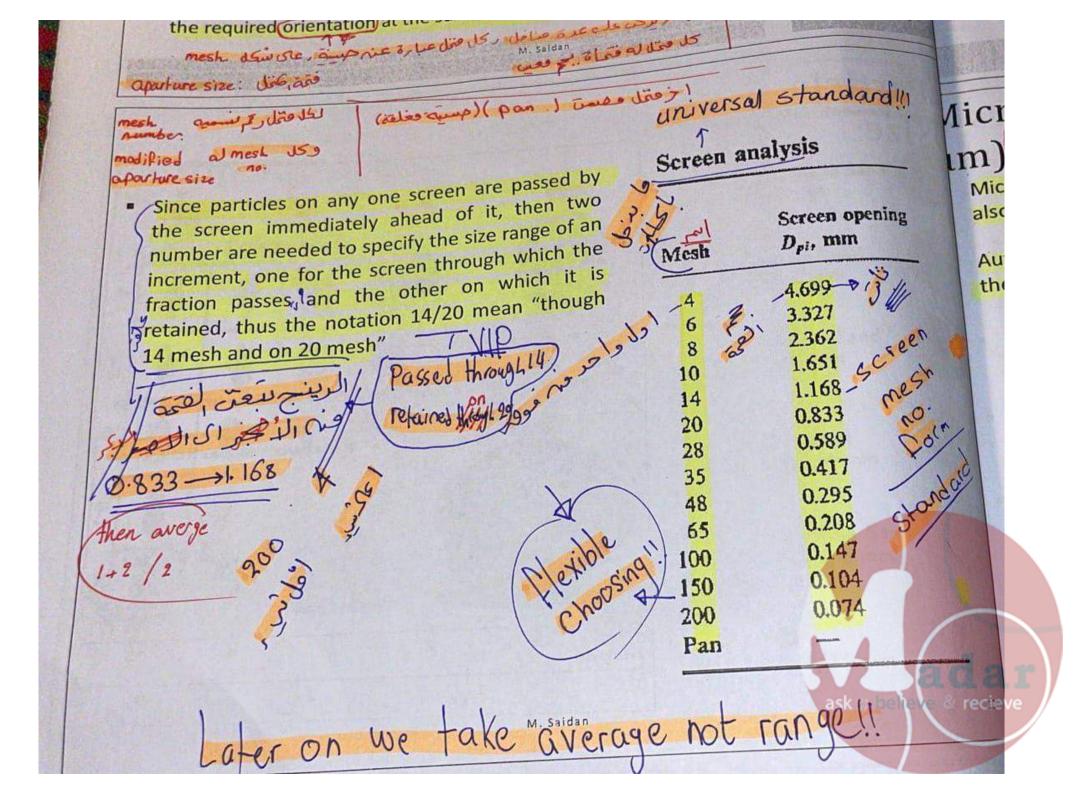
http://en.wikipedia.org/wiki/Sphericity





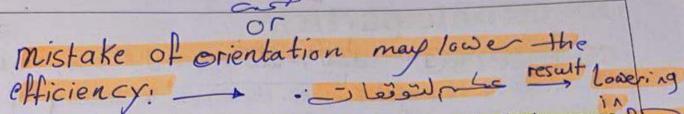






Screenin

g



The efficiency of screening is defined as the ratio of the mass of material which passes باماد الحيصر القل عني المادة الحيصر القل عني المادة الحيصر القل عني المادة الماد

وتوقع اغا عر

This will differ according to the size of the material.

Capable No val Justin

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It may be assumed that the rate of passage of particles of a given size through the screen is proportional to the number or mass of particles of that size on the screen at any instant

Thus if w is the mass of particles of a particular size on the screen at a time t, then:

 $\frac{\mathrm{d}w}{\mathrm{d}t} = -kw$

where k is a constant for a given size and shape of particle and for a given screen. Thus, the mass of particles $(w_1 - w_2)$ passing the screen in time t is given by:

$$\ln \frac{w_2}{w_1} = \sqrt{kt}$$

$$w_2 = w_1 e^{-kt}$$

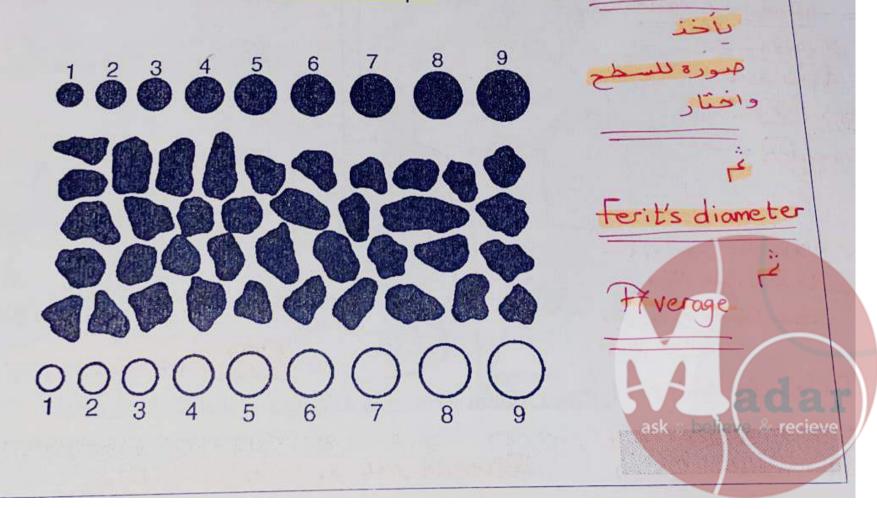
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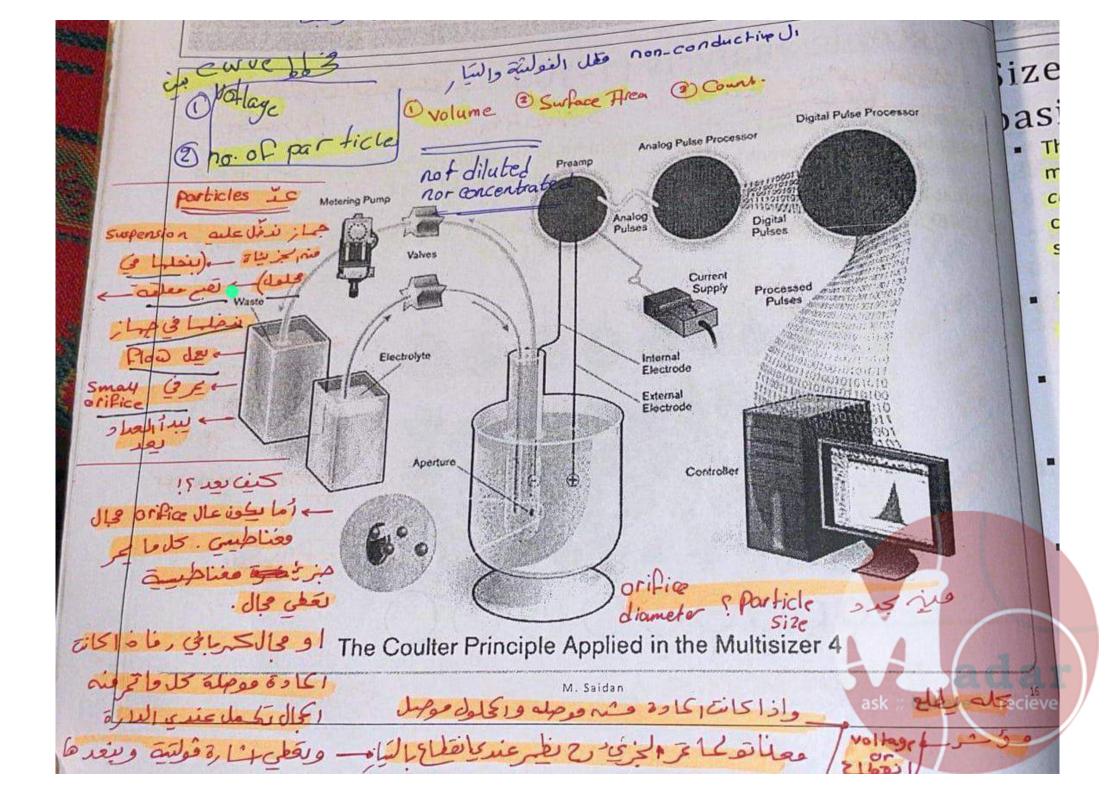
or:

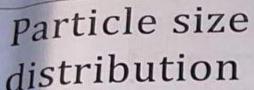
Microscopic analysis (1-100 معنة عندة العربات (1-100 μm) العربات الع

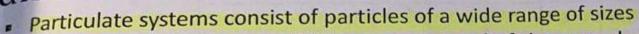
Microscopic examination permits measurement of the projected area of the particle and also enables an assessment to be made of its two-dimensional shape.

Automatic methods of scanning have been developed. By using the electron microscope, the lower limit of size can be reduced to about 0.001 μm.









A quantitative indication of the mean size and of the spread of sizes of particulate systems should be given and represented by means of: _____ 10% hove aport size from _____ to

√ a cumulative mass fraction curve,

Seives والمنسوال x in which the proportion of particles (x) smaller than a certain size (d) is plotted against that size (d).

In most practical determinations of particle size, the size analysis will be obtained as a series of steps, each step representing the proportion of particles lying within a certain small range of size.

From these results a cumulative size distribution can be built up and this can then be approximated by a smooth curve provided that the size intervals are sufficiently small.

Size distribution curve — cumulative

basis X: cumulative.

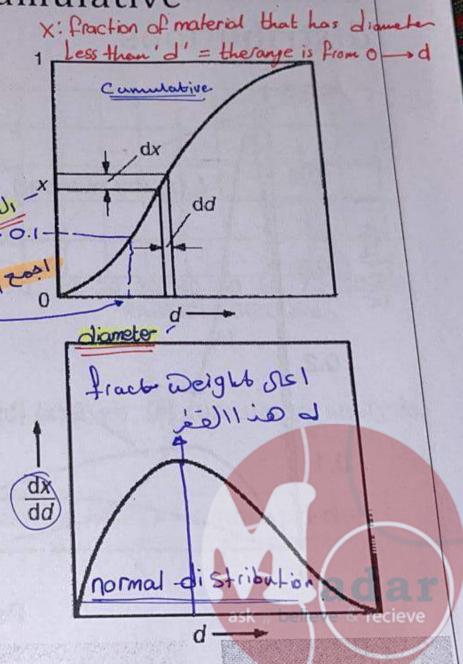
The distribution of particle sizes can be seen more readily by plotting a size frequency curve, in which the slope (dx/dd) of the cumulative curve is plotted against particle size (d). العنبة لما يج اعل فنه عنه الما العنبة لما يج اعل فنه عنه العنبة لما يج اعلى فنه عنه العنبة لما يعنبه العنبه العنبة لما يعنبه العنبه العنبة لما يعنبه العنبه ا

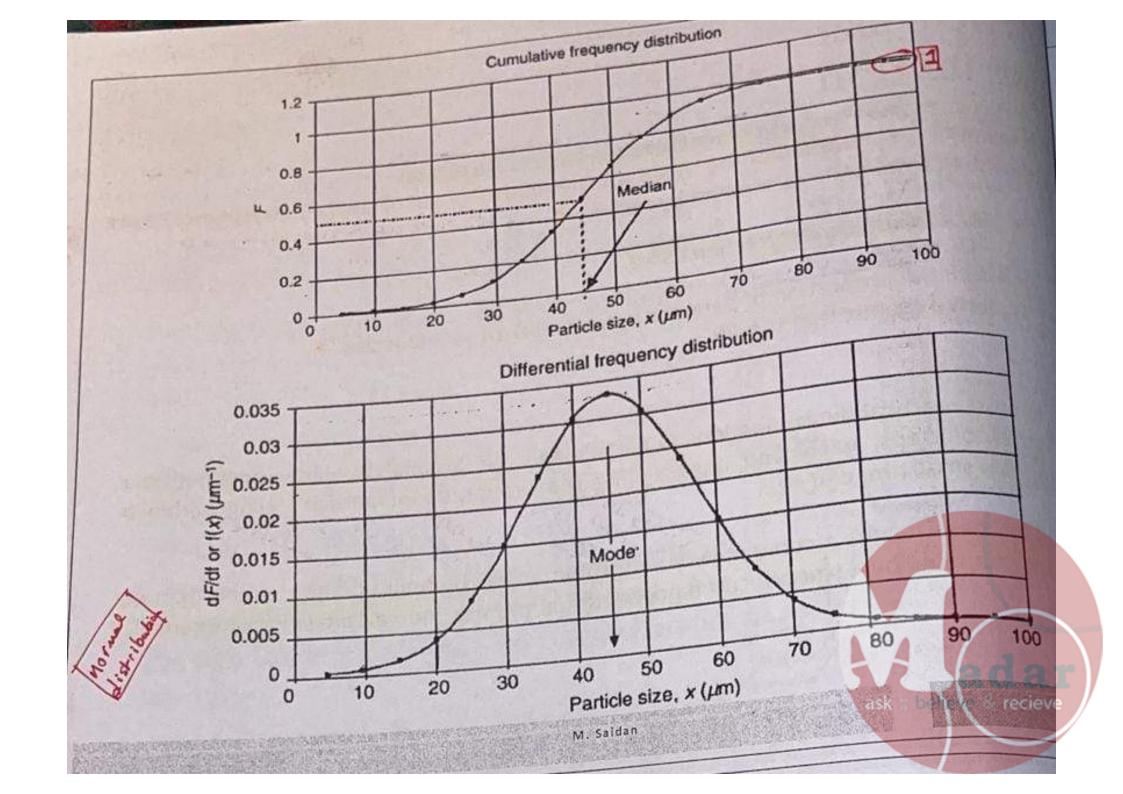
The most frequently occurring size is then shown by the maximum of the curve.

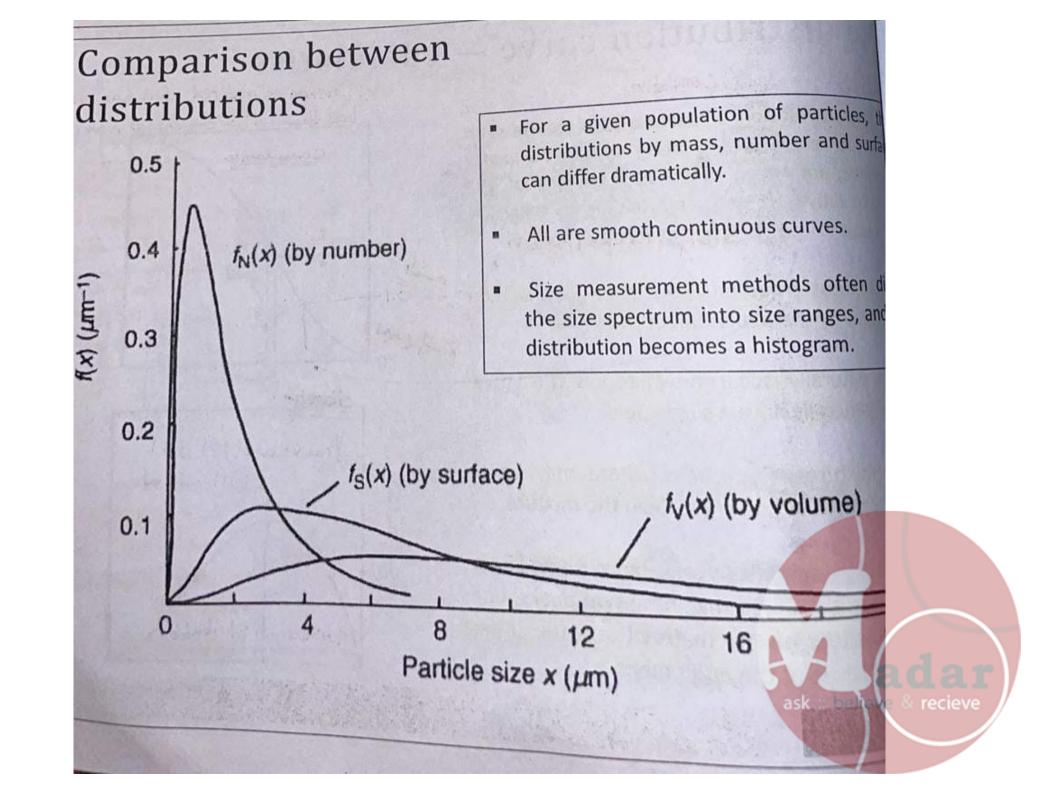
For naturally occurring materials the curve will generally have a single peak.

For mixtures of particles, there may be as many peaks as components in the mixture.

If the particles are formed by crushing larger particles, the curve may have two peaks, one characteristic of the material and the other characteristic of the equipment.







Mixed particle size and size analysis

- In a sample of uniform particles of diameter D_p the total volume of the particles is m/ρ_p
- The volume of one particle is v_p , the number of particles in the sample N is

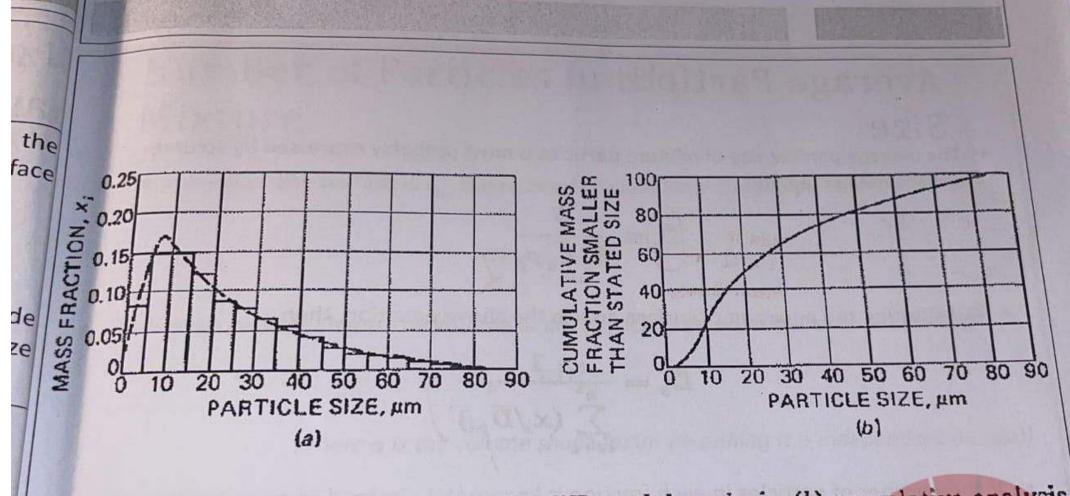
for ticlos is:
$$N = \frac{m}{\rho_p v_p}$$
for ticlos is:
$$mues = \rho_p v_p$$

The total surface area of particles is:

articles is:
$$A = Ns_p = \frac{6m}{\Phi_s \rho_p D_p}$$

$$Specific surface area uniform
one particle
ons to mixtures of particles having various sizes a$$

Now to apply these equations to mixtures of particles having various sizes and densities, the mixtures should be screened and sorted into fractions (each of constant density and constant size).



Particle-size distribution for powder: (a) differential analysis; (b) cumulative analysis.

✓ Cumulative analysis is more precise

Specific Surface Area of Mixture

If particle density and sphericity are known in each fraction, the surface area of particles is calculated by:

$$A_{iv} = \frac{6x_1}{\Phi_s \rho_d \bar{D}_{p1}} + \frac{6x_2}{\Phi_s \rho_p \bar{D}_{p2}} + \dots + \frac{6x_n}{\Phi_s \rho_p \bar{D}_{pn}}$$

$$= \frac{6}{\Phi_s \rho_p} \sum_{i=1}^n \frac{x_i}{\bar{D}_{pi}} \frac{\text{Column } 3}{\text{Column } 4}$$

where subscripts = individual increments

 $x_i =$ mass fraction in a given increment

n = number of increments

 \bar{D}_{pi} = average particle diameter, taken as arithmetic average smallest and largest particle diameters in increment

Average Particle

■ The average particle size of mixture particles is most probably expressed by volumesurface mean diameter

volume
$$\bar{D}_s \equiv \frac{6}{\Phi_s A_w \rho_p}$$

Substituting the equation of surface area in the above equation, then

$$\overline{D}_s = \frac{1}{\sum_{i=1}^{n} (x_i/\overline{D}_{pi})}$$

If the number of particles in each fraction is known (N_i) , instead of mass fraction, then the arithmetic mean diameter is used:

$$\bar{D}_N = \frac{\sum\limits_{i=1}^{N} (N_i \bar{D}_{pi})}{\sum\limits_{i=1}^{n} N_i} = \frac{\sum\limits_{i=1}^{N} (N_i \bar{D}_{pi})}{N_T}$$
 in

Where the (N_T) is the number of particles in the entire sample

The mass mean diameter is calculated by:

$$\bar{D}_{w} = \sum_{i=1}^{n} x_{i} \bar{D}_{pi}$$

Dividing the total volume of the sample by the number of particles in the mixture gives the average volume of a particle. The diameter of such a particle is the volume mean diameter which is calculated by:

$$\bar{D}_V = \left[\frac{1}{\sum\limits_{i=1}^n (x_i/\bar{D}_{pi}^3)}\right]^{1/3}$$

April James 105



Number of Particles in the Mixture

The volume of one particle is v_p , the number of particles in the sample N is

$$N = \frac{m}{\rho_p v_p}$$

For a given particle shape, the volume of any particle is proportional to its (diameter)³

$$v_p = aD_p^3$$

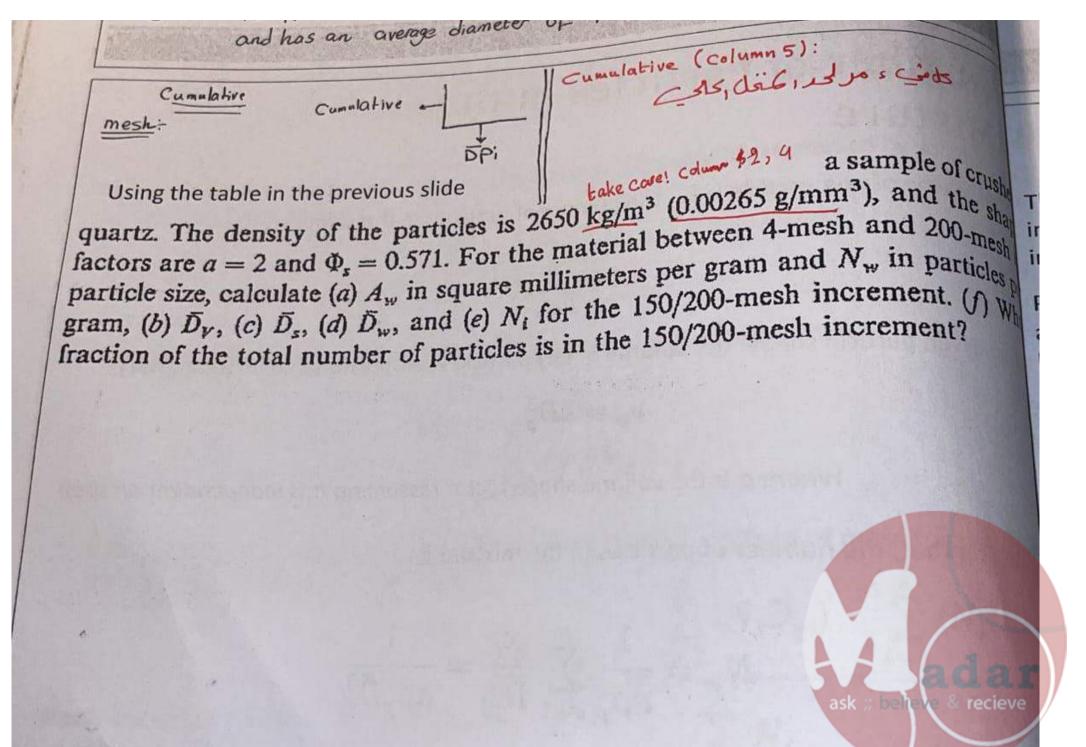
Where a is the volume shape factor (assuming it is independent on size))

Based on that, the number of particles in the mixture is:

$$N_{w} = \frac{1}{a\rho_{p}} \sum_{i=1}^{n} \frac{x_{i}}{\bar{D}_{pi}^{3}} = \frac{1}{a\rho_{p}\bar{D}_{v}^{3}}$$



	assessment	not standard Dat	Storbed: we take any ast range of curn fractions	ular
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Particulate Solid in behavior of solid particles in bulk! Storage in tanks.

The properties of solids in bulk are a function of the properties of the individual particles including their shapes and sizes and size distribution, and of the way in which the particles interact with one another.

 Particulate solids present considerably greater problems than fluids in storage, in removal at a controlled rate from storage, and when introduced into vessels or reactors where they become involved in a process. Liquid, gas ais as a

spaces when two solid particles face each other · One of the most important characteristics of any particulate mass is its voidage (the fraction of the total volume which is made up of the free space between the particles and is filled Porousity: within a particle high sulface ofen size! with fluid.) Free volume

bettween solid porticle Low voidage ==== high density of packing of the particles. مارسم للفلوبد باكرورمة فلاله. voidage decreases with lowering particlesize

 The way in which the particles pack depends not only on their physical properties, including shape and size distribution, but also on the way in which the particulate mass has been کل وحدہ کنفترست تی ہے introduced to its particular location. orientation

المستقرار الجذي ولي وور تحير في ال المستقرار الجذي وليب وور تحير في ال المستقرار الجذي وليب وور تحير في ال المستقرار الجذي وليب والمستقرار الجذي وليب وور تحير في ال المستقرار الجذي وليب وور تحير في ال المستقرار الجذي وليب وور تحير في المستقرار الجذي وليب وور تحير في المستقرار المجذوب والمستقرار المستقرار المستقر dense

> The more rapidly material is poured on to a surface or into a vessel, the more densely

will it pack.

Isometric, spherical : packing more, density high

Low symmetry, Low sphericity: packing low, density low!!

Usperily I Low packing

shed

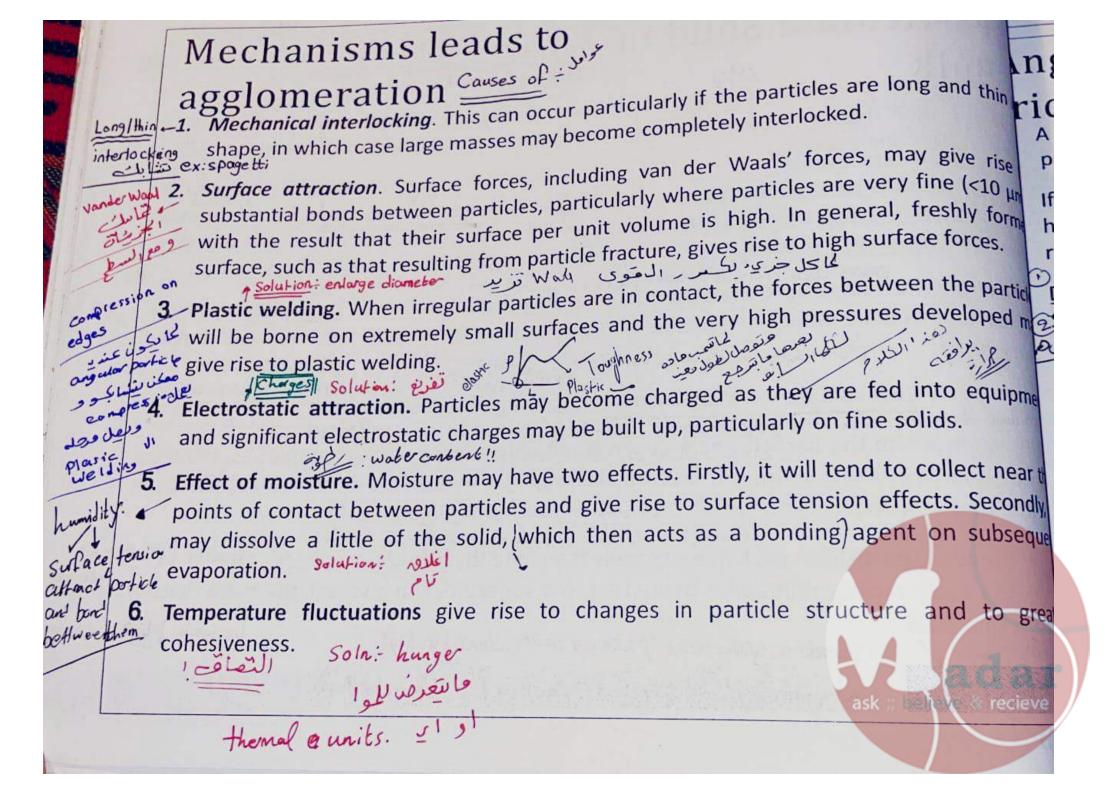
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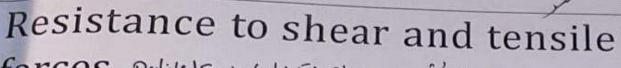
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Agglomeration of particles: Publicle jesses and it you have to do force on Re Agglomeration arises from interaction between particles, as a result of which they adh If the particles tend to agglomerate, poor flow properties may be expected. If a significant amount of the material is in the form of particles smaller than 10 µm, the particles deviate substantially from isometric form, it may be inferred that the characteristics will be poor. siggeregation | 015 81 وابعن المحرا لخلط لعن كالمرافع المحرا لخلط المرافع المحرا المرافع المحرا المحر المحرا المحرا





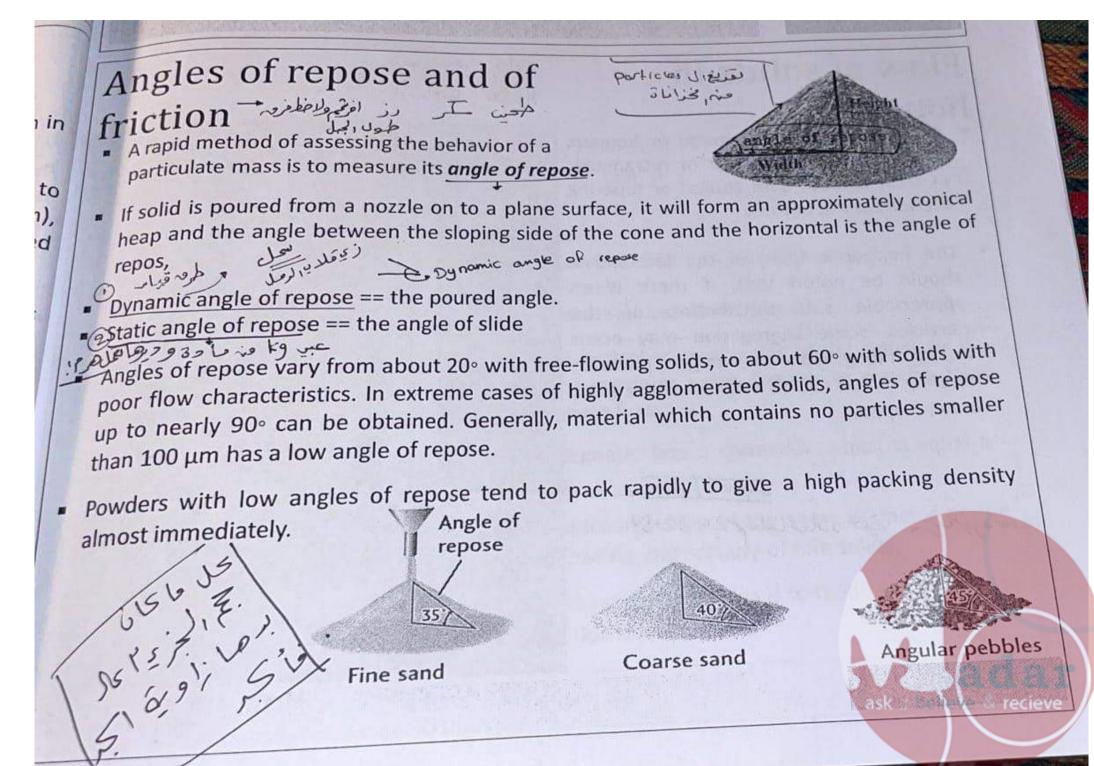
forces Particulate in shear into bulk sine in sheld

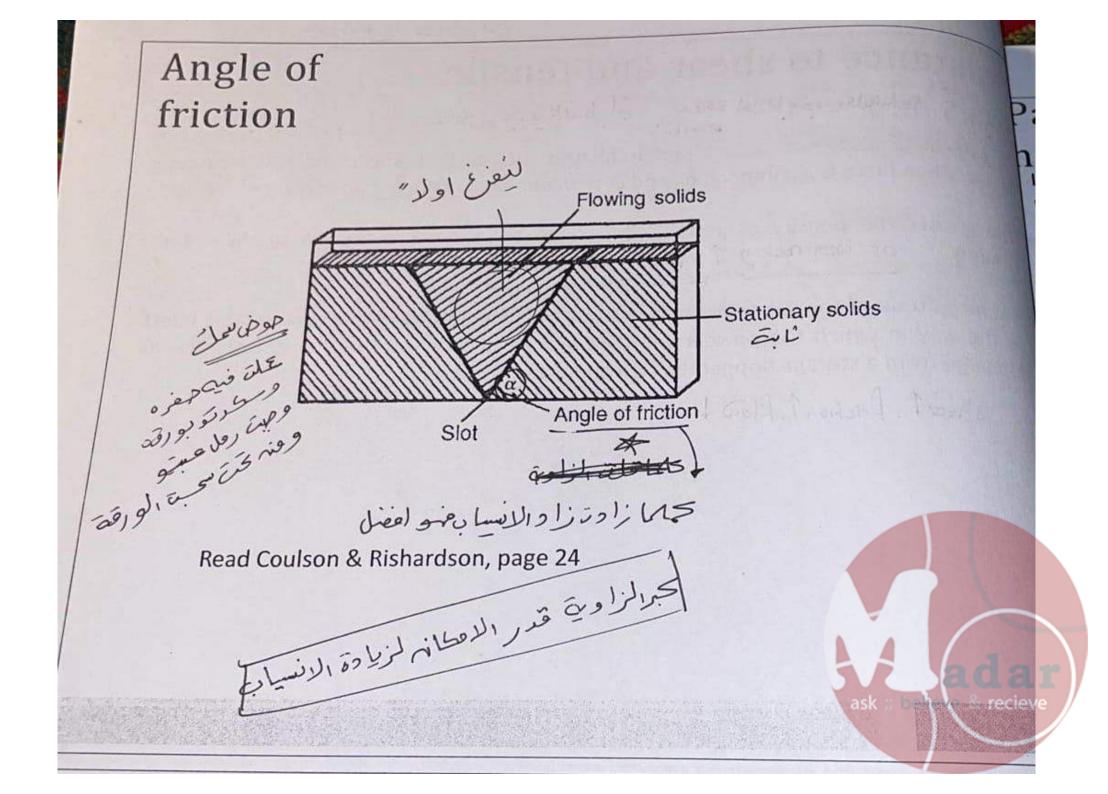
- A particulate mass may offer a significant resistance to both shear and tensile forces, either when there is agglomeration and even in non-agglomerating powders,
- The greater the density of packing, the higher will be this resistance to shear and tension.

 as well packing 1, Plan 1, resistance 1
- The magnitude of the shear and tensile strength of the powder has a considerable effect on the way in which the powder will flow, and particularly on the way in which it will discharge from a storage hopper through an outlet nozzle.

shear 1, Priction 1, Flow 1







Flow of solids in

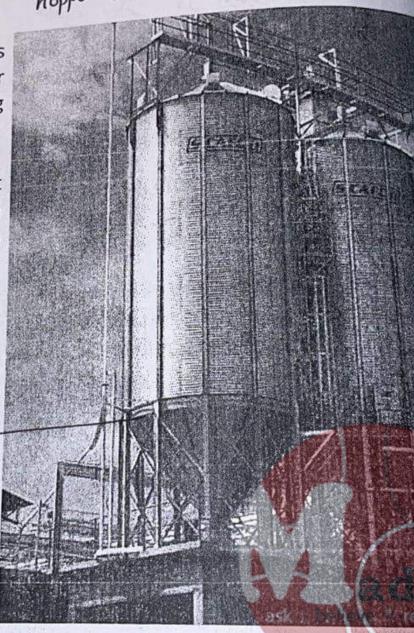
hoppers

Frequently, solids are stored in hoppers which are usually circular or rectangular in cross-section, with conical or tapering sections at the bottom.

The hopper is filled at the top and it should be noted that, if there is an appreciable size distribution of the particles, some segregation may occur during filling with the larger particles tending to roll to the outside of the piles in the hopper.

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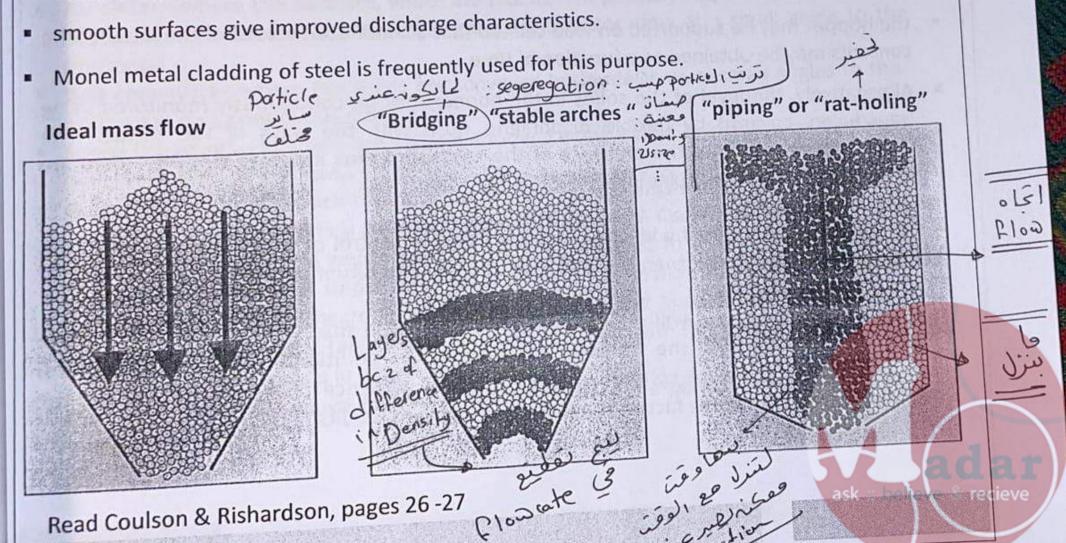
ارتفاع عام : Cylo: ارتفاع الله الم



Particles discharge from the

hopper المان عند Vessel منه كالم بمعلق الم

In general, tall thin hoppers give better flow characteristics than short wide ones and the use of long small-angle conical sections at the base is advantageous.



Flow of solids through

orifices

$$G = \frac{\pi}{4} \rho_s d_{\text{eff}}^{2.5} g^{0.5} \left(\frac{1 - \cos \beta}{2 \sin^3 \beta} \right)^{0.5}$$

where: G is the mass flowrate,

 ρ_s is the density of the solid particles.

deff is the effective diameter of the orifice orifice particle diameter

g is the acceleration due to gravity, and

 β is the acute angle between the cone wall and the horizontal.

air

- alid particles is usually the Limiting rate: botherick The discharge rate of solid particles is usually controlled by the size of the orifice or the aperture at the base of the hopper, though sometimes screw feeders or rotating table feeders may be incorporated to encourage an even flowrate. ripic is also le as le piston.
- The effective diameter is the actual orifice diameter less a correction which is equal to between 1 and 1.5 times the particle diameter - effective diameter = actual x 1.5
- It has been found that the attachment of a discharge pipe of the same diameter as the orifice immediately beneath it increases the flowrate, particularly of fine solids.
- Another method of increasing the discharge rate of fine particles is to fluidize the particles in the neighborhood of the orifice by the injection of air.

measure flow rate in siglo: بنركبوعاى مَدان للوزن بعدين نفرع جزدمنو وناخذ فرق ونقسمه عاى الوقت mass lost Heasurement and control of solids flowrate agglomeration (1): Flow to be is in - 13 cintar = 51 Con The flowrate of solids can be measured either as they leave the hopper or as they are soli conveyed. Grav the hopper may be supported on load cells so that a continuous record of the mass of the Air by ' contents may be obtained as a function of time. hor Alternatively, the level of the solids in the hopper may be continuously monitored using Be transducers covered by flexible diaphragms flush with the walls of the hopper. The ho diaphragm responds to the presence of the solids and thus indicates whether there are Sc he solids present at a particular level. BI The problems associated with the measurement and control of the flowrate of solids ar C much more complicated than those in the corresponding situation with liquids. The flow characteristics will depend, not only on particle size, size range and shape, by also on how densely the particles are packed. In addition, surface and electric properties and moisture content all exert a strong influence on flow behavior, and the combined effect of these factors is almost impossible to predict in advance. in - V come o Cure De JLCI Floorafell com Callibration curo es alien Cla asser Clare of radiation @ الكنافة بيتحول m

Methods for flow

Fitting an orifice plate at the discharge point from the hopper.

Using a belt-type feeder in which the mass of material on the belt is continuously measured by load cells for example or by a nuclear densitometer which measures the degree of absorption of gamma rays transmitted vertically through the solids on the belt which is travelling at a controlled speed.

Applying an impulse method in which a solids stream impacts vertically on a sensing plate orientated at an angle to the vertical. The horizontal component of the resulting force is measured by as load cell attached to the plate.

Flow Control

The rate of feed of solids may be controlled using screw feeders, rotating tables or vibrating feeders, such as magnetically vibrated troughs.

 Volumetric rates may be controlled by regulating the speeds of rotation of star feeders or rotary valves.

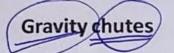
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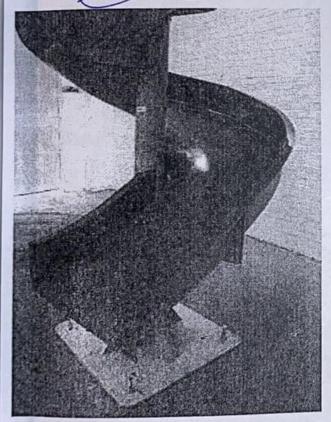
Conveying of

نقل المسط فه محانه الك اخر! ..

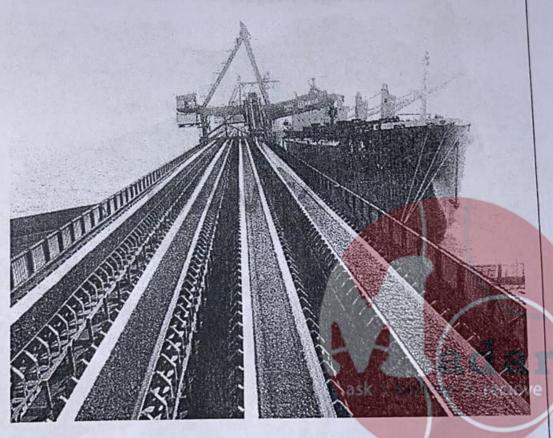
- Gravity chutes—down which the solids fall under the action of gravity.

 Air clides where the solids fall under the action of gravity.
- Air slides—where the particles, which are maintained partially suspended in a channel by the upward flow of air through a porous distributor, flow at a small angle to the horizontal.
- Belt conveyors—where the solids are conveyed horizontally, or at small angles to the horizontal, on a continuous moving belt.
- Screw conveyors—in which the solids are moved along a pipe or channel by a rotating helical impeller, as in a screw lift elevator. وهو لمالع بغزي بال ه الله عنون بال
- Bucket elevators—in which the particles are carried upwards in buckets attached to a continuously moving vertical belt (See Figure 1.16. page 29, Coulson & Richardson)
- Vibrating conveyors—in which the particles are subjected to an asymmetric vibration and travel in a series of steps over a table. During the forward stroke of the table the particles are carried forward in contact with it, but the acceleration in the reverse stroke is so high that the table slips under the particles. With fine powders, vibration of sufficient intensity results in a fluid-like behavior.
- Pneumatic/hydraulic conveying installations—in which the particles are transported in a stream of air/water. ◆ ●

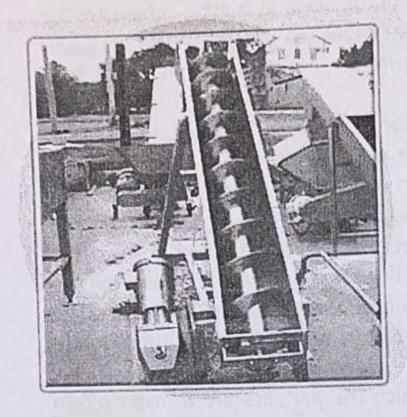


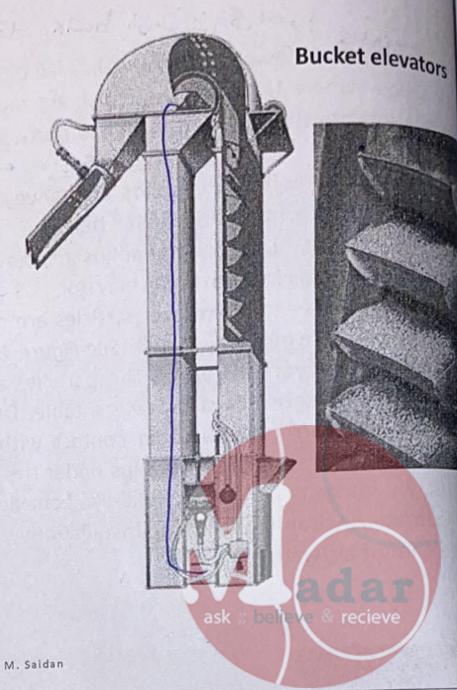


Belt conveyors

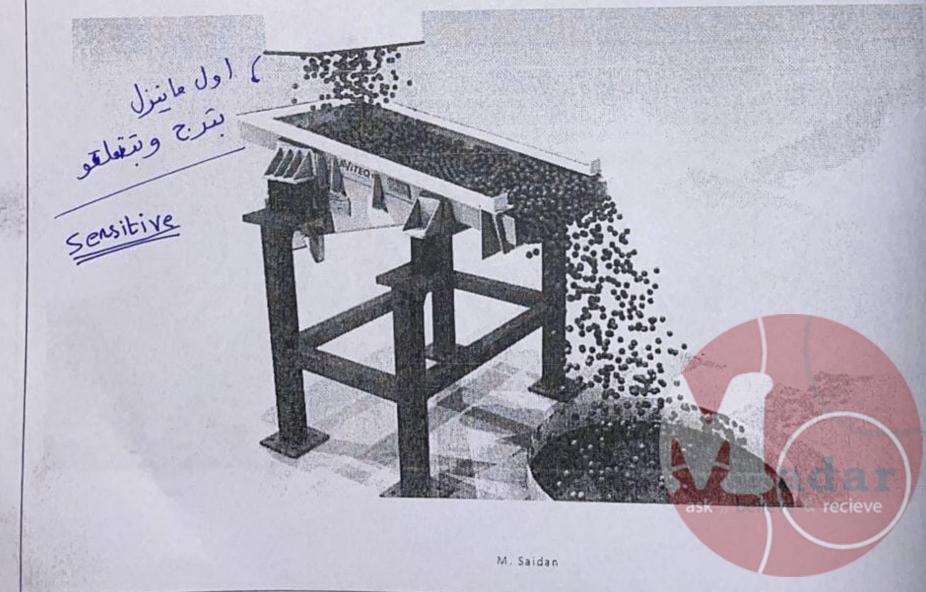


Screw conveyors





Vibrating conveyors



Mixing / blending.

How tomix/time?!

mechanisms

to prepare a mixture of solid particles+ ? Time needed to reach ideal mixing!

A. Convective mixing, in which groups of particles are moved from one position to another,

* Porticles of A is dominant!! so it diffuses!!

B. Diffusion mixing, where the particles are distributed vover a freshly developed interface,

A. Shear mixing, where slipping planes are formed Same Fine!!

ندمَّل حزيئاة واحل فراعات جزيبان أخره!!

يؤدي اي الهيكال

These mechanisms operate to extents in different kinds of mixers and with different kinds of particles.

A <u>trough mixer</u> with a ribbon spiral involves almost pure convective mixing, and

a simple barrel-mixer involves mainly a form of diffusion mixing.



The degree of mixing

Statistical analysis -> standard deviation that must be within the range!

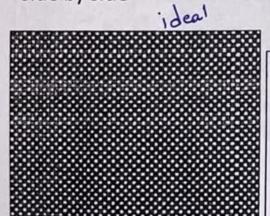
المعب تحد يوجودة It is difficult to quantify the degree of mixing, عب تحد يوجودة

Statistical variation in composition among samples withdrawn at any time from a mix i ي كل عانهل ماتش

commonly used as a measure of the degree of mixing of solid particles. باخدعينة: وقت : عداما وقت الما في عناه الوقة الوقة الما في الم

The standard deviation s (the square root of the mean of the squares of the individual deviations) or the variance s2 is generally used.

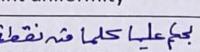
No perfect mixing: just a degree of randomness in which two similar particles may well be side by side



4 Connét acheive it !!

- No amount of mixing will lead to the formation of a uniform mosaic.
- > Just overall uniformity but not point uniformity

بحرعلها كلما فشرنقطة نقطة





- time has to be obtimized to obtain the best degree of mixing.

For a completely random mix of uniform particles:

Fully mixed
$$s_r^2 = \frac{p(1-p)}{n}$$
and the best $s_r^2 = \frac{p(1-p)}{n}$
Agreell

where s_r^2 is the variance for the mixture, p is the overall proportion of particles of one colour, and n is the number of particles in each sample.

However, in a completely unmixed system

unmixed at
$$t=0$$
 $\leftarrow s_0^2 = p(1-p)$

mixer identification c

which is independent of the number of particles in the sample



M Colden

When a material is partly mixed, then the degree of mixing may be represented by some term b

degree of mixing.
$$b = \frac{(s_0^2 - s_r^2)}{(s_0^2 - s_r^2)}$$

$$1 - b = \frac{(s^2 - s_r^2)}{(s_0^2 - s_r^2)}$$

- For diffusive mixing, **b** will be independent of sample size provided the sample is small.
- With convective mixing, 1 b depends on the size of the sample

Exampl

e

The performance of a solids mixer was assessed by calculating the variance occurring in the mass fraction of a component amongst a selection of samples withdrawn from the mixture. The quality was tested at intervals of 30 s and the data obtained are:

c ²		minimum Vorie	ince		
sample variance (-)	0.025	0.006	0.015	0.018	0.019
mixing time (s)	30	bex time!!!	90	120	150

If the component analyzed represents 20 per cent of the mixture by mass and each of the samples removed contains approximately 100 particles, comment on the quality of the mixture produced and present the data in graphical form showing the variation of the mixing index with time.

ask ;; belje 🙋 & recie

Solutio

n

For a completely unmixed system:

$$s_0^2 = p(1-p) = 0.20(1-0.20) = 0.16$$

For a completely random mixture:

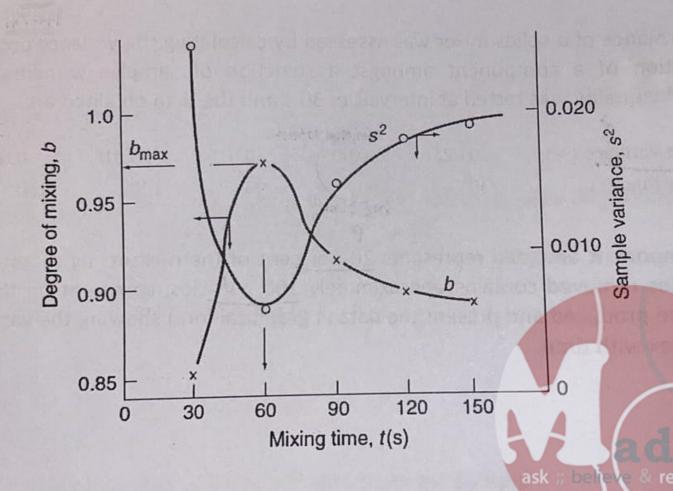
$$s_r^2 = p(1-p)/n = 0.20(1-0.20)/100 = 0.0016$$

The degree of mixing b is given by equation 1.35 as: $b = (s_0^2 - s^2)/(s_0^2 - s_r^2)$ In this case, $b = (0.16 - s^2)/(0.16 - 0.0016) = 1.01 - 6.313s^2$ The calculated data are therefore:

<i>t</i> (s)	30	60	90	120	150
$\frac{1}{s^2}$	0.025	0.006	0.015	0.018	0.019
b,	0.852	0.972	0.915	0.896	0.890

Plante avorage 259 12/2 (Sight fine





It is clear that the degree of mixing is a maximum at t = 60s

The rate of mixing

Mixing involves obtaining an equilibrium condition of uniform randomness, the relation between **b** and time might be expected to take the general form:

rate
$$-b = 1 - e^{-ct}$$

- (a) the total volume of the material/
- (b) the inclination of the drum,
- (c) the speed of rotation of the drum,
- (d) the particle size of each component,
- (e) the density of each component, and
- (f) the relative volume of each component.



Solid Particles Separation

Sedimentation.

- Separation depends on the selection of a process in which the behavior of the material significantly influenced by some physical property.
- Sieving method is used if a material is to be separated into various size fractions. Sing depends primarily on the size of the particles, also other physical properties such as shape of the particles and their tendency to agglomerate may also be involved.
- Separation depends on the differences in the behavior of the particles in a moving fland in this case the size and the density of the particles are the most important factors a shape is of secondary importance.
- Other separation processes make use of differences in electrical or magnetic properties
 the materials or in their surface properties.

or to extract Solids!

solids are removed from fluids in order to purify the fluid although, in some cases, and particularly with liquids, it is the solid material that is the product.

Separation processes are:

Precepitation: Chemical reaction results in product that isn't soluble so by gravity it setters down > Sedimentation, in which the solids are allowed to settle by gravity through the the settle liquid from which they are removed, usually as a pumpable sludge.

, filterpaped mesh Filtration, in which the solids are collected on a medium, such as a porous material or a layer of fine particles, through which the liquid is pumped.

الطرد المركزي!

Centrifugal separation in which the solids are forced on to the walls of a vessel which is rotated to provide the centrifugal force.

CAPECS+opecs.

اذا عندل عدم على على عدم عنيان قلف وكنافة فلف وهورين دافل و عدم الفارية عندل عدم عدم الفارية الفارية

In many cases it is possible to use the method to separate a mixture of two materials into its constituents, or to separate a mixture of particles of the same material into a number of size fractions.

Particlesize $\frac{d_B}{d_A} = \left(\frac{\rho_A - \rho_J}{\rho_B - \rho_J}\right)^j \quad \underline{\text{Pluid}}$

where j = 0.5 for fine particles, where Stokes' law applies, and j = 1 for coarse particles where Newton's law applies.

مجربال mm الجزي،

اذا كانه عندل وجده الم وصدة و وحدة . المحافة وصدة و وحدة . المحافة وصدة المحافة وصدة المحافة والمحافة والمحافة المعنى المحافة المعنى المحافة المعنى المحافة المعنى المحافة المعنى المحافة والمحافة والمحافة المحافة ا

4

recieve

Exampl

A mixture of quartz and galena of a size range from 0.015 mm to 0.065 mm is to be separated into two pure fractions using a hindered settling process. What is the minimum apparent density of the fluid that will give this separation? The density of galena is 750 kg/m3 and the density of quartz is 2650 kg/m3.

Stokes' law applies

$$\frac{0.065}{0.015} = \left(\frac{7500 - \rho}{2650 - \rho}\right)^{0.5}$$

$$\rho = 2377 \text{ kg/m}^3$$

Newton's law applies

$$\frac{0.065}{0.015} = \left(\frac{7500 - \rho}{2650 - \rho}\right)^{1.0}$$

$$\rho = 1196 \text{ kg/m}^3$$

Thus, the required density of the fluid is between 1196 and 2377 kg/m3.

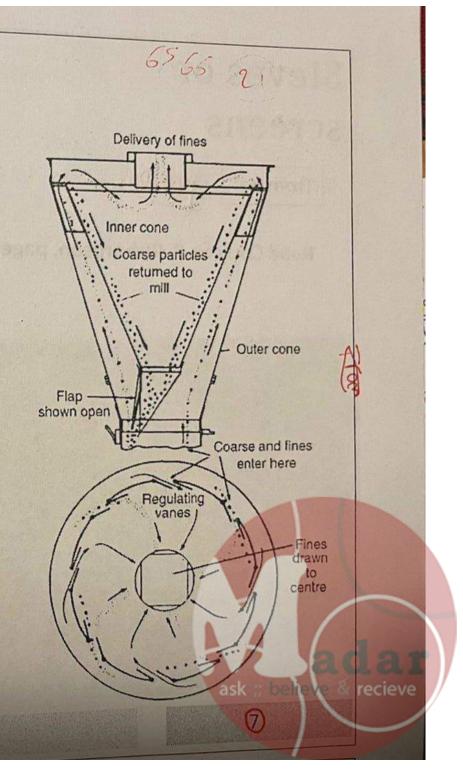
Centrifugal separators

Top clear

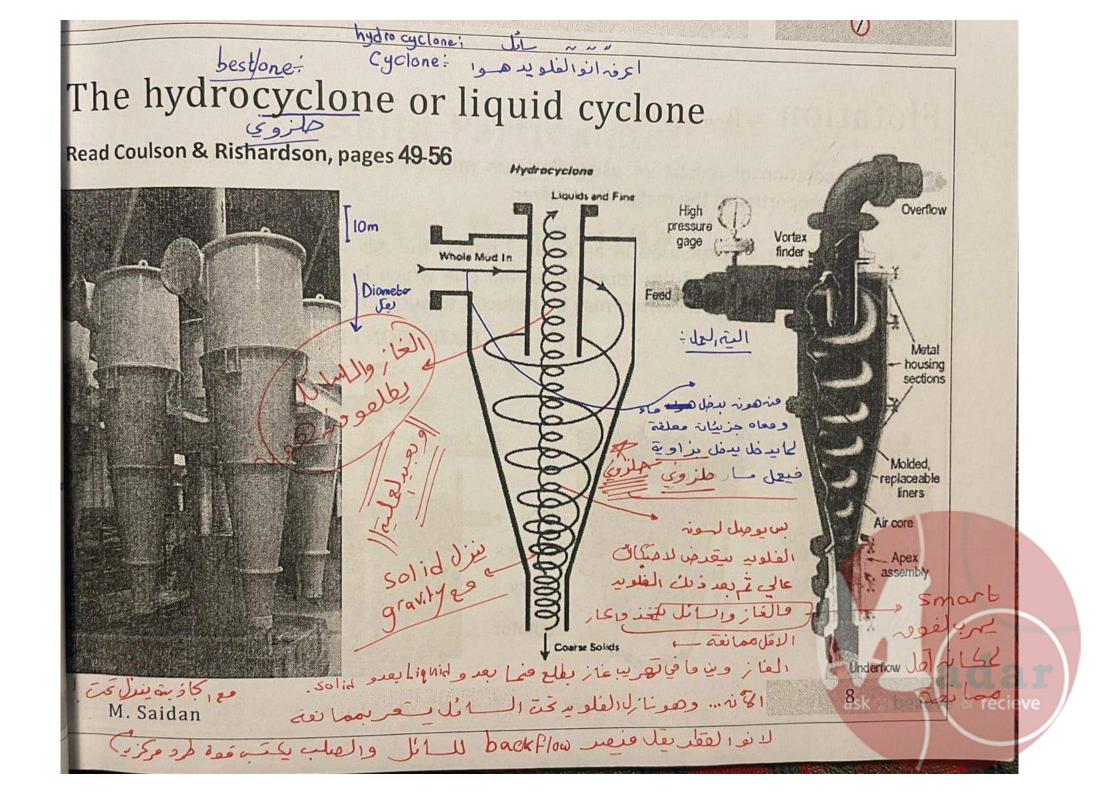
J density,

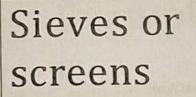
Read Coulson & Rishardson, pages 47-48

بتدخل جزيئات داخل فلويد على مصم مه و Centribus ما و المنظم من المنظم من المنظم المنظم



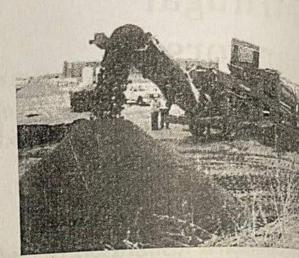
M. Saidan



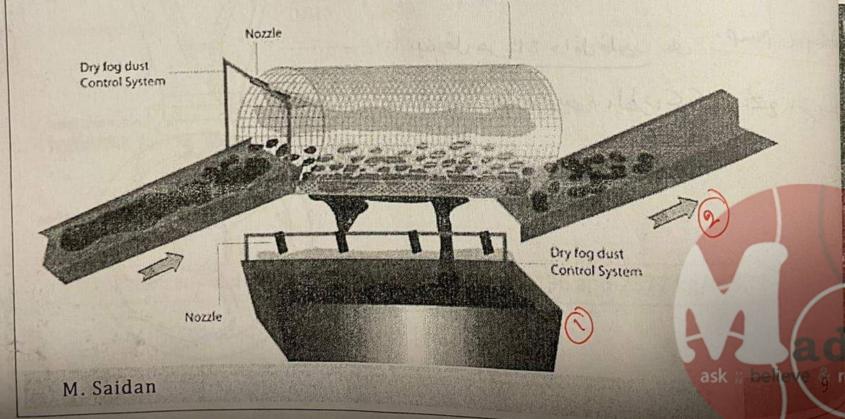


Trommel - movable

Read Coulson & Rishardson, pages 56-58

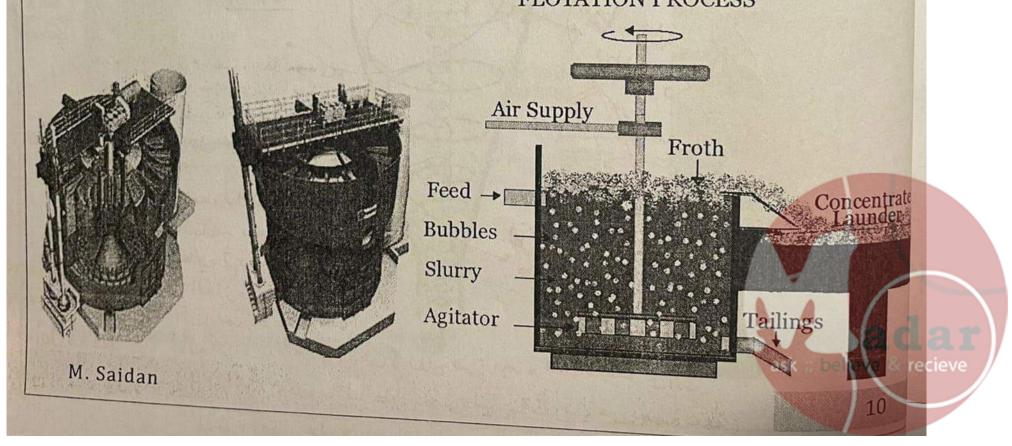


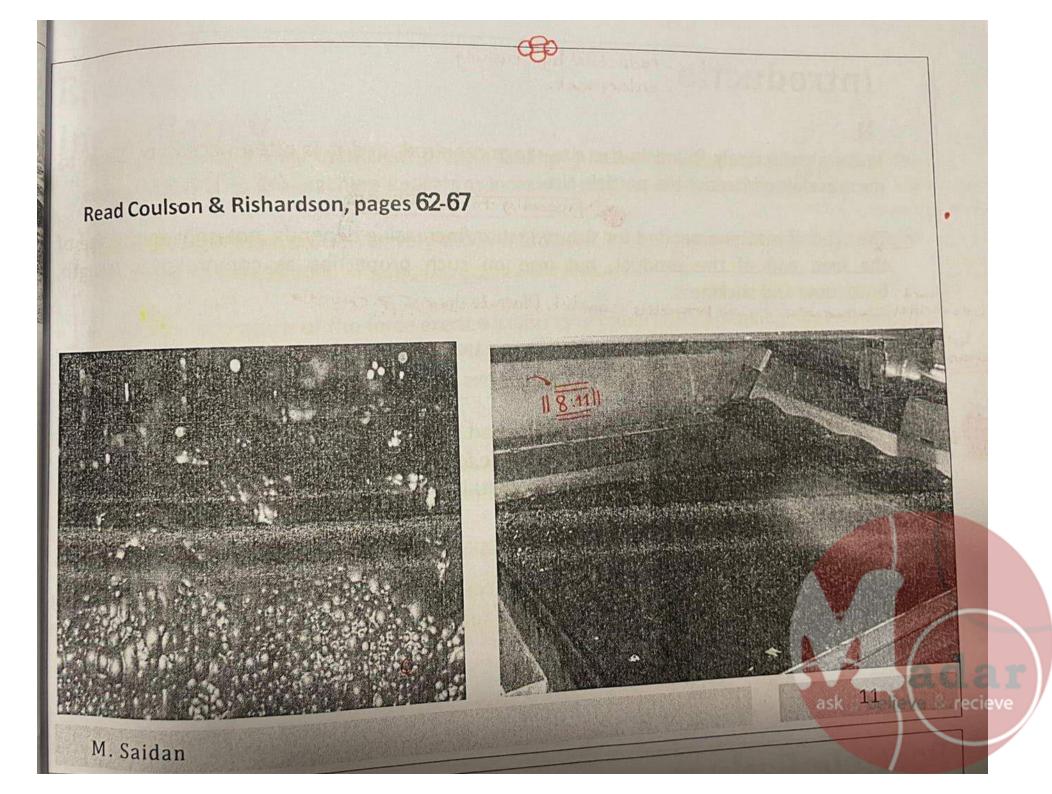
Rotating Mesh Drum



Flotation - for oily particles

- The separation of a mixture using flotation methods depends on differences in surface properties of the materials involved.
- If the mixture is suspended in an aerated liquid, the gas bubbles will tend to adh preferentially to one of the constituents—the one which is more difficult to wet by liquid—and its effective density may be reduced to such an extent that it will rise to surface.
 FLOTATION PROCESS





Introductio-reduction by crushing - enlargment.

- Materials are rarely found in the size range required, and it is often necessary either to decrease or to increase the particle size.
- The type of machine needed for size reduction/increasing depends, not only on the size the feed and of the product, but also on such properties as compressive strength ord brittleness and stickiness.

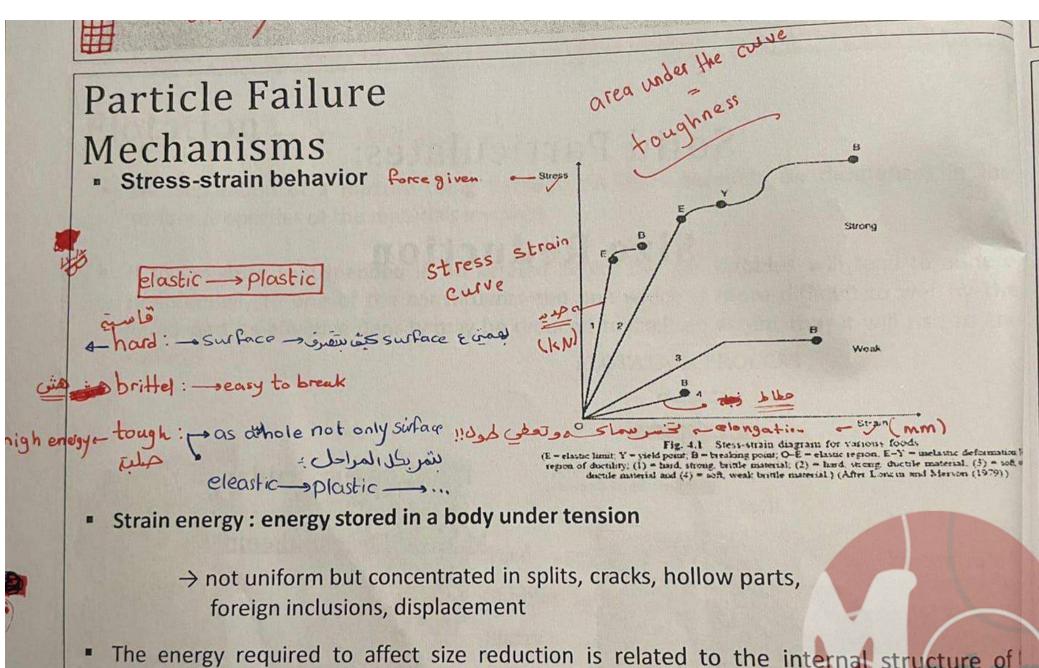
 Filter > material J. properties visitly > properties visitly
- Size reduction or comminution is an important step in the processing of many some materials:
 - √ To create particles in a certain size and shape
- To increase the surface area available for next process
 - To liberate valuable minerals held within particles benefit cation!!
- Size reduction process : extremely energy-intensive and is a very inefficient process.
- During the course of the size reduction processes, much energy is expended in causiful plastic deformation and this energy may be regarded as a waste as it does not result fracture.

highly energy — Plastic — Leformation

Consuming:

ask ;; believe & recieve





The energy required to affect size reduction is related to the internal structure of material and the process consists of two parts, first opening up any small fissures which already present, and secondly forming new surface.

Energy

Inefficiency

Size reduction is a very inefficient process and only between 0.1 and 2.0 per cent of the energy supplied to the machine appears as increased surface energy in the solids. 2 / of the energy we only need to break

The efficiency of the process is very much influenced by the manner in which the load is 98%. applied and its magnitude. ال المحمد عنوب ع particle المحمد المح

انول لغورسيس In addition the nature of the force exerted is also very important depending, for example, on whether it is predominantly a compressive, an impact or a shearing force. If the applied force is insufficient for the elastic limit to be exceeded, and the material is compressed, energy is coldine elastic di stored in the particle.

> When the load is removed, the particle expands again to its original condition without doing useful work. The energy appears as heat and no size reduction is affected.

A somewhat greater force will cause the particle to fracture, however, and in order to obtain the most effective utilization of energy the force should be only slightly in excess of the crushing strength of the material. The surface of the particles will generally be of a very irregular nature so that the force is initially taken on the high spots, with the result that very high stresses and temperatures may be set up locally in the material.

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Breakage of single

particles
All large lumps of material contain cracks and size reduction occurs as a result of crack propagation that occurs above a critical parameter, F, where

$$F = \frac{\tau^2 a}{Y}$$

where: $a = \operatorname{crack}$ length, $\tau = stress$, and Y =Young's modulus.

- At lower values of F, elastic deformation occurs without fracture and the energy input is completely ineffective in achieving size reduction.
- > The method of application of the force to the particles may affect the breakage pattern. fracture mechanisms;
- The four basic patterns are:
 - (a) Impact —particle concussion by a single rigid force. ا معلدات
 - (b) Compression particle disintegration by two rigid forces. —
 - (c) Shear —produced by a fluid or by particle –particle interaction. ____
- (d) (d) Attrition arising from particles scraping against one another or against a rigid

surface. - particle aid, &

Energy for size reduction from size size

- It is impossible to estimate accurately the amount of energy required in order to effect; size reduction of a given material,
- A number of empirical laws have been proposed:

$$\frac{\mathrm{d}E}{\mathrm{d}L} = -CL^{p}$$

which states that the energy dE required to effect a small change dL in the size of unit mass of material is a simple power function of the size. If p = -2, then integration gives:



$$E = C\left(\frac{1}{L_2} - \frac{1}{L_1}\right)$$

$$\text{from } L_1 - L_2$$



Rittinger's

law c

Writing $C = K_R f_c$, where f_c is the crushing strength of the material, then Rittinger's l_{q_0} first postulated in 1867, is obtained as:

$$E = K_R f_c \left(\frac{1}{L_2} - \frac{1}{L_1} \right)$$

Since the surface of unit mass of material is proportional to 1/L, the interpretation of the law is that the energy required for size reduction is directly proportional to the increase in surface.

where P is the power required, mo is the feed rate to crusher



Kick's

If
$$p = -1$$
, then:

$$\frac{\mathrm{d}E}{\mathrm{d}L} = -CL^{p}$$

and, writing
$$C = K_K f_c$$
:

$$E = C \ln \frac{L_1}{L_2}$$

$$E = K_K f_c \ln \frac{L_1}{L_2}$$

This supposes that the energy required is directly related to the reduction ratio L1/L2 which means that the energy required to crush a given amount of material from a 50 mm to a 25 mm size is the same as that required to reduce the size from 12 mm to 6 mm.

regardless what is L. and Le minimi Le 35 50

energy needed to reduce particle eize from 1000-500 visionis
is the same as to reduce from 500.50-25

Bond crushing law and work

Bond has suggested a law intermediate between Rittinger's and Kick's laws, by putting p = -3/2 in equation Thus:

$$E = 2C \left(\frac{1}{L_2^{1/2}} - \frac{1}{L_1^{1/2}} \right)$$
$$= 2C \sqrt{\left(\frac{1}{L_2} \right) \left(1 - \frac{1}{q^{1/2}} \right)}$$
$$q = \frac{L_1}{L_2}$$

where:

the reduction ratio. Writing $C = 5E_i$, then:

$$E = E_i \sqrt{\left(\frac{100}{L_2}\right) \left(1 - \frac{1}{q^{1/2}}\right)}$$

Bond terms E_i the work index, and expresses it as the amount of energy required to reduce unit mass of material from an infinite particle size to a size L_2 of 100 μ m, that is $q = \infty$. The size of material is taken as the size of the square hole through which 80 per cent of the material will pass.

Exampl

A material is crushed in a Blake jaw crusher such that the average size of particle is reduce from 50 mm to 10 mm with the consumption of energy of 13.0 kW/(kg/s). What would the consumption of energy needed to crush the same material of average size 75 mm to average size of 25 mm:

- a) assuming Rittinger's law applies?
- b) assuming Kick's law applies?

Which of these results would be regarded as being more reliable and why?

Solution

a) Rittinger's law.

This is given by: $E = K_R f_c [(1/L_2) - (1/L_1)]$

13.0 $K_R f_c[(1/10) - (1/50)]$ Thus:

 $K_R f_c = (13.0 \times 50/4) = 162.5 \text{ kW/(kg mm)}$ and:

Thus the energy required to crush 75 mm material to 25 mm is:

$$E = 162.5[(1/25) - (1/75)] = 4.33 \text{ kJ/kg}$$



b) Kick's law.

This is given by: $E = K_K f_c \ln(L_1/L_2)$

Thus: $13.0 = K_K f_c \ln(50/10)$

and: $K_K f_c = (13.0/1.609) = 8/08 \text{ kW/(kg/s)}$

Thus the energy required to crush 75 mm material to 25 mm is given by:

$$E = 8.08 \ln(75/25) = 8.88 \text{ kJ/kg}$$



Energy utilisation

Energy is utilized in crushing as follows:

why process is inefficient:

- (a) In producing elastic deformation of the particles before fracture occurs.
- (b) In producing inelastic deformation which results in size reduction.
- (c) In causing elastic distortion of the equipment.
- (d) In friction between particles, and between particles and the machine.
- (e) In noise, heat and vibration in the plant, and
- (f) In friction losses in the plant itself.
- It is estimated that only about 10 per cent of the total power is usefully employed.



Nature of the material to be crushed

- The choice of a machine for a given crushing operation is influenced by the nature of the product required and the quantity and size of material to be handled.
- The more important properties of the feed apart from its size are as follows:
 - 1. Hardness
 - 2. Structure
 - 3. Moisture content
 - 4. Crushing strength
 - 5. Friability
 - 6. Stickiness
 - 7. Soapiness



The Mohr Scale of Hardness

Materials are arranged in order of increasing hardness in the Mohr Scale in which the first four items rank as soft and the remainder as hard. The Mohr Scale of Hardness is:

hardness - LET

2. Rock salt or gypsum

3. Calcite

4. Fluorspar

5. Apatite

6. Felspar

7. Quartz

8. Topaz

9. Carborundum

10. Diamond.

to select the appropriate machine!



Solid Particles Separation

Sedimentation.

- Separation depends on the selection of a process in which the behavior of the material significantly influenced by some physical property.
- Sieving method is used if a material is to be separated into various size fractions. Since depends primarily on the size of the particles, also other physical properties such as the shape of the particles and their tendency to agglomerate may also be involved
- Separation depends on the differences in the behavior of the particles in a moving fluid, and in this case the size and the density of the particles are the most important factors and shape is of secondary importance.
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الطرد المركزي!

Centrifugal separation in which the solids are forced on to the walls of a vessel which is rotated to provide the centrifugal force. Sho

CAPECS+opecs.

اذا عندل عدم و عدينان فلفي وكنافة فقلفة مفورين داخل فلويدله كنا قة ... لفعالم يجب العب بكنا في الفلويد همَّ نفواد كم الله عنه المعام المع

In many cases it is possible to use the method to separate a mixture of two materials into its constituents, or to separate a mixture of particles of the same material into a number of size fractions.

Particlesize

 $\frac{d_B}{d_A} = \left(\frac{\rho_A - \rho_A}{\rho_B - \rho_A}\right)^j \xrightarrow{\text{Pluid}}$

where j = 0.5 for fine particles, where Stokes' law applies, and j = 1 for coarse particles where Newton's law applies.

Fine lie due is 131

وطلع Density بنعتا رفلوید له کتامه بین الول

والمثاني

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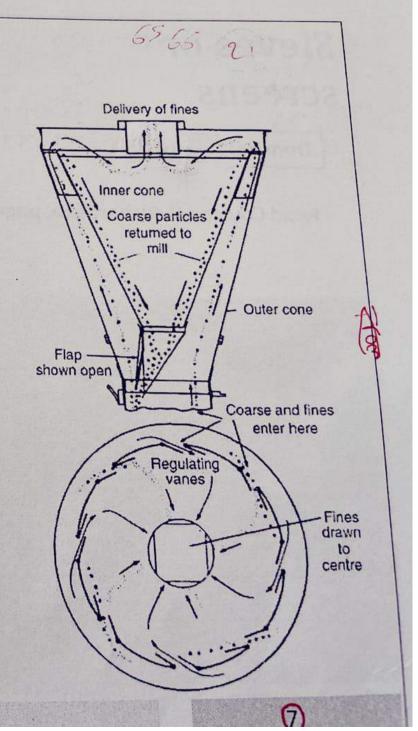
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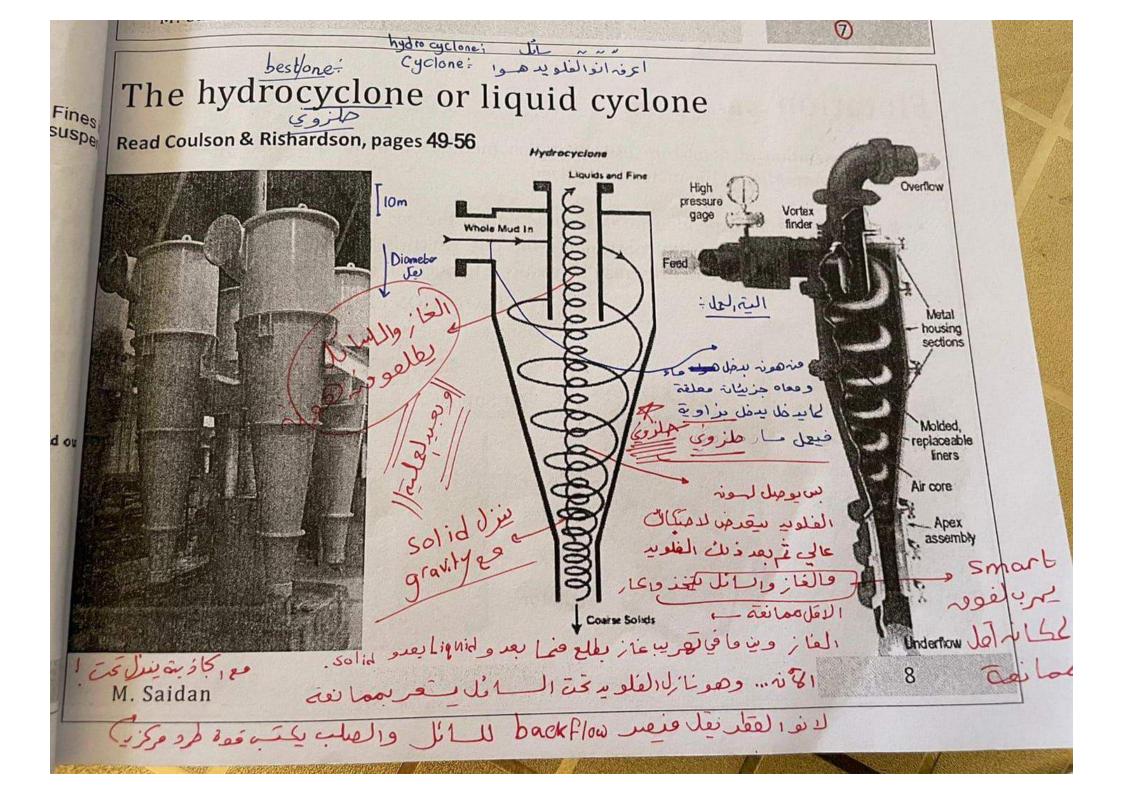
Centrifugal separators

Top clear
bottom I density,

Read Coulson & Rishardson, pages 47-48

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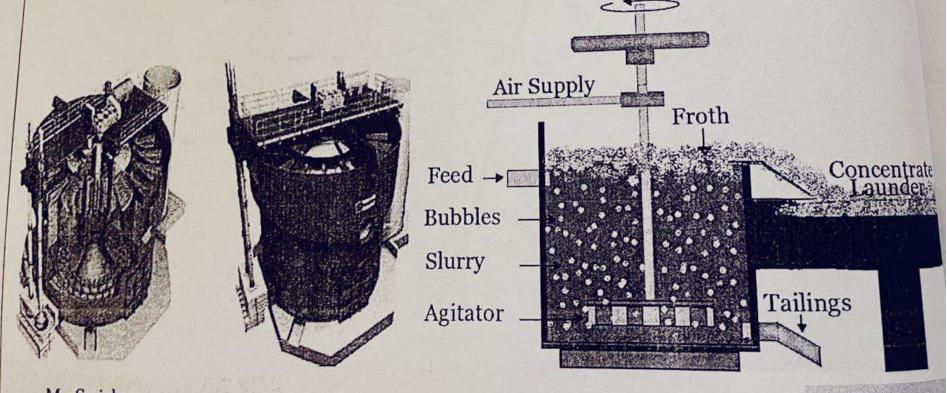


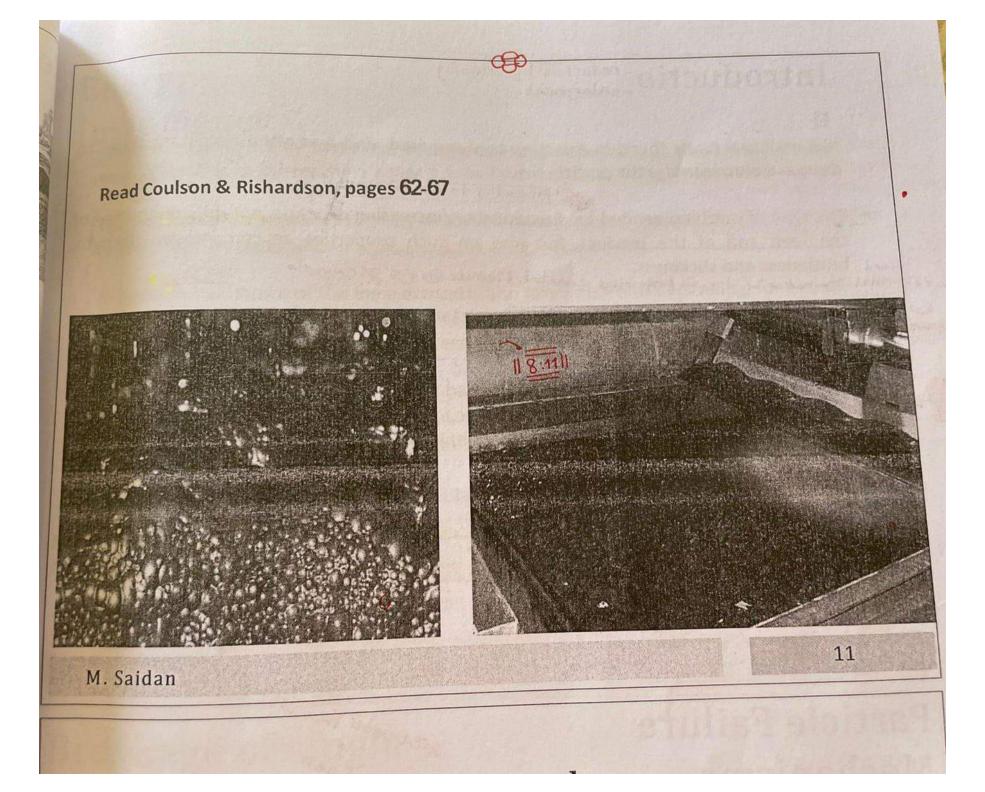




Flotation - for oily particles

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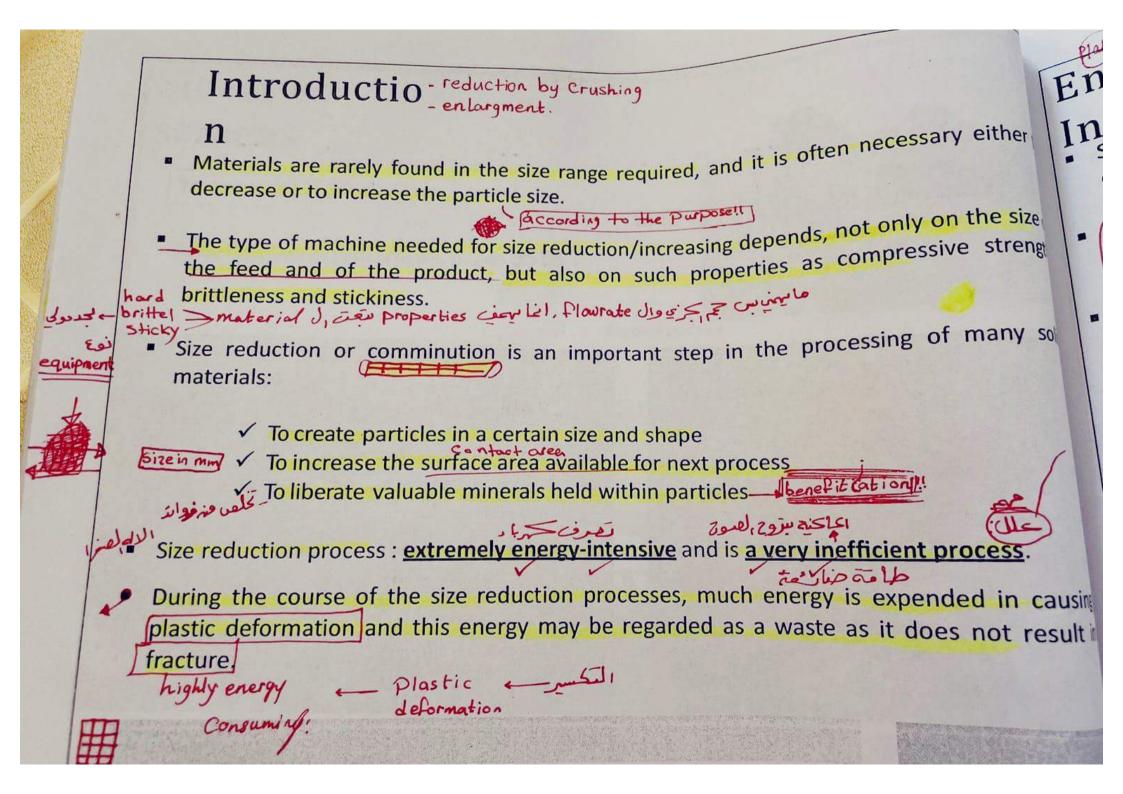


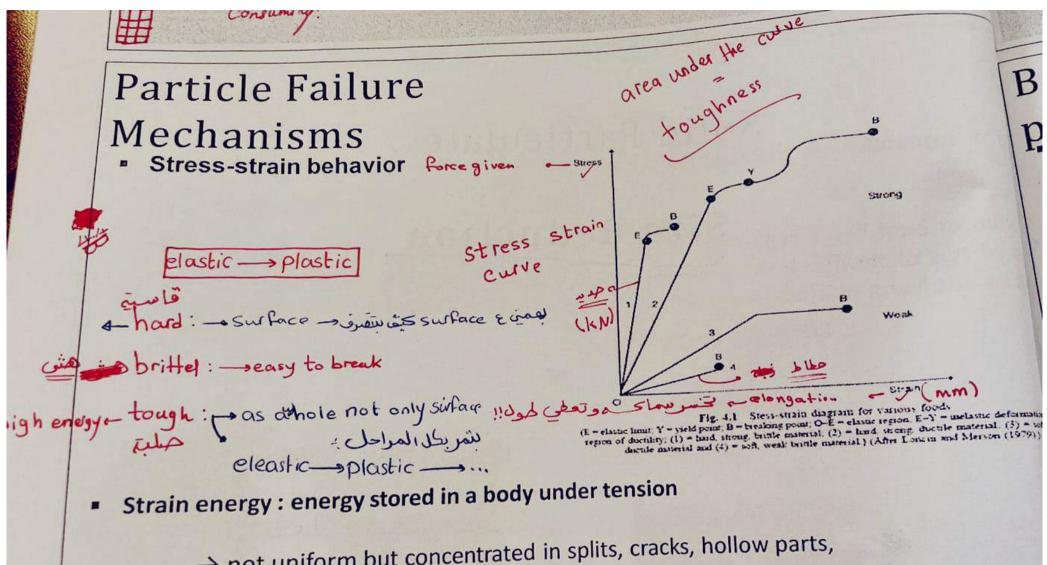
Solid Particulates:

Size Reduction

Dr. Motasem Saidan

M. Saidan@gmail.com





- → not uniform but concentrated in splits, cracks, hollow parts, foreign inclusions, displacement
- The energy required to affect size reduction is related to the internal structure of material and the process consists of two parts, first opening up any small fissures which already present, and secondly forming new surface.

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Size reduction is a very inefficient process and only between 0.1 and 2.0 per cent of the energy supplied to the machine appears as increased surface energy in the solids. 2% of the energy we only need to break

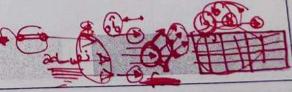
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- اول مشيل الفورس ترجع ل شكل المقبل !!

 When the load is removed, the particle expands again to its original condition without doing useful work. The energy appears as heat and no size reduction is affected.
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بقروع الكيرف ستوف و maximum energy كتى توصل الولتقلع معلى العنية وتتكس مطعليا شوى زيادة فتوى!!



Breakage of single

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where: a = crack length. $\tau = stress$, and Y = Young's modulus.

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Energy for size

reduction from size ____size

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which states that the energy dE required to effect a small change dL in the size of unit mass of material is a simple power function of the size. If p = -2, then integration gives:



$$E = C\left(\frac{1}{L_2} - \frac{1}{L_1}\right)$$

$$Prom L_1 - L_2$$

Rittinger's

law c

Writing $C = K_R f_c$, where f_c is the crushing strength of the material, then Rittinger's law, first postulated in 1967. first postulated in 1867, is obtained as: Strength of material

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Since the surface of unit mass of material is proportional to 1/L, the interpretation of this law is that the energy required for size reduction is directly proportional to the increase in surface.

$$E=P/\dot{m}^o$$

where P is the power required, mo is the feed rate to crusher

Kick's

$$\frac{\mathrm{d}E}{\mathrm{d}L} = -CL^{p}$$

If p = -1, then:

$$E = C \ln \frac{L_1}{L_2}$$

and, writing $C = K_K f_c$:

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This supposes that the energy required is directly related to the reduction ratio L1/L2 which means that the energy required to crush a given amount of material from a 50 mm to a 25 mm size is the same as that required to reduce the size from 12 mm to 6 mm.

Bond crushing law and work index c

Bond has suggested a law intermediate between Rittinger's and Kick's laws, by putting p = -3/2 in equation Thus:

$$E = 2C \left(\frac{1}{L_2^{1/2}} - \frac{1}{L_1^{1/2}} \right)$$
$$= 2C \sqrt{\left(\frac{1}{L_2} \right) \left(1 - \frac{1}{q^{1/2}} \right)}$$
$$q = \frac{L_1}{L_2}$$

where:

the reduction ratio. Writing $C = 5E_i$, then:

$$E = E_i \sqrt{\left(\frac{100}{L_2}\right) \left(1 - \frac{1}{q^{1/2}}\right)}$$

Bond terms E_i the work index, and expresses it as the amount of energy required to reduce unit mass of material from an infinite particle size to a size L_2 of 100 μ m, that is $q = \infty$. The size of material is taken as the size of the square hole through which 80 per cent of the material will pass.

Exampl

A material is crushed in a Blake jaw crusher such that the average size of particle is reduce from 50 mm to 10 mm with the consumption of energy of 13.0 kW/(kg/s). What would be the consumption of energy of 13.0 kW/(kg/s). the consumption of energy needed to crush the same material of average size 75 mm to average size of 25 mm:

- a) assuming Rittinger's law applies?
- b) assuming Kick's law applies?

Which of these results would be regarded as being more reliable and why?

Solution

a) Rittinger's law.

 $E = K_R f_c[(1/L_2) - (1/L_1)]$ This is given by:

13.0 $K_R f_c[(1/10) - (1/50)]$ Thus:

 $K_R f_c = (13.0 \times 50/4) = 162.5 \text{ kW/(kg mm)}$ and:

Thus the energy required to crush 75 mm material to 25 mm is:

$$E = 162.5[(1/25) - (1/75)] = 4.33 \text{ kJ/kg}$$

b) Kick's law.

This is given by: $E = K_K f_c \ln(L_1/L_2)$

Thus: $13.0 = K_K f_c \ln(50/10)$

and: $K_K f_c = (13.0/1.609) = 8/08 \text{ kW/(kg/s)}$

Thus the energy required to crush 75 mm material to 25 mm is given by:

$$E = 8.08 \ln(75/25) = 8.88 \text{ kJ/kg}$$

Energy utilisation

Energy is utilized in crushing as follows:

why process is inefficient:

- (a) In producing elastic deformation of the particles before fracture occurs.
- (b) In producing inelastic deformation which results in size reduction.
- (c) In causing elastic distortion of the equipment.
- (d) In friction between particles, and between particles and the machine.
- (e) In noise, heat and vibration in the plant, and
- (f) In friction losses in the plant itself.
- It is estimated that only about 10 per cent of the total power is usefully employed.

Nature of the material to be crushed

- The choice of a machine for a given crushing operation is influenced by the nature of the product required and the quantity and size of material to be handled.
- The more important properties of the feed apart from its size are as follows:
 - 1. Hardness
 - 2. Structure
 - 3. Moisture content
 - 4. Crushing strength
 - 5. Friability
 - 6. Stickiness
 - 7. Soapiness

The Mohr Scale of Hardness

Materials are arranged in order of increasing hardness in the Mohr scale in which the first four items rank as soft and the remainder as hard. The Mohr Scale of Hardness is:

hardness - Lis

2. Rock salt or gypsum 6. Felspar

- 3. Caleite
- 4. Fluorspar

5. Apatite

7. Quartz

8. Topaz 9. Carborundum

10. Diamond.

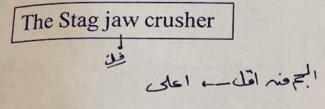
to select the appropriate machine!

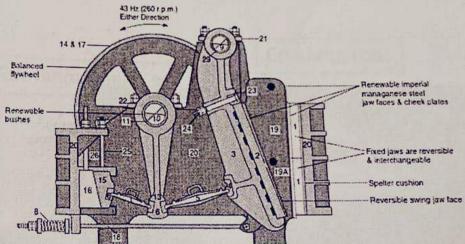
Types of Crushing Equipment crusher distribution.

Coarse crushers	Intermediate crushers	Fine crushers
Stag jaw crusher Dodge jaw crusher Gyratory crusher Other coarse crushers	Crushing rolls Disc crusher Edge runner mill Hammer mill Single roll crusher Pin mill Symons disc crusher	Buhrstone mill Roller mill NEI pendulum mill Griffin mill Ring roller mill (Lopulco Ball mill Tube mill Hardinge mill Babcock mill

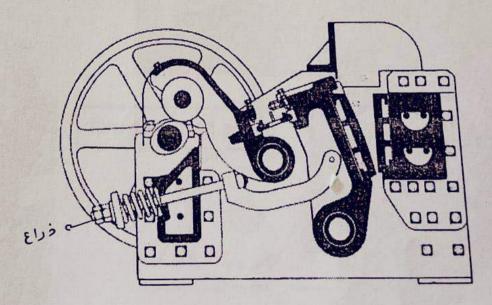
Read through page 106-137 R&C Reference Book

Coarse Crushers





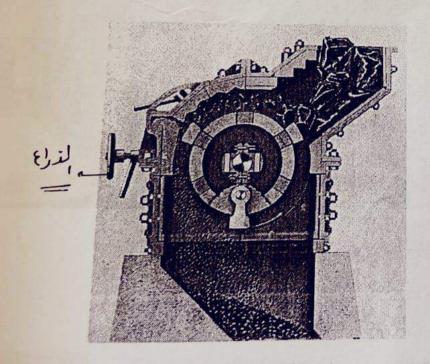
The Dodge jaw crusher

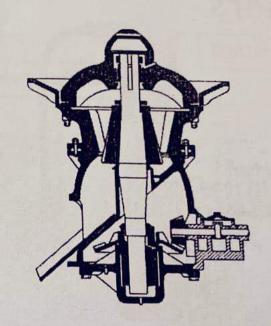


Coarse الذراء عديد المناعد ال

The gyratory crusher

Continuous flow!



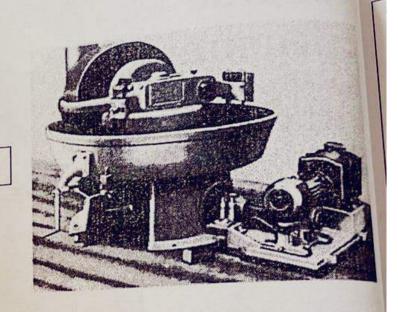


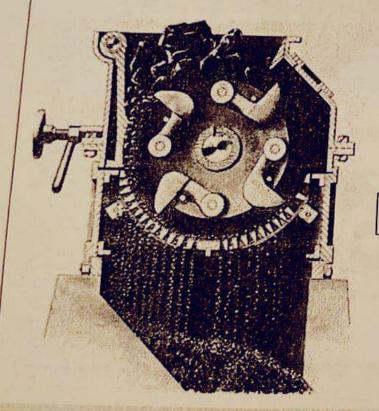
Rotary materials breaker

Intermediate crushers

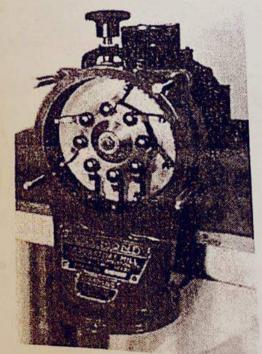
The edge runner mill

بها روثست

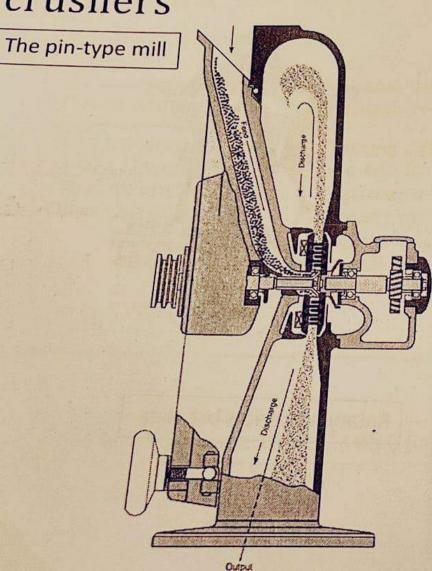


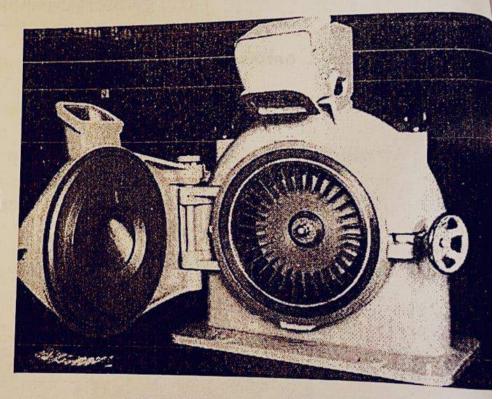


The hammer mill



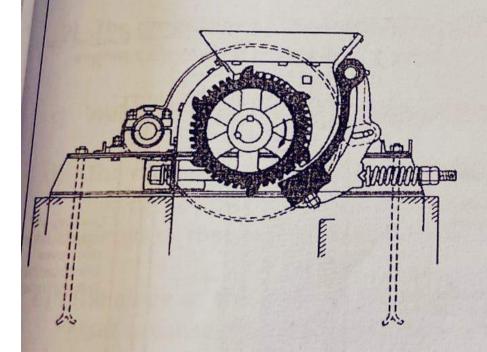
Intermediate crushers



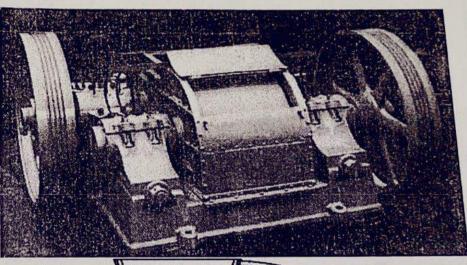


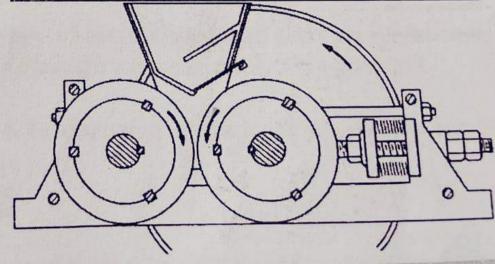
Intermediate crushers

The single roll crusher



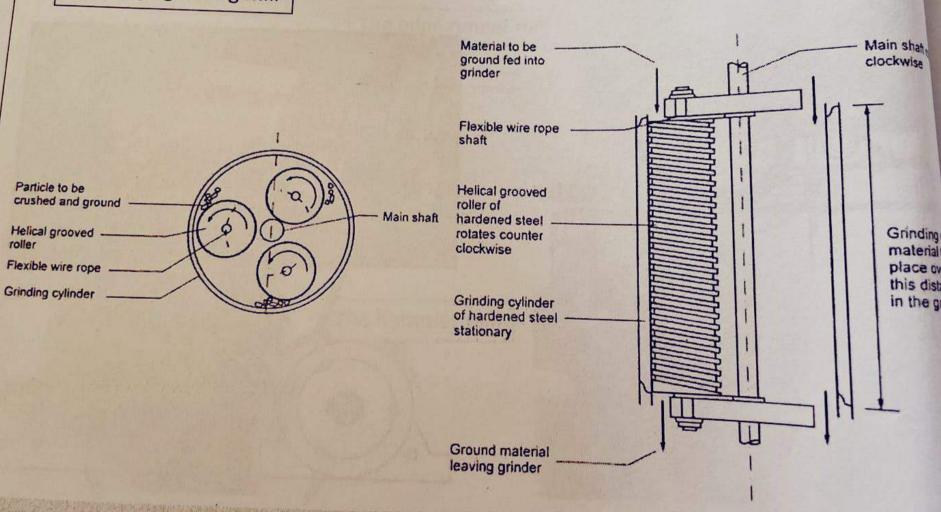
Crushing rolls



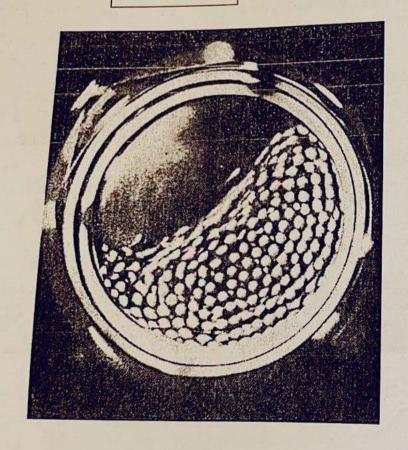


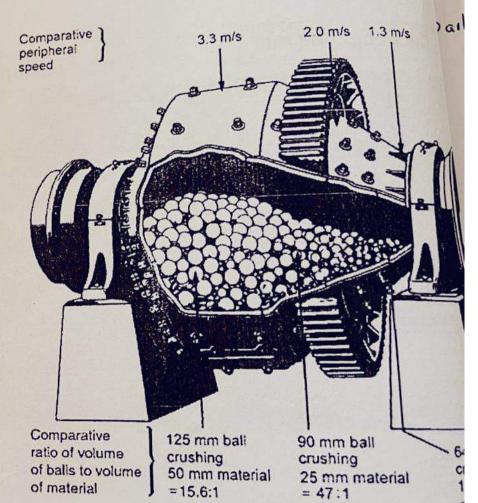
Fine crushers

The Szego grinding mill



The ball mill





Ball mill: Factors influencing the size of the product ملعبه دور بح المنتج

- The rate of feed. With high rates of feed, less size reduction is effected since the material is in the mill for a shorter time.

 high flow rate time of crushing is low sel in the low rate time -
- b) The properties of the feed material. The larger the feed the larger is the product under given operating conditions. A smaller size reduction is obtained with a hard material.
- وع زیادہ ابع بر بسرید متوہ العدم _ رسید متوہ العدم ال
- d) The diameter of the balls. Small balls facilitate the production of fine material although they do not deal so effectively with the larger particles in the feed. For most economical operation, the smallest possible balls should be used.
- e) The slope of the mill. An increase in the slope of the mill increases the capacity of the plant because the retention time is reduced, although a coarser product is obtainned.
- f) Discharge freedom. Increasing the freedom of discharge of the product has the same effect as increasing the slope. ** adjust lophimise!!

Factors influencing the size of the product

الكرات شرطت بي بعني به الكرات شرطت بي بعني به الكرات و ما تكونه و المعالمة بحث بيالجمار و المعالمة بها المحالمة المحالم

g) The speed of rotation of the mill. At low speeds of rotation, the balls simply roll over one another and little crushing action is obtained. At slightly higher speeds, the balls are projected short distances across the mill, and at still higher speeds they are thrown greater distances and considerable wear of the lining of the mill takes place. At very high speeds, the balls are carried right round in contact with the sides of the mill and little relative movement or grinding takes place again. The minimum speed at which the balls are carried round in this manner is called the critical speed of the mill and, under these conditions, there will be no resultant force acting on the ball when it is situated in contact with the lining of the mill in the uppermost position, that is the centrifugal force will be exactly equal to the weight of the ball.

30% تقريباً ليقدا كرارج

(h) The level of material in the mill. Power consumption is reduced by maintaining a low level of material in the mill, and this can be controlled most satisfactorily by fitting a suitable discharge opening for the product. If the level of material is raised, the cushioning action is increased and power is wasted by the production of an excessive quantity of undersize material.

Advantages of the ball mill (استادی)

(رسطات في) عثانه بيشل عاله الميللع !! م

- i. The mill may be used wet or dry although wet grinding the product.
- ii. The costs of installation and power are low. بتمسرتلف من ذا تنفسل
- iii. The ball mill may be used with an inert atmosphere for the grinding of explosive materials.
- iv. The grinding medium is cheap.
- v. The mill is suitable for materials of all degrees of (vi) It may be used for batch or continuous operation.

Specialized applications

- I. / Electrohydraulic crushing
- II. Ultrasonic grinding
- III. Cryogenic grinding
- IV. Explosive shattering

Read through R&C page: 137

Size Enlargement of

Particles
The finer the particles, the greater is their specific surface, and the gravitational forces acting on the particles may not be great enough to keep them apart during flow.

- The flowability of particulate systems can sometimes be improved by the use of "glidants", which are very fine powders which are capable of reducing interparticle friction by forming surface layers on the particles, thereby combating the effects of friction arising from surface roughness; they can also reduce the effects of electrostatic charges.
- Fine particles may be difficult to discharge from hoppers as particles may cling to the walls and also form bridges at the point of discharge.
- Particle size may be a critical factor since very fine particles may be exhaled, and very large particles may have a negligible effect on health. In this respect, it may be noted that the particular health hazard imposed by asbestos is largely associated with the size range and shape of the particles and their tendency to collect in the lungs.
- The size of particles may be increased from molecular dimensions by growing them by crystallization from both solutions and melts. Here, dissolving and recrystallizing may provide a mechanism for controlling both particle size and shape. It may be noted, that fine particles may also be condensed out from both vapors and gases.

Size enlargement

processes

Spray drying In this case, particle size is largely determined by the size of the droplet of liquid or suspension, which may be controlled by a suitably designed spray nozzle. The aggregates of dried material are held together as a result of the deposition of small amounts of solute on the surface of the particles. For a given nozzle, the drop sizes will be a function of both flowrate and liquid properties, particularly viscosity, and to a lesser extent outlet temperature. general, viscous liquids tend to form large drops yielding large aggregates. Pressure

Norrie axis

Niscous Paste circia

Niscous Paste circia

D Tempreture

عند , کواد عد طرب خلوا مع می spray drier Us Lds in الراس تبعو فنه فوقه فيو عاده ٥٠ ای عزی د سل منه نزول مکونه لحے) اوا کیر ادالم Roof gas disperser Inlet flue gas (upper) Rotary atomizer Central gas disperser Outlet flue gas Inlet flue gas النوزليانهم ب Paste (lower) و منه نخت شفل هوا SDA product discharge __

Rosticle \$120, هار . والمي داخل الم المجنيان متبخر تداركه مزاع ولنقسم الصبية لقسمين

..... Size enlargement

processes plantil

- * Prilling in which relatively coarse droplets are introduced into the top of a tall, narrow tower and allowed to fall against an upward flow of air. This results in somewhat large particles than those formed in spray dryers.
- Fluidized beds in this case, an atomized liquid or suspension is sprayed on to a bed of hold fluidized particles and layers of solid build up to give enlarged particles the size of which is largely dependent on their residence time, that is the time over which successive layers of solids are deposited. These are used, particularly with large particles. In this case, the rapid circulation within the bed gives rise to a high level of inter-particle impacts.
- Pug mills and extruders that is a combination of ribbing and shearing and mixing. Densification and extrusion are both achieved in a single operation. The feed, which generally has only a small water content, is subjected to a high energy input which leads to a considerable rise in temperature. The action is similar to that occurring in an extruder. High degrees of compaction are achieved, leading to the production of pellets with low porosity with the result that less binder is required.

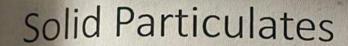




Elevated temperatures: with many materials, agglomeration may be achieved by health as a result of which softening occurs in the surface layers. For the formation of pometal sheets and discs, high temperatures are required.

Pressure compaction: If a material is subjected to very high compaction forces, it may formed into sheets, briquettes or tablets. In the tableting machines used for produce pills of pharmaceuticals, the powder is compressed into dies, either with or without addition of a binder.

Powder compaction may also be achieved in roll processes, including briquetting, which compression takes place between two rollers rotating at the same speed —that without producing any shearing action. In pellet mills, a moist feed is forced through the holes where the resistance force is attributable to the friction between the powder at the walls of the dies.



Motion of Particles in Fluid

Dr. Motasem Saidan

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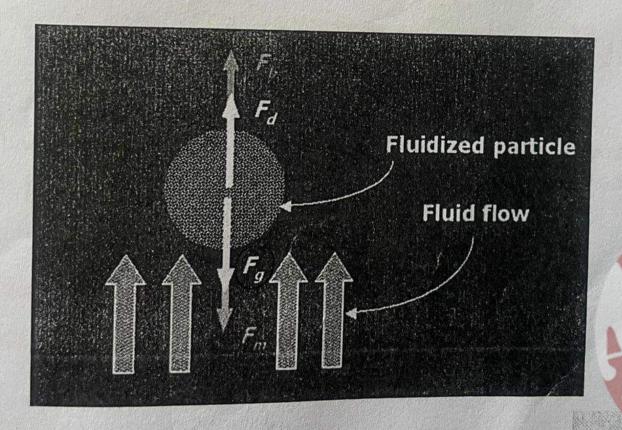
Mechanics of Particle Motion

For a rigid particle moving through a fluid, there are 3 forces acting on the body

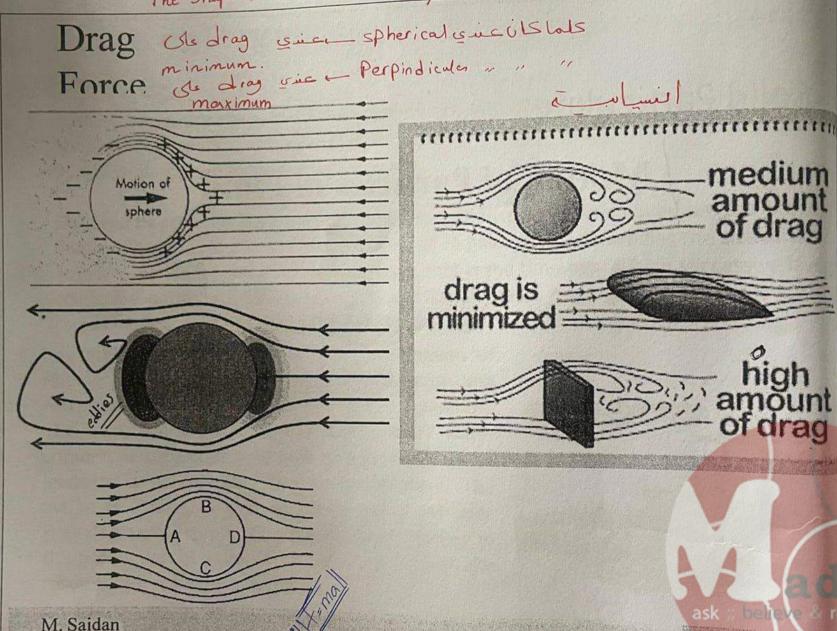
- The external force (gravitational or centrifugal force) /

- The buoyant force (opposite but parallel direction to external force) /--->

- The drag force (opposite direction to the particle motion)



The shape of the object mainly affects the drag force! . ~



One-dimensional Motion of Particle through Fluid

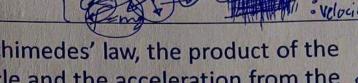
Consider a particle of mass m moving through a fluid under the action of an experience of mass m moving through a fluid under the action of an experience of the control of force F_e . Let the velocity of the particle relative to the fluid be u, let the buoyan on the particle be F_b and let the drag be F_D , then

of the particle set
$$\frac{du}{dt} = F_e - F_b - F_D$$
 (1)

The external force (Fe) - Expressed as a product of the mass (m) and the acceler (ae) of the particle from this force

$$F_e = ma_e \qquad (2)$$





The buoyant force (F_b) — Based on Archimedes' law, the product of the mass of the fluid displaced by the particle and the acceleration from the Archime des external force.

- The volume of the particle is volume of displaced Pluid. $Vp = \frac{m}{\rho_p}$

حرة لو غطستها عاء رح بهسوندي فريم بالـ٧ الفريه هاد صو ٧ تبع ليك تم تبع لكرة.

The mass of fluid displaced is $m = \frac{m}{\rho} \rho$

$$m=\frac{m}{\rho_p}\rho$$

where ρ is the density of the fluid. The buoyant force is given by

$$F_b = \frac{m\rho a_e}{\rho_p} \qquad (3)$$

The drag force (FD)

$$F_D = \frac{C_D^{\prime o'}}{2} \qquad (4)$$

where C_D is the drag coefficient, A_p is the projected area of the particle in the plane perpendicular to the flow direction.

> By substituting all the forces in the Eq. (1)

divided by mi

$$\frac{du}{dt} = a_e - \frac{\rho a_e}{\rho_p} - \frac{C_D u^2 \rho A_p}{2m} = a_e \frac{\rho_{p-} \rho}{\rho_p} - \frac{C_D u^2 \rho A_p}{2m}$$
 (5)

Case 1: Motion from gravitational force

$$\frac{du}{dt} = g \frac{\rho_{p} - \rho}{\rho_{p}} - \frac{C_{D} u^{2} \rho A_{p}}{2m} \tag{6}$$



M. Saidan

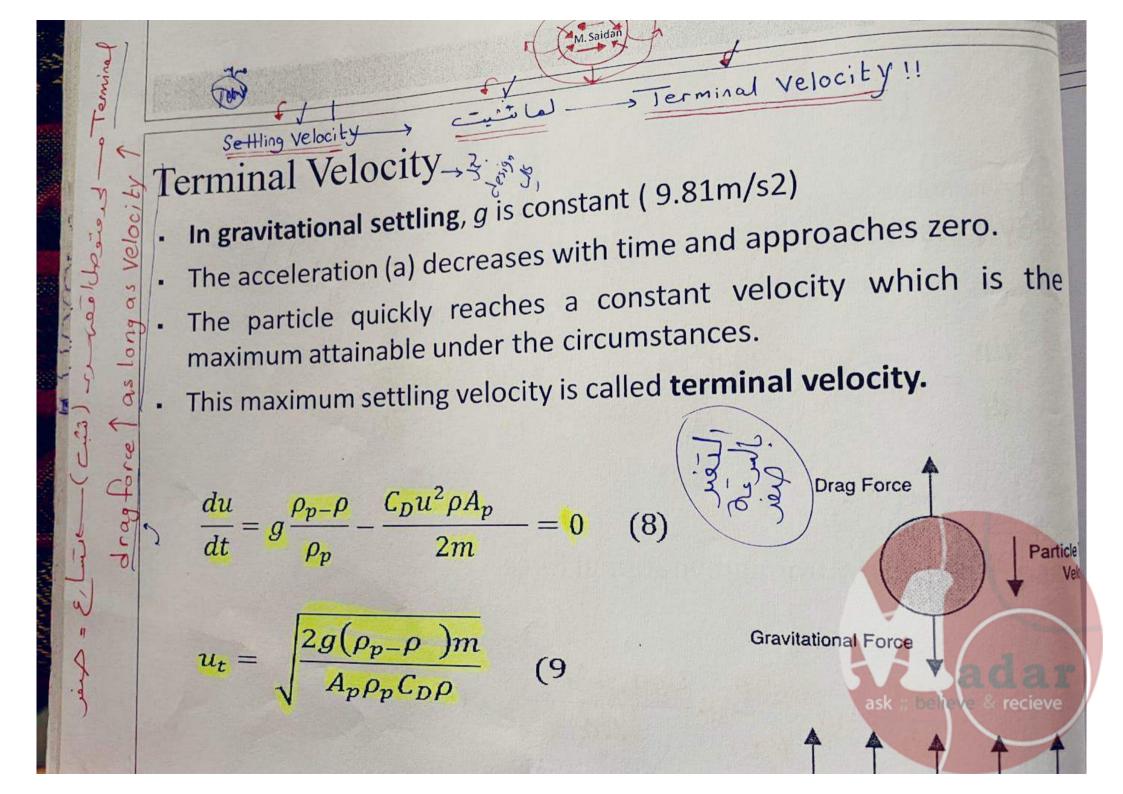
Case 2: Motion in a centrifugal field

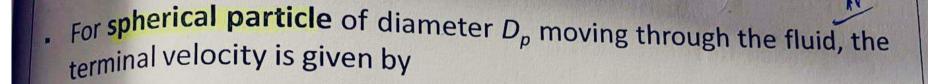
$$\frac{du}{dt} = r\omega^{2} \frac{\partial u^{2}}{\partial \rho_{p}} - \frac{C_{D}u^{2}\rho A_{p}}{2m}$$
(7)

r = radius of path of particles

 ω = angular velocity, rad/s

In this equation, **u** is the velocity of the particle relative to the fluid and is directed outwardly along a radius.





$$m = \frac{1}{6}\pi D_p^3 \rho_p$$
 $A_p = \frac{1}{4}\pi D_p^2$

. Substitution of m and A_p into the equation for u_t gives the equation for gravity settling of spheres

$$u_t = \sqrt{\frac{4gD_p(\rho_p - \rho)}{3C_D\rho}} \tag{12}$$

$$u_t = \sqrt{\frac{4gD_p(\rho_p - \rho)}{3C_D\rho}} \tag{13}$$

- . In motion from a centrifugal force, the velocity depends on the radius
- The acceleration is not constant if the particle is in motion with respect to the fluid.
- In many practical use of centrifugal force, is small ($\frac{du}{dt} = \sim 0$) thus, it can be neglected to give

$$\frac{du}{dt} = r\omega^2 \frac{\rho_{p} - \rho}{\rho_p} - \frac{C_D u^2 \rho A_p}{2m} = 0 \quad (10)$$

$$u_t = \omega \sqrt{\frac{2r(\rho_{p-}\rho)m}{A_p\rho_pC_D\rho}}$$
 (11)

Reynolds Number (Drag, Reynold) relation:

Particle Reynolds Number

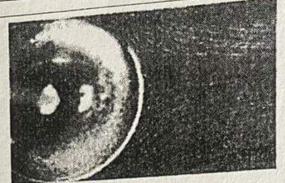
$$Re = rac{uD_p
ho}{\mu}$$
 $u:$ velocity of fluid stream $D_p:$ diameter of the particle $\rho:$ density of fluid $\mu:$ viscosity of fluid

> For the case of creeping flow, that is flow at very low velocities relative to the sphere, the

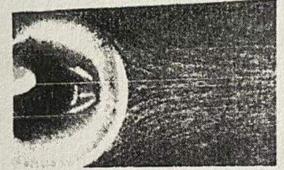
> This equation is known as Stokes' law and it is applicable only at very low values of the particle Reynolds number and deviations become progressively greater as Re increases Organ Piction Jizo

Particle moves with constant fluid, V.V

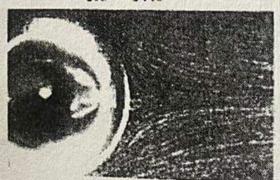
- As R_e increases, skin friction becomes proportionately less and, at values greater than about 20, flow separation occurs with the formation of vortices in the wake of the sphere.
- At high Reynolds numbers, the size of the vortices progressively increases until, at values of between 100 and 200, instabilities in the flow give rise to vortex shedding. The effect of these changes in the nature of the flow on the force exerted on the particle is now considered.



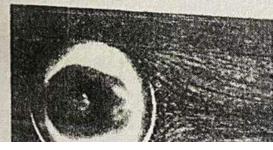
Re = 9.15



Re = 17.9

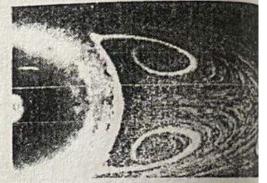


Re = 25.5

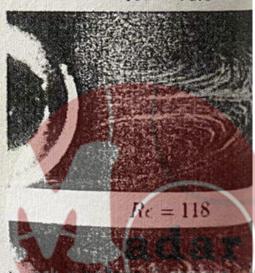




 $R\epsilon = 37.7$



Re = 73.6



pefficient

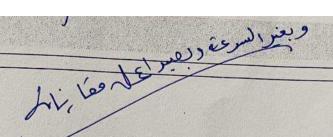
1 Syladolis. 2 dimensionless

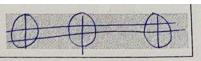
- prag coefficient is a function of Reynolds number (NRE).
- The drag curve applies only under restricted conditions:
 - The particle must be a solid sphere;
 - The particle must be far from other particles and the vessel wall so that the flow pattern around the particle is not distorted;
 - It must be moving at its terminal velocity with respect to the fluid. iii.
- The most satisfactory way of representing the relation between drag force and velocity involves the use of two dimensionless groups:

The first group is the particle Reynolds number $Re'(=ud\rho/\mu)$. The second is the group $R'/\rho u^2$, in which R' is the force per unit projected area of particle in a plane perpendicular to the direction of motion.

X-axis







For a sphere, the projected area is that of a circle of the same diameter as the sphere.

Thus:

and

$$\sum_{\alpha'} \frac{F'}{R'} = \frac{F}{(\pi d^2/4)}$$

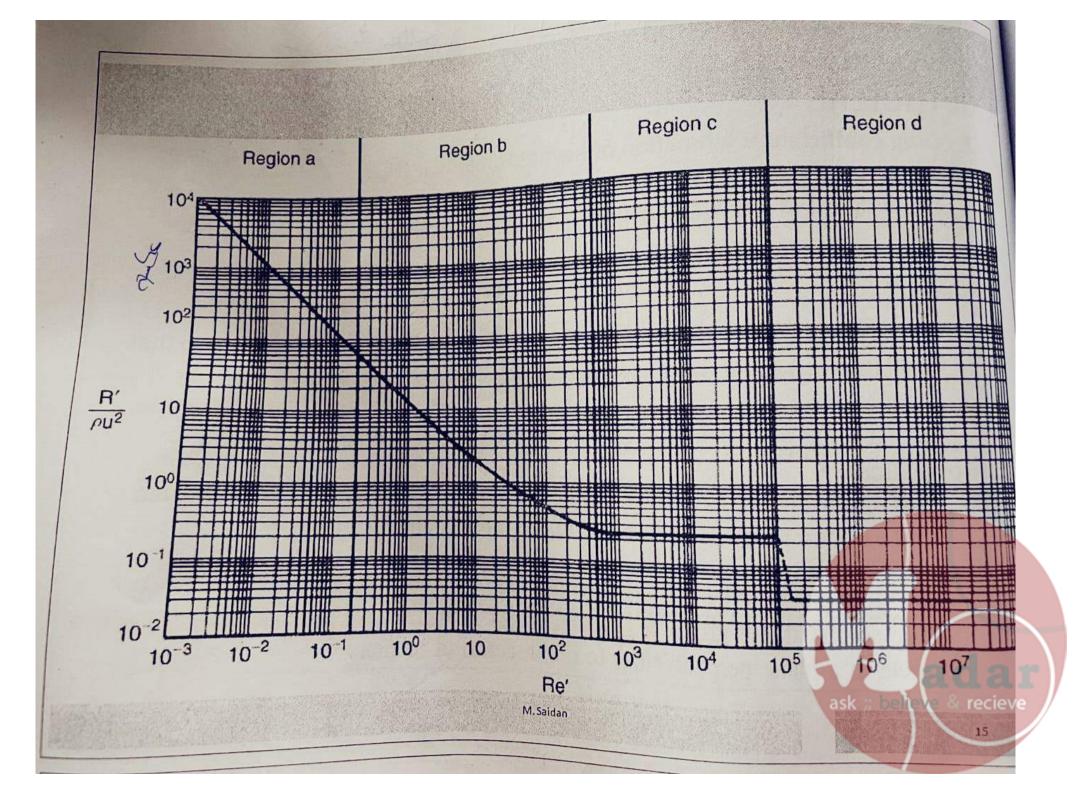
$$\frac{R'}{\rho u^2} = \frac{4F}{\pi d^2 \rho u^2}$$

 $R'/\rho u^2$ is a form of drag coefficient, often denoted by the symbol C'_D . Frequently, a drag coefficient C_D is defined as the ratio of R' to $\frac{1}{2}\rho u^2$.

Thus:

$$C_D = 2C_D' = \frac{2R'}{\rho u^2}$$

It is seen that C'_D is analogous to the friction factor $\phi (= R/\rho u^2)$ for pipe flow, and C_D is analogous to the Fanning friction factor f.



Region (a)
$$(10^{-4} < Re' < 0.2)$$

At very low values of the Reynolds number, the force F is given by Stokes' law:

$$\sqrt{\frac{3}{\rho}} \frac{y}{c'} \frac{R'}{\rho u^2} = 12 \frac{\mu}{u d\rho} = 12 Re'^{-1}$$

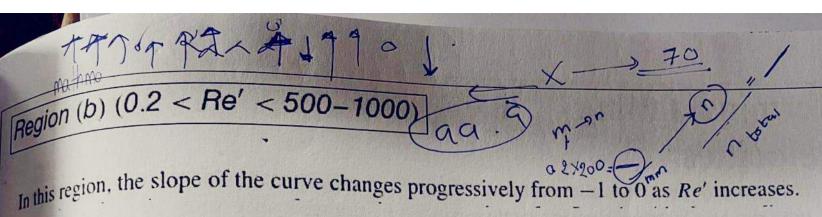
In this region, the relationship between $\frac{R'}{\rho u^2}$ and Re' is a straight line of slope

The limit of 10^{-4} is imposed because reliable experimental measurements have been made at lower values of Re'.

$$R' = 12\rho u^2 \left(\frac{\mu}{ud\rho}\right) = \frac{12u\mu}{d}$$

$$F = \frac{12u\mu}{d} \frac{1}{4}\pi d^2 = 3\pi \mu du$$





Thus:

$$\frac{R'}{\rho u^2} = 12Re'^{-1} + \frac{\text{Cool additions}}{0.22} - \frac{\text{Rem}}{\text{diagnity}}$$

$$\frac{\text{Regular that}}{\text{Shope}}$$

Stokes' law

Additional non-viscous effects

A reasonable approximation for values of Re' up to about 1000:

$$\frac{R'}{\rho u^2} = 12Re'^{-1}(1 + 0.15Re'^{0.687})$$

$$R' = \frac{12u\mu}{d} (1 + 0.15Re^{(0.687)})$$

$$F = 3\pi\mu du (1 + 0.15Re^{(0.687)})$$

$$F = 3\pi \mu du (1 + 0.15 Re^{0.687})$$

Region (c) (500-1000 < Re' < ca 2 × 10⁵)

In this region, Newton's law is applicable and the value of $R'/\rho u^2$ is approximately constant giving:

$$\frac{R'}{\rho u^2} = 0.22$$

$$R' = 0.22\rho u^2$$

$$F = 0.22\rho u^2 \frac{1}{4}\pi d^2 = 0.055\pi d^2 \rho u^2$$

Region (d) (Re' > ca
$$2 \times 10^5$$
)

When Re' exceeds about 2×10^5 , the flow in the boundary layer changes from streamline to turbulent and the separation takes place nearer to the rear of the sphere. The drag force is decreased considerably and:

$$\frac{R'}{\rho u^2} = 0.05$$

$$R' = 0.05 \rho u^2$$

$$F = 0.0125 \pi d^2 \rho u^2$$



Terminal falling velocities:

- If a spherical particle is allowed to settle in a fluid under gravity, its velocity will increase until the accelerating force is exactly balanced by the resistance force.
- The accelerating force due to gravity is given by:

$$=(\frac{1}{6}\pi d^3)(\rho_s-\rho)g$$

where ρ_s is the density of the solid.

The terminal falling velocity u_0 corresponding to region (a) is given by:

Or if the particle has started from rest, the drag force is given by :

$$(\frac{1}{6}\pi d^3)(\rho_s - \rho)g = 3\pi\mu du_0$$

and:

$$u_0 = \frac{d^2g}{18\mu}(\rho_s - \rho)$$

The terminal falling velocity corresponding to region (c) is given by:

$$(\frac{1}{6}\pi d^3)(\rho_s - \rho)g = 0.055\pi d^2\rho u_0^2$$

or:

$$u_0^2 = 3dg \frac{(\rho_s - \rho)}{\rho}$$

Under terminal falling conditions, velocities are rarely high enough for *Re'* to approach 10⁵, with the small particles generally used industry.

Assumption

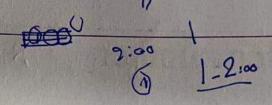
S

In the expressions given for the drag force and the terminal falling velocity, the following assumptions have been made:

- A. That the settling is not affected by the presence of other particles in the fluid. The condition is known as "free settling". When the interference of other particles appreciable, the process is known as "hindered settling".
- B. That the walls of the containing vessel do not exert an appreciable retarding effect.
- C. That the fluid can be considered as a continuous medium, that is the particle is lar compared with the mean free path of the molecules of the fluid, otherwise the particle may occasionally "slip" between the molecules and thus attain a velocity higher that calculated.



Terminal Velocity for two materials



If for a particle of material A of diameter d_A and density ρ_A , Stokes' law is applicable, then the terminal falling velocity uo is given by equation

$$u_{0A} = \frac{d_A^2 g}{18\mu} (\rho_A - \rho)$$

Similarly, for a particle of material B:

$$u_{0B} = \frac{d_B^2 g}{18\mu} (\rho_B - \rho)$$

The condition for the two terminal velocities to be equal is then:

$$\frac{d_B}{d_A} = \left(\frac{\rho_A - \rho}{\rho_B - \rho}\right)^{1/2} \longrightarrow \text{Stocks}$$

If Newton's law is applicable, equation 3.25 holds and:

$$u_{0A}^2 = \frac{3d_A g(\rho_A - \rho)}{\rho}$$
$$u_{0B}^2 = \frac{3d_B g(\rho_B - \rho)}{\rho}$$

and

For equal settling velocities:

$$u_{0B}^{2} = \frac{3d_{B}g(\rho_{B} - \rho)}{\rho}$$

$$\frac{d_{B}}{d_{A}} = \left(\frac{\rho_{A} - \rho}{\rho_{B} - \rho}\right)^{\frac{1}{2}}$$



In general, the relationship for equal settling velocities is:

$$\frac{d_B}{d_A} = \left(\frac{\rho_A - \rho}{\rho_B - \rho}\right)^S$$

where $S = \frac{1}{2}$ for the Stokes' law region, S = 1 for Newton's law and, as an approximation, $\frac{1}{2} < S < 1$ for the intermediate region.



Galileo number

(Ga)

The dimensionless group $(R'_0/\rho u_0^2)Re_0^2$ does not involve u_0 since:

$$\frac{R_0'}{\rho u_0^2} \frac{u_0^2 d^2 \rho^2}{\mu^2} = \frac{2dg(\rho_s - \rho)}{3\rho u_0^2} \frac{u_0^2 d^2 \rho^2}{\mu^2}$$
$$= \frac{2d^3(\rho_s - \rho)\rho g}{3\mu^2}$$

The group $\frac{d^3\rho(\rho_s-\rho)\dot{g}}{u^2}$ is known as the Galileo number Ga

$$\frac{R_0'}{\rho u_0^2} R e_0'^2 = \frac{2}{3} G a$$



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Table 3.4. Values of log Re' as a function of $log\{(R'/\rho u^2)Re'^2\}$ for spherical particles

$\frac{1}{\log\{(R'/\rho u^2)Re'^2\}}$	0.0	0.1	0.2	0.3	0.4	0.5	0.6	0.7	0.8
iog((N / ph /))					El Balleni			3.620	3.720
2	3.919	2.018	2.117	2.216	2.315	2.414	2.513	2.612	2.711
0	2.908	1.007	1.105	1.203	1.301	1.398	1.495	1.591	1.686
ī	1.874	1.967	0.008	0.148	0.236	0.324	0.410	0.495	0.577
2	0.738	0.817	0.895	0.972	1.048	1.124	1.199	1.273	1.346
3	1.491	1.562	1.632	1.702	1.771	1.839	1.907	1.974	2.040
4	2.171	2.236	2.300	2.363	2.425	2.487	2.548	2.608	2.667
5	2.783	2.841	2.899	2.956	3.013	3.070	3.127	3.183	3.239

Log values R

then take noverse

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Exampl

What is the terminal velocity of a spherical steel particle, 0.40 mm in diameter, settling in an oil of density 820 kg/m³ and viscosity 10 mN s/m²? The density of steel is 7870 kg/m³.

solution

For a sphere:

$$\frac{R'_0}{\rho u_0^2} R e_0'^2 = \frac{2d^3(\rho_s - \rho)\rho g}{3\mu^2}$$

$$= \frac{2 \times 0.0004^3 \times 820(7870 - 820)9.81}{3(10 \times 10^{-3})^2}$$

$$= 24.2$$

$$\log_{10} 24.2 = 1.384$$

From Table 3.4:

$$\log_{10} Re'_0 = 0.222$$

Thus:

$$Re'_0 = 1.667$$

and:



Non-Spherical Particles

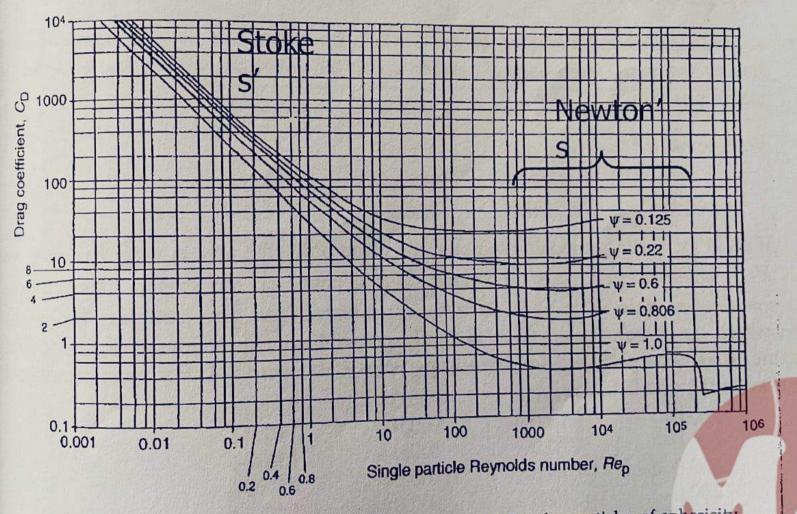


Figure 1.3 Drag coefficient C_D versus Reynolds number Re_p for particles of sphericity ψ ranging form 0.125 to 1.0 (note Re_p uses the equivalent volume diameter)

Terminal falling velocities of non-spherical particles

For a non-spherical particle:

total drag force,
$$F = R_0' \frac{1}{4} \pi d_p^2 = (\rho_s - \rho) g k' d$$

Thus:

$$\frac{R_0'}{\rho u_0^2} = \frac{4k' d_p g}{\pi \rho u_0^2} (\rho_s - \rho)$$

$$\frac{R_0'}{\rho u_0^2} R e_0'^2 = \frac{4k' \rho d_p^3 g}{\mu^2 \pi} (\rho_s - \rho)$$

and:

$$\frac{R_0'}{\rho u_0^2} R e_0'^{-1} = \frac{4k' \mu g}{\pi \rho^2 u_0^3} (\rho_s - \rho)$$

Heywood Approach

- A mean projected diameter of the particle dp is defined as the diameter of a circle having the same area as the particle when viewed from above and lying in its most stable position.
- Volume= $K' d_{p}^{3}$

If d_p is the mean projected diameter, the mean projected area is $\pi d_p^2/4$ and the volume is $k'd_p^3$, where k' is a constant whose value depends on the shape of the particle. For a spherical particle, k' is equal to $\pi/6$. For rounded isometric particles, that is particles in which the dimension in three mutually perpendicular directions is approximately the same, k' is about 0.5, and for angular particles k' is about 0.4. For most minerals k' lies between 0.2 and 0.5.

Table 3.7. Corrections to $\log Re'$ as a function of $\log \{(R'/\rho u^2)Re'^2\}$ for non-spherical particles

$\log\{(R'/\rho u^2)Re'^2\}$	k' = 0.4	k' = 0.3	k' = 0.2	k' = 0.1
2	-0.022	-0.002	+0.032	+0.131
ī	-0.023	-0.003	+0.030	+0.131
0	-0.025	-0.005	+0.026	+0.129
1	-0.027	-0.010	+0.021	+0.122
2	-0.031	-0.016	+0.012	+0.111
2.5	-0.033	-0.020	0.000	+0.080
3	-0.038	-0.032	-0.022	+0.025
3.5	-0.051	-0.052	-0.056	-0.040
4	-0.068	-0.074	-0.089	-0.098
4.5	-0.083	-0.093	-0.114	-0.146
5	-0.097	-0.110	-0.135	-0.186
5.5	-0.109	-0.125	-0.154	-0.224
6	-0.120	-0.134	-0.172	-0.255

Table 3.8. Corrections to $\log Re'$ as a function of $\{\log(R'/\rho u^2)Re'^{-1}\}$ for non-spherical particles

$\log\{(R'/\rho u^2)Re'^{-1}\}$	k' = 0.4	k' = 0.3	k' = 0.2	k' = 0.1
$\bar{4}$	+0.185	+0.217	+0.289	
4.5 3 3.5	+0.149	+0.175	+0.231	
3	+0.114	+0.133	+0.173	+0.282
3.5	+0.082	+0.095	+0.119	+0.170
$\frac{1}{2}$	+0.056	+0.061	+0.072	+0.062
2.5	+0.038	+0.034	+0.033	-0.018
Ī	+0.028	+0.018	+0.007	-0.053
1.5	+0.024	+0.013	-0.003	-0.061
0	+0.022	+0.011	-0.007	-0.062
	+0.019	+0.009	-0.008	-0.063
2	+0.017	+0.007	-0.010	-0.064 -0.065
3	+0.015	+0.005	-0.012 -0.013	-0.066
4 5	+0.013 +0.012	+0.003 +0.002	-0.013	-0.066

Example
What will be the terminal velocities of mica plates, 1 mm thick and ranging in area from 6 to 600 mm², settling in an oil of density 820 kg/m³ and viscosity 10 mN s/m²? The density of mica is 3000 kg/m³.

Solution

	smallest particles	largest particles
A'	$6 \times 10^{-6} \text{ m}^2$	$6 \times 10^{-4} \text{ m}^2$
d_p	$\sqrt{(4 \times 6 \times 10^{-6}/\pi)} = 2.76 \times 10^{-3} \text{ m}$	$\sqrt{(4 \times 6 \times 10^{-4}/\pi)} = 2.76 \times 10^{-2} \text{ m}$
$\frac{d_p}{d_p^3}$	$2.103 \times 10^{-8} \text{ m}^3$	$2.103 \times 10^{-5} \text{ m}^3$
volume	$6 \times 10^{-9} \text{ m}^3$	$6 \times 10^{-7} \text{ m}^3$
k'	0.285	0.0285

$$\left(\frac{R'_0}{\rho u^2}\right) Re'_0^2 = \frac{4k'}{\mu^2 \pi} (\rho_s - \rho) \rho d_\rho^3 g$$
 (equation 3.52)
= $(4 \times 0.285/\pi \times 0.01^2)(3000 - 820)(820 \times 2.103 \times 10^{-8} \times 9.81)$
= 1340 for the smallest particles and, similarly, 134,000 for the largest particles.

Thus:

(R'_0)	smallest particles	largest particles
$\log\left(\frac{R_0'}{\rho u_0^2}Re_0'^2\right)$	3,127	5.127
log Re' ₀ Correction from Table 3.6	1.581	2.857 (from Table 3.4)
Corrected log Re'	-0.038	-0.300 (estimated)
Re'_0	1.543	2.557
<i>u</i> ₀	34.9 0.154 m/s	361 0.159 m/s
le son a		ack to have & reci

Thus it is seen that all the mica particles settle at approximately the same velocity.

Introductio

م بنرتب particles على شكل معين بح معين فوجد بعض بعض ويضل فراغات بن ها ي particles

Most of technical process, liquid or gases flow through beds of solid

particles.

Flow of fluid - bed described through granular bed -

Example: + hrough granular bed

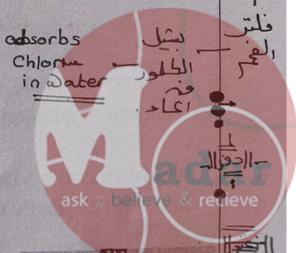
Solid

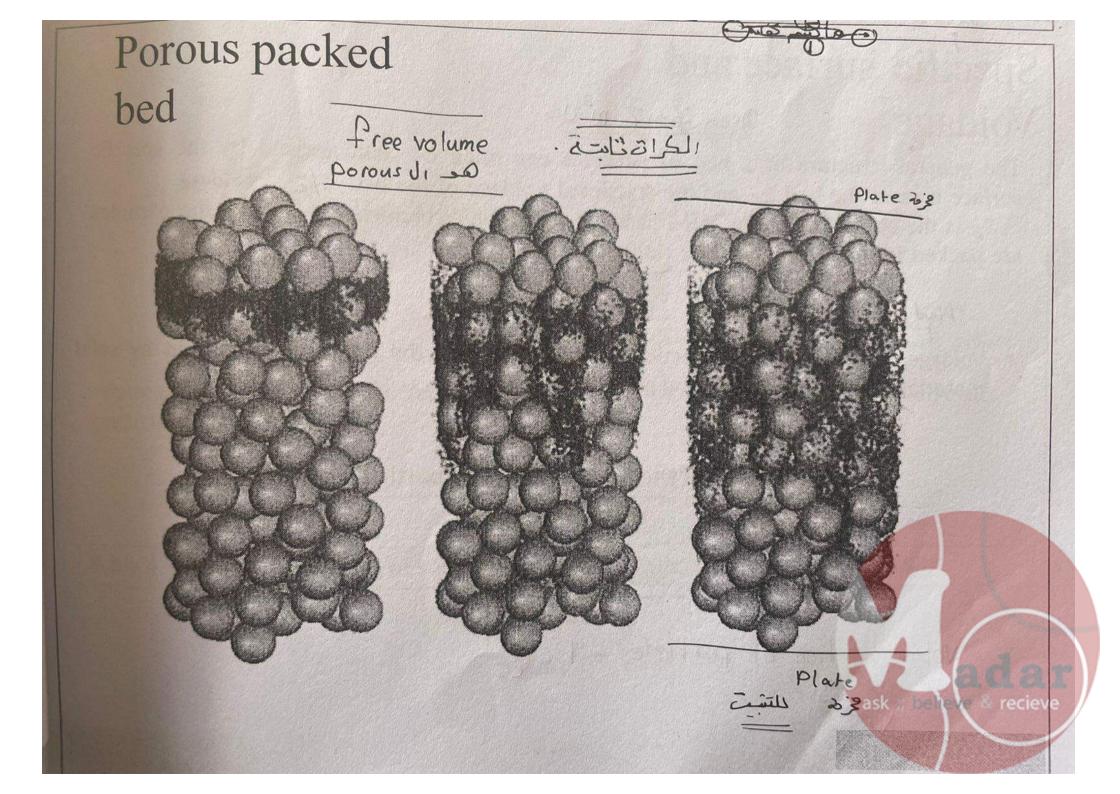
i. A single fluid flow through a bed of granular solid

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Single fluid flow through a granular bed or porous medium involves in;

- √ fixed bed reactor
- √ filtration
- √ adsorption
- seepage of underground water or petroleum زی خزانه الدیزل





Darcy Law and Permeability

- Permeability measurement is conducted to determine the surface power المادة
- Darcy's Law: the average velocity, as measured over the whole area of the bed, is directly proportional to the driving pressure and inversely proportional to the thickness of the bed.

rtional to the driving pressure and inversely properties bed.

$$u_c = \frac{K(-\Delta P)}{l} = \frac{1}{B} \frac{(-\Delta P)}{\mu l}$$

$$u_c = \frac{K(-\Delta P)}{l} = \frac{1}{B} \frac{(-\Delta P)}{\mu l}$$

where $-\Delta P$ is the pressure drop across the bed,

- 1 is the thickness of the bed,
- u_c is the average velocity of flow of the fluid, defined as (1/A)(dV/dt).
- A is the total cross sectional area of the bed,
- V is the volume of fluid flowing in time t, and
- K is a constant depending on the physical properties of the bed and fluid.

where μ is the viscosity of the fluid and B is termed the permeability coefficient for the bed, and depends only on the properties of the bed.

Specific surface and Voidage Prenifer F m3 US

The general structure of a bed of particles can often be characterised by the specific surface area of the bed S_B and the fractional voidage of the bed $e./2 \rightarrow voidage$ S_B is the surface area presented to the fluid per unit volume of bed when the particles

are packed in a bed. Its units are (length)-1.

Plaid block (3

not occupied by so Voidage/porosity (ε) -The fraction of the volume of the bed material. It is dimensionless and given by;

$$\epsilon = \frac{\text{volume of the bed} - \text{volume of particles}}{\text{volume of bed}}$$

 $= 1 - \frac{\text{volume of particles}}{\text{volume of bed}}$ Volume fraction of partieles = 1 - e



Specific surface area of the particles (S or

av) or Sp

The Specific surface area of the particles equals to the surface area of a particle divided by its volume. Its units are (length)^-1

$$a_{v} = \frac{S_{p}}{v_{p}}$$

 $a_v = \frac{S_p}{v_p}$ S_p : surface area of a particle in m² V_p : volume of particle in m³

VIP a: specific surface area of bed: S VIP S or Sp or av: "" particle

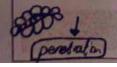
For a spherical particle,

$$h_v = \frac{6}{D_p}$$
 D_p : diameter is D_p : diame

$$S = \frac{\pi d^2}{\pi (d^3/6)} = \frac{6}{d}$$
volume

 $A_{v} = \frac{6}{D_{p}}$ $D_{p}: \text{ diameter in m}$ $S = \frac{\pi d^{2}}{\pi (d^{3}/6)} = \frac{6}{d}$ $Cked \text{ bed of non-spherical particle, the effective polyment of the speciments of the effective polyment of$ For a packed bed of non-spherical particle, the effective particle diameter D_p is defined as

$$D_P = \frac{6}{a_v}$$



volume of cylinder & you have to know them!





Since $(1 - \varepsilon)$ is the volume fraction of particles in the bed,

$$a = a_v(1 - \varepsilon) = \frac{6}{D_p} (1 - \varepsilon)$$
feget sphere
$$\varepsilon_{vell : vell : vel$$

where a is the ratio of total surface area in the bed to total volume of bed (void volume plus particle volume) in m^{-1} .

Table 4.1. Properties of beds of some regular-shaped materials

(R&C page 193)



Example

A packed bed is composed of cylinders having a diameter D = 0.02 m and a length h = D. The bulk density of the overall packed bed is 962 kg/m^3 and the density of the solid cylinders is 1600 kg/m^3 .

- (a) Calculate the void fraction &.
- (b) Calculate the effective diameter D_p of the particles.
- (c) Calculate the value of a

Solution: For part (a), taking 1.00 m³ of packed bed as a basis, the total mass of the bed is $(962 \text{ kg/m}^3)(1.00 \text{ m}^3) = 962 \text{ kg}$. This mass of 962 kg is also the mass of the solid cylinders. Hence, volume of cylinders = $962 \text{ kg/(1600 kg/m}^3) = 0.601 \text{ m}^3$. Using Eq. (3.1-6),

$$\varepsilon = \frac{\text{volume of voids in bed}}{\text{total volume of bed}} = \frac{1.000 - 0.601}{1.000} = 0.399$$

For the effective particle diameter D_p in part (b), for a cylinder where h=D, the surface area of a particle is

$$S_p = (2)\frac{\pi D^2}{4} \text{ (ends)} + \pi D(D) \text{ (sides)} = \frac{3}{2}\pi D^2$$

The volume v_p of a particle is

$$v_p = \frac{\pi}{4}D^2(D) = \frac{\pi D^3}{4}$$



Then,

$$a_v = \frac{S_p}{v_p} = \frac{\frac{3}{2}\pi D^2}{\frac{1}{4}\pi D^3} = \frac{6}{D}$$

Finally,

$$D_p = \frac{6}{a_v} = \frac{6}{6/D} = D = 0.02 \text{ m}$$

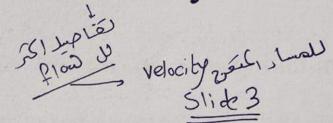
Hence, the effective diameter to use is $D_p = D = 0.02$ m. For part (c),

$$a = \frac{6}{D_p}(1 - \varepsilon) = \frac{6}{0.02}(1 - 0.399) = 180.3 \text{ m}^{-1}$$



Fluid flow through beds (Carman-Kozeny equations)

Streamline flow—Carman-Kozeny equation



- The pore space in the bed is assumed to be a tube with equivalent diameter which satisfies the following assumptions:
 - The internal surface area is equal to the surface area of particles
 - The free space is equal to that in granular bed.

If the free space in the bed is assumed to consist of a series of $m_1 = \frac{d_m'^2}{K'\mu} \frac{(-\Delta P)}{l'}$

$$u_1 = \frac{d_m'^2}{K'\mu} \frac{(-\Delta P)}{l'} \quad \boxed{2}$$

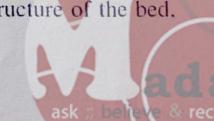
where: d'_m is some equivalent diameter of the pore channels,

K' is a dimensionless constant whose value depends on the structure of the bed.

I' is the length of channel, and

 u_1 is the average velocity through the pore channels.

 μ is the viscosity of the fluid,



$$d_m' = \frac{e}{S_B} = \frac{e}{S(1-e)}$$

 $\frac{e}{S_B} = \frac{\text{volume of voids filled with fluid}}{\text{wetted surface area of the bed}}$ $= \frac{\text{cross-sectional area normal to flow}}{\text{wetted perimeter}}$

$$\frac{e}{S_B} = \frac{1}{4}$$
 (hydraulic mean diameter)

$$S_B = S(1 - e)$$

e is the fractional voidage



Then taking $u_1 = u_c/e$ and $l' \propto l$,

$$u_{c} = \frac{1}{K''} \frac{e^{3}}{S_{B}^{2}} \frac{1}{\mu} \frac{(-\Delta P)}{l}$$

$$= \frac{1}{K''} \frac{e^{3}}{S^{2}(1-e)^{2}} \frac{1}{\mu} \frac{(-\Delta P)}{l}$$

K'' is generally known as Kozeny's constant and a commonly accepted value for K'' is K'' is dependent on porosity, particle shape, and other factors.

The permeability coefficient is given by:

$$B = \frac{1}{K''} \frac{e^3}{S^2 (1 - e)^2}$$

M. Saidan



Inserting a value of 5 for K''

$$u_c = \frac{1}{5} \frac{e^3}{(1 - e)^2} \frac{-\Delta P}{S^2 \mu l}$$

For spheres: S = 6/d and:

$$u_c = \frac{1}{180} \frac{e^3}{(1-e)^2} \frac{-\Delta P d^2}{\mu l}$$
$$= 0.0055 \frac{e^3}{(1-e)^2} \frac{-\Delta P d^2}{\mu l}$$

For non-spherical particles, the Sauter mean diameter d_s should be used in place

ask ;; believe & recieve

MA Coldon

Streamline and turbulent flow

Since equation applies to streamline flow conditions, though CARMAN and others have extended the analogy with pipe flow to cover both streamline and turbulent flow conditions through packed beds.

In this treatment a modified friction factor $R_1/\rho u^2_1$ is plotted against a modified Reynolds number Re_1 . This is analogous to plotting $R/\rho u^2$ against Re for flow through a pipe

The modified Reynolds number Re_1 is obtained by taking the same velocity and characteristic linear dimension d'_m as were used in deriving equation 4.9. Thus:

$$Re_1 = \frac{u_c}{e} \frac{e}{S(1-e)} \frac{\rho}{\mu}$$
$$= \frac{u_c \rho}{S(1-e)\mu}$$



R1 can be related to the properties of the bed and pressure gradient as follows:

Considering the forces acting on the fluid in a bed of unit cross-sectional area and thickness l, the volume of particles in the bed is l(1-e) and therefore the total surface is S l(1-e).

Thus the resistance force is R_1S l(1-e) This force on the fluid must be equal to that produced by a pressure difference of ΔP across the bed. Then, since the free cross-section of fluid is equal to e:

and

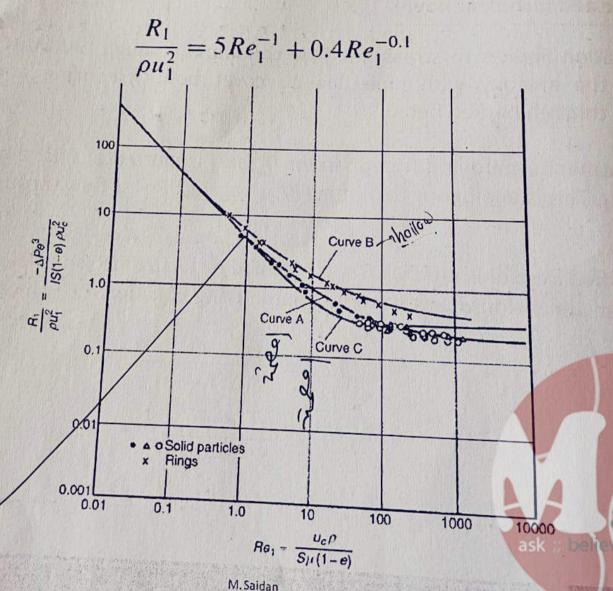
Thus

$$(-\Delta P)e = R_1 SI(1 - e)$$

$$R_1 = \frac{e}{S(1 - e)} \frac{(-\Delta P)}{I}$$

$$\frac{R_1}{\rho u_1^2} = \frac{e^3}{S(1 - e)} \frac{(-\Delta P)}{I} \frac{1}{\rho u_c^2}$$

Carman found that when $R_1/\rho u_1^2$ was plotted against Re_1 using logarithmic coordinates the flow through randomly packed beds of solid particles could be contapproximately by a single curve (curve A, Figure 4.1), whose general equation is:



From equation 4.16 it can be seen that for values of Re_1 less than about 2, the \sec_{00} term is small and, approximately:

$$\frac{R_1}{\rho u_1^2} = 5Re_1^{-1} \tag{4.18}$$

$$u_c = \frac{1}{5} \left(\frac{1}{1 - e} \right) \left(\frac{\rho u_c^2}{S\mu} \right) \left(\frac{R_1}{\rho u_1^2} \right)$$

$$\frac{R_1}{\rho u_1^2} = 5 \left(\frac{S(1 - e)\mu}{u_c \rho} \right)$$

$$= 5Re_1^{-1}$$



As the value of Re_1 increases from about 2 to 100, the second term in equation 4.16 becomes more significant and the slope of the plot gradually changes from -1.0 to about $-\frac{1}{4}$. Above Re_1 of 100 the plot is approximately linear. The change from complete streamline flow to complete turbulent flow is very gradual because flow conditions are not the same in all the pores. Thus, the flow starts to become turbulent in the larger pores, and subsequently in successively smaller pores as the value of Re_1 increases.

Rings, which as described later are often used in industrial packed columns, tend to deviate from the generalised curve A on Figure 4.1 particularly at high values of Re_1 .

Sawistowski⁽⁹⁾ compared the results obtained for flow of fluids through beds of hollow packings (discussed later) and has noted that equation 4.16 gives a consistently low result for these materials. He proposed:

$$\frac{R_1}{\rho u_1^2} = 5Re_1^{-1} + Re_1^{-0.1}$$

This equation is plotted as curve B in Figure 4.1.

Ergun semi-empirical correlation

For flow through ring packings which as described later are often used in industrial packed columns, Ergun⁽¹⁰⁾ obtained a good semi-empirical correlation for pressure drop as follows:

$$\frac{-\Delta P}{l} = 150 \frac{(1-e)^2}{e^3} \frac{\mu u_c}{d^2} + 1.75 \frac{(1-e)}{e^3} \frac{\rho u_c^2}{d}$$
om equation 4.3): (4.20)

Writing d = 6/S (from equation 4.3):

$$\frac{-\Delta P}{Sl\rho u_c^2} \frac{e^3}{1 - e} = 4.17 \frac{\mu S(1 - e)}{\rho u_c} + 0.29$$

$$\frac{R_1}{\rho u_1^2} = 4.17 Re_1^{-1} + 0.29$$
(4.21)

or:

This equation is plotted as curve C in Figure 4.1. The form of equation 4.21 is somewhat similar to that of equations 4.16 and 4.17, in that the first term represents viscous losses which are most significant at low velocities and the second term represents kinetic energy losses which become more significant at high velocities. The equation is thus applicable over a wide range of velocities and was found by Ergun to correlate experimental data well for values of $Re_1/(1-e)$ from 1 to over 2000.

Dependence of K'' on bed structure

Tortuosity

CARMAN(1) has shown that:

$$K'' = \left(\frac{l'}{l}\right)^2 \times K_0 \tag{4.22}$$

where (l'/l) is the tortuosity and is a measure of the fluid path length through the bed compared with the actual depth of the bed,

 K_0 is a factor which depends on the shape of the cross-section of a channel through which fluid is passing.



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Non-spherical particles

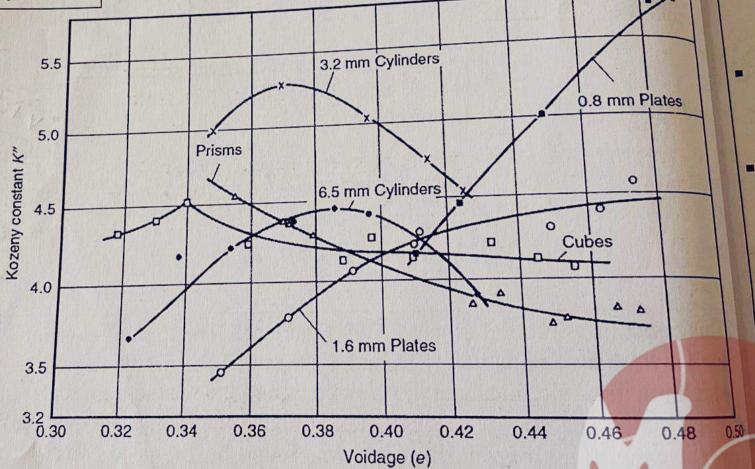


Figure 4.2. Variation of Kozeny's constant K'' with voidage for various shapes

M. Saidan

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Table 4.2. Experimental values of K'' for beds of high porosity

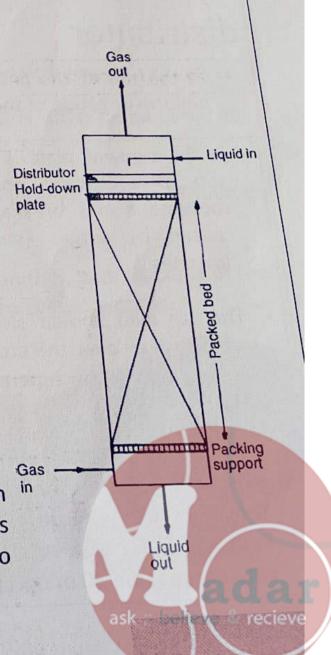
Voidage e	Experimental value of K"		
	Brinkman ⁽³⁾	Davies ⁽²¹⁾	Silk fibres LORD ⁽²⁰⁾
0.5	5.5		
0.6	4.3		
0.8	5.4	6.7	5.35
0.9	8.8	9.7	6.8
0.95	15.2	15.3	9.2
0.98	32.8	27.6	15.3



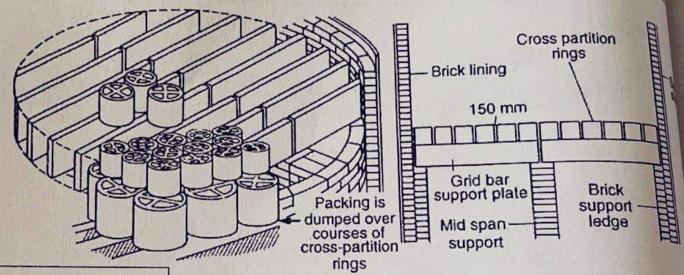
Packed

Columns

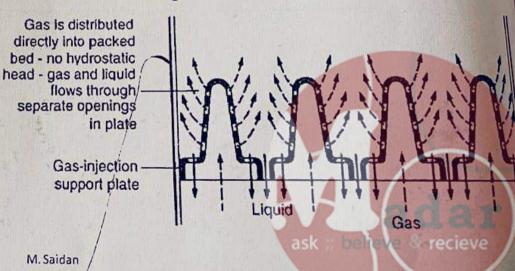
- Packed columns consist of shaped particles contained within a column, their behavior will in many ways be similar to that of packed beds which have already been considered.
- Packed towers are used for bringing two phases in contact with one another and there will be strong interaction between the fluids.
- Normally one of the fluids (liquid) will preferentially wet the packing and will flow as a film over its surface; the second fluid (gas) then passes through the remaining volume of the column.
- An example of the liquid-gas system is an absorption process where a soluble gas is scrubbed from a mixture of gases by means of a liquid.
- In the construction of packed towers, the shell of the column may be constructed from metal, ceramics, glass, or plastics material. The column should be mounted truly vertically to help uniform liquid distribution.



The bed of packing rests on a support plate which should be designed to have at least of the passage of the gas so as to offer as low a resistance as possible.



The gas injection plate is designed to provide separate passageways for gas and liquid so that they need not vie for passage through the same opening. This is achieved by providing the gas inlets to the bed at a point above the level at which liquid leaves the bed.



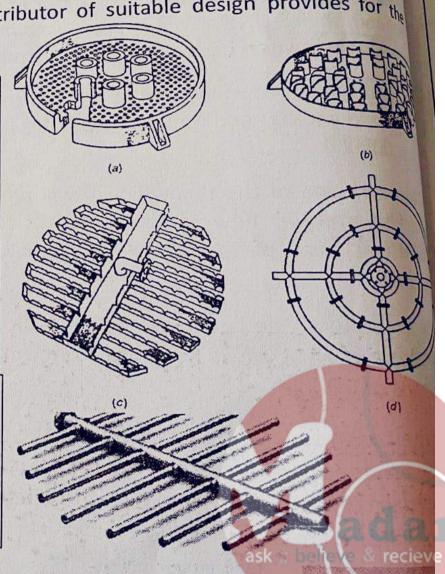
Types of liquid distributor

At the top of the packed bed a liquid distributor of suitable design provides for the

uniform irrigation of the packing

A "hold-down" plate is often placed at the top of a packed column to minimize movement and breakage of the packing caused by surges (sudden increase) in flowrates.

- ✓ The gas inlet should also be designed for uniform flow over the cross-section and the gas exit should be separate from the liquid inlet
- ➤ Columns for both absorption and distillation vary in diameter from about 25 mm for small laboratory purposes to over 4.5 m for large industrial operations; these industrial columns may be 30 m or more in height.



packing (Rea

(Read through R&C page 216 - 221)

Spackings, grids, and structured packings.

page 216 - 221)

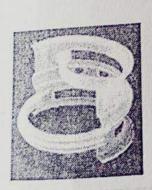
packings, grids, and structured packings.

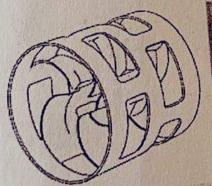
Most of these packings are available in a wide range of materials such as ceramics, metals, glass, plastics, carbon, and sometimes rubber.

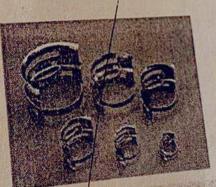
Non-porous solid should be used if there is any risk of crystal formation in the pores when the packing dries, as this can give rise to serious damage to the packing elements.

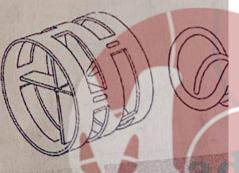
Channeling, that is non-uniform distribution of liquid across the column cross-section, is much less marked with shaped packings, and their resistance to flow is much less.

Shaped packings also give a more effective surface per unit volume because surface contacts are reduced to a minimum and the film flow is much improved compared with broken solids.





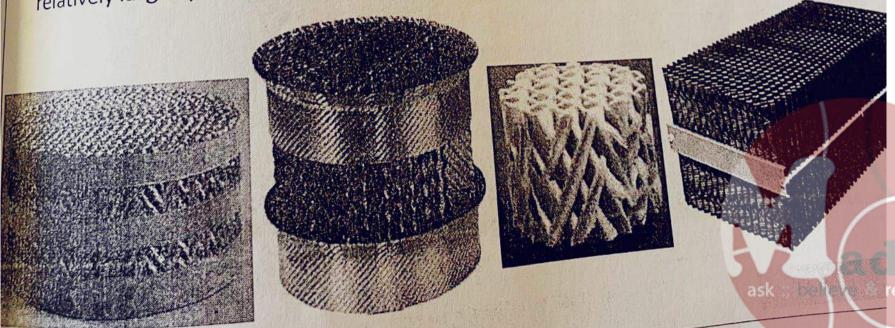


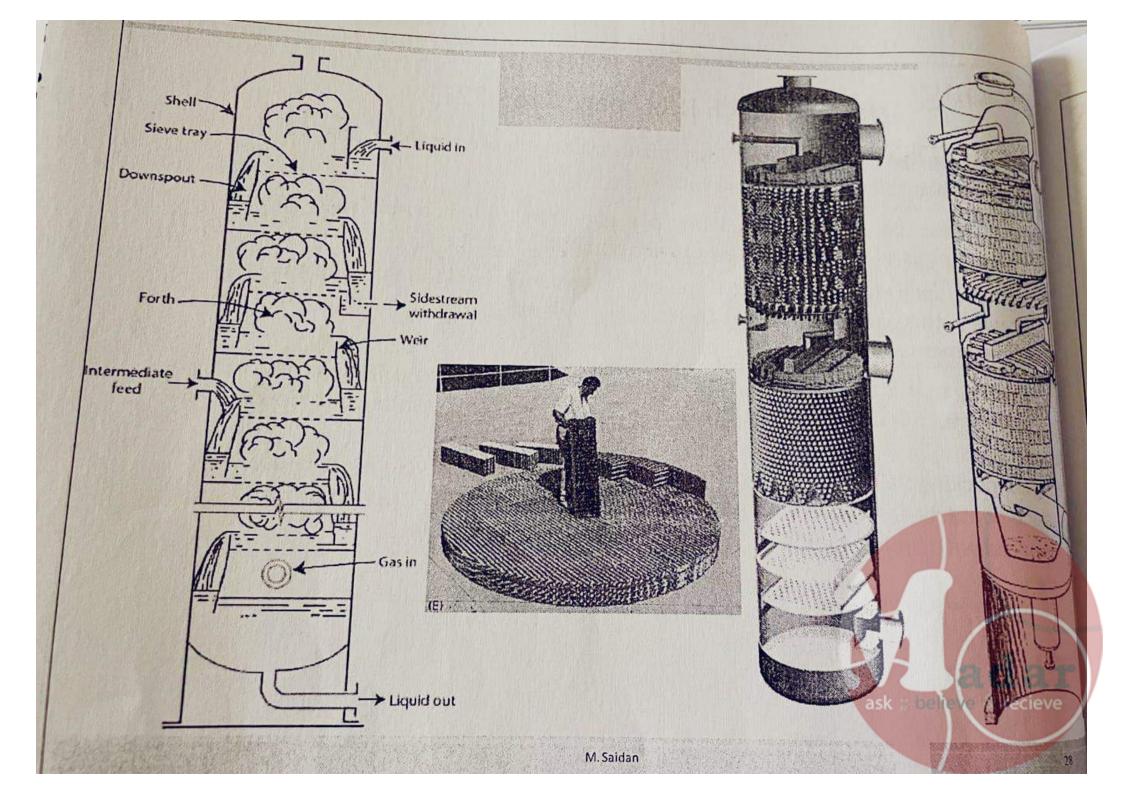


packing

Table 4.3. Design data for various packings

- S The voidage obtainable with these packings varies from about 0.45 to 0.95
- To obtain high and uniform voidage and to prevent breakage, it is often found better to dump the packings into a tower full of liquid.
- Grid packings, which are relatively easy to fabricate, are usually used in columns of square section, and frequently in cooling towers.
- They may be made from wood, plastics, carbon, or ceramic materials, and, because of the relatively large spaces between the individual grids, they give low pressure drops.





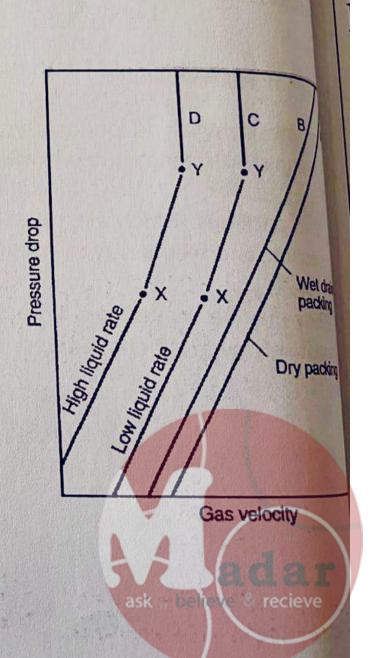
Fluid flow in packed columns

Pressure drop

The drop in pressure arising from the flow of a single phase through granular beds is considered and the same general form of approach is usefully adopted for the flow of two fluids through packed columns.

Loading (X) and flooding (Y) points

Hold-up



Read through R&C pages 222-228